

# Total Synthesis of Natural Products

## CH438

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**Teaching Assitant:**

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*Lecture Time: Thursday 15:00-17:00*  
*Lecture Room: BCH 1103*

*Exercise(s): ...*

# Outline

- **Introduction**
- **Classical examples (Longifolene, Reserpine, Strychnine...)**
- **Syngernism between strategy and total synthesis**

**Domino, multicomponent reaction**

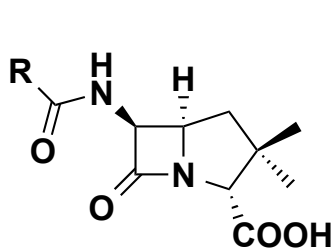
**Oxidative coupling, Pattern Recognition**

**Hidden Symmetry, C-H Functionalization**

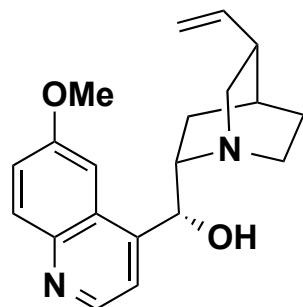
**Asymmetric organocatalysis**

.....

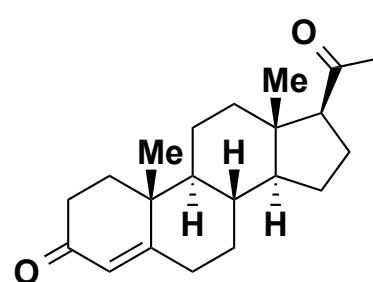
# Why Natural Products: Source of Drugs, Structural diversity and Complexity



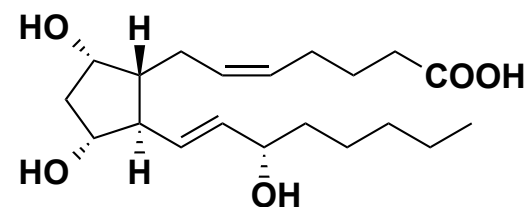
Penicillin



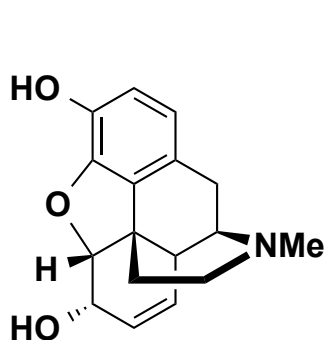
Quinine



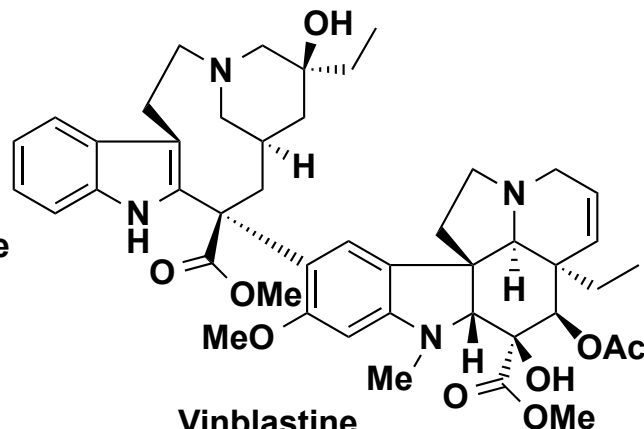
Progesterone



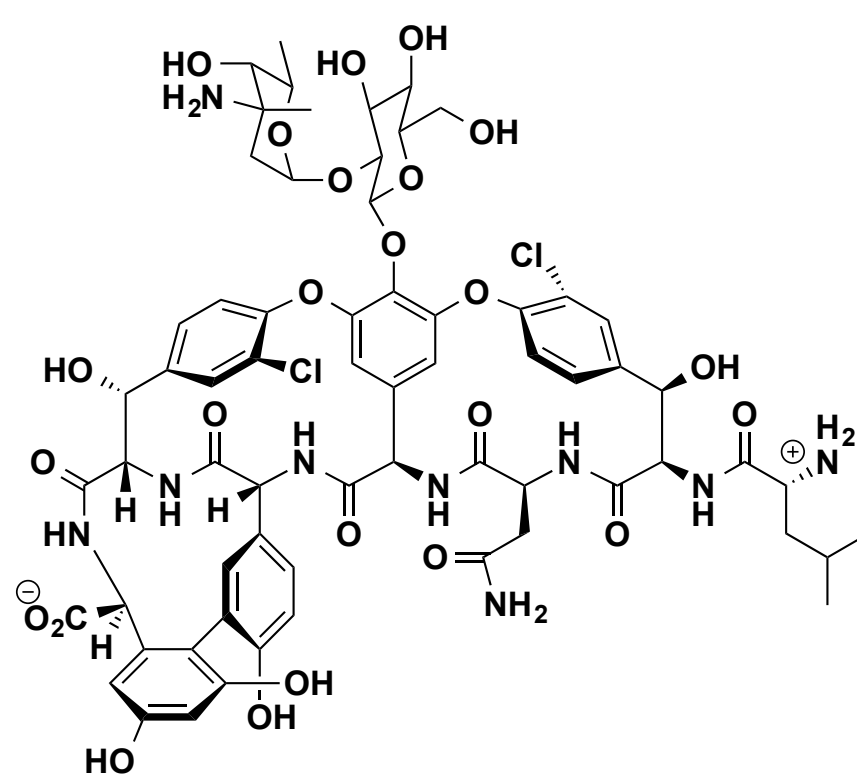
PGF2a



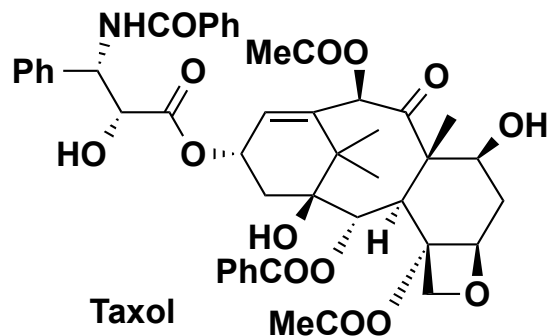
Morphine



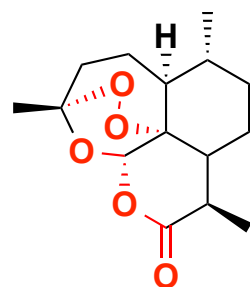
Vinblastine



Vancomycin



Taxol



Artemisinin

# **What is Total Synthesis ?**

**Total Synthesis: Why, What's for, What's after**

# Terms

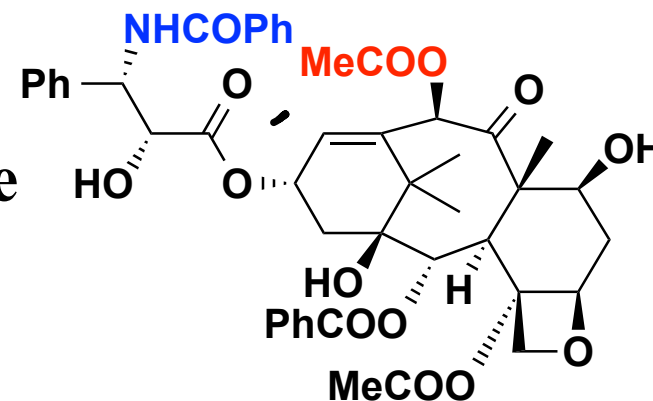
**Total Synthesis:** The production of a molecule from commercially or readily available chemicals through a series of chemical transformations.  
*Not limited to Natural Product*

**Formal total synthesis:** the chemical synthesis of an **advanced** intermediate that has already been transformed into the desired target.

# Terms

## Partial synthesis or semisynthesis:

Designates the synthesis of a given molecule from an **advanced** precursor related to it.

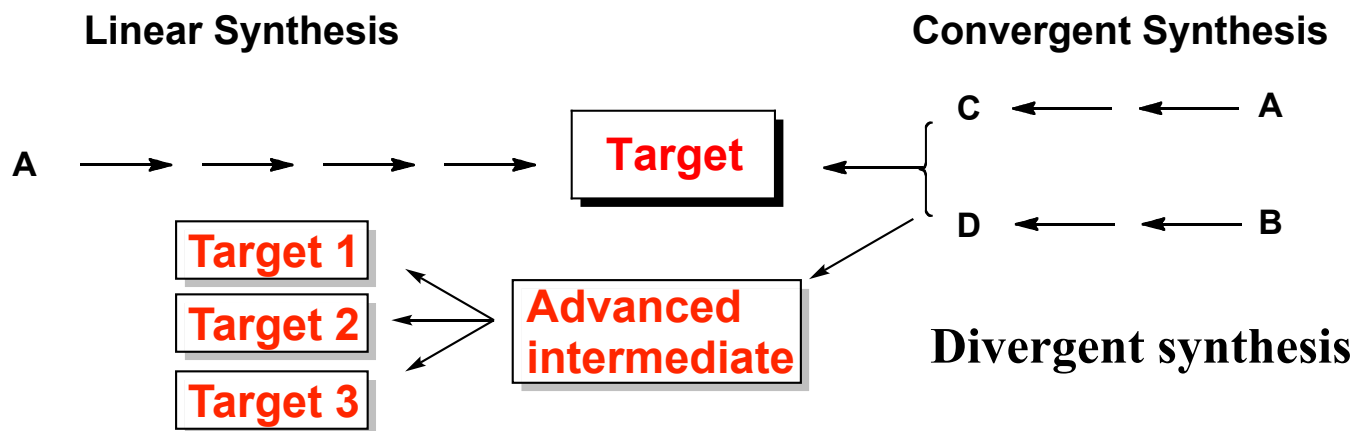


**Relay Synthesis:** defines the process in which an advanced synthetic intermediate is readily accessible by degradation of the target molecule.

# Total Synthesis

The production of a molecule from commercially or readily available chemicals through a series of chemical transformations

Not limited to natural products



**Target-oriented Synthesis**

**Methodology-oriented Synthesis**

**Divergent Synthesis (Unified Strategy)**

**To be the first**

**or**

**To be the best**

## **Total Synthesis: To be the First vs To be the Best**

**The race to be the first to get there is still real**

**The philosophy of being the first wasn't necessary  
prolific to the fundamental discoveries along the way.**

**The imperative of being the first to cross the line  
is somehow down-regulated nowadays.**

**To be the best:**

**A relative term that is difficult to quantify,  
evolves constantly.**

**Key to Synthesis:**

- a) Selection of target  
(structure, bioactivity, physical property)**
- b) *Synthetic strategy***
- c) New reaction, new reactivity**

# Total Synthesis: Why



**Why did you want to climb Mount Everest?**

**George Mallory (Mountaineer): "Because it's there"**

**R. B. Woodward: "Because nobody else could do it"**

**A Paradigm shift:** *"Can we make everything"* became *"How well can we make everything"* or *"Can we impact science (society) on the way making molecules"*



**Climbing Mount Everest:** Became accessible to non-professionals

Mostly *follow the same pathway*

**Total synthesis:** Reserved only to skilled synthetic chemists

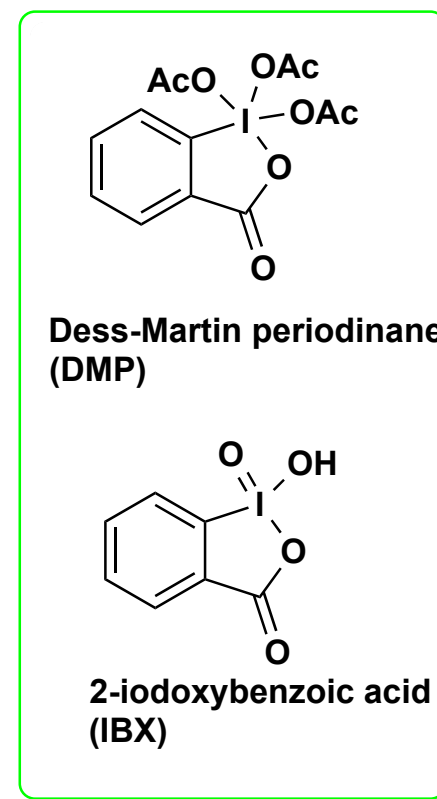
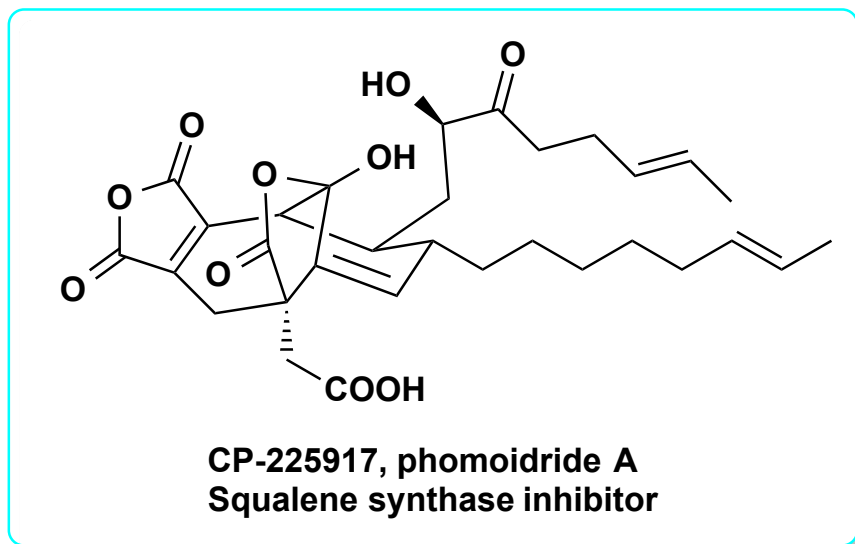
Need constantly to *find new routes* to cross the line

# **Total Synthesis: Why, What's for, What's after**

# Total Synthesis: For the discovery of new reactions

Many new synthetic processes have been discovered as a result of a perceived need in connection with specific problems involving **novel or complicated structures** and a deliberate search for suitable methodology

E. J. Corey (From "Classic in Total Synthesis II" Nicolaou, KC, Snyder, SA)



On the way to the total synthesis of CP molecules, a series of new reactions have been discovered based on the well-known oxidants, DMP and IBX. For a review, see:

Nicolaou, K. C.; Baran, P. S. *Angew. Chem. Int. Ed.* **2002**, *41*, 2678

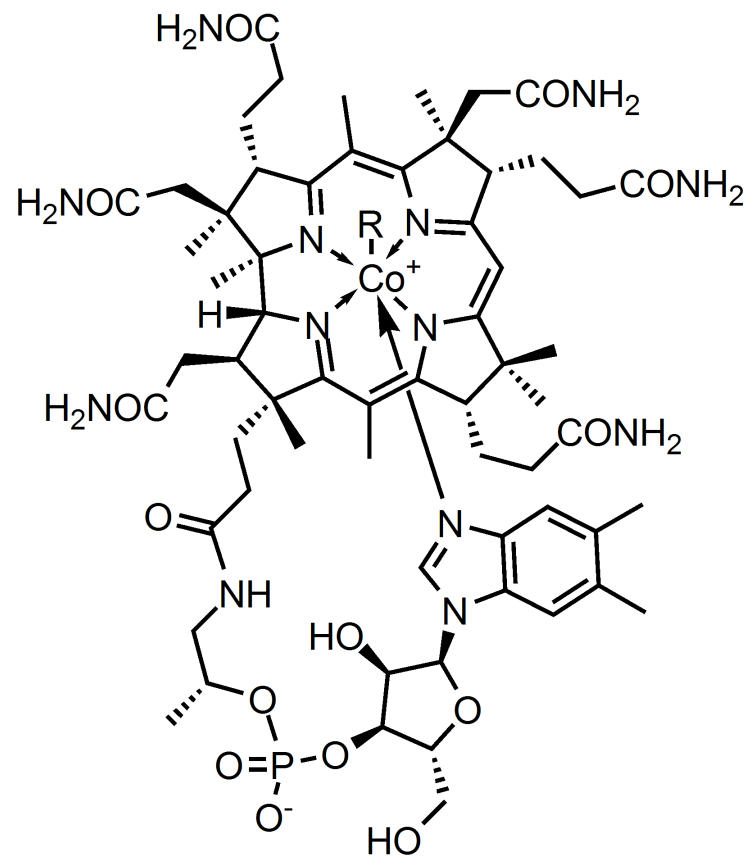
## Total Synthesis: For Serendipity...

*Most of the chemical reactions are discovered, not invented...*

*Even the “simplest” transformation may fail to go as expected in a complex molecular settings. Observation of an unexpected reaction pathway may lead to the development of a new reaction.*

*Something unexpected is bound to happen when you're dealing with the synthesis of a complex natural product.*

# Total Synthesis: For the discovery of new fundamental principles in organic chemistry



R = 5'-deoxyadenosyl, Me, OH, CN

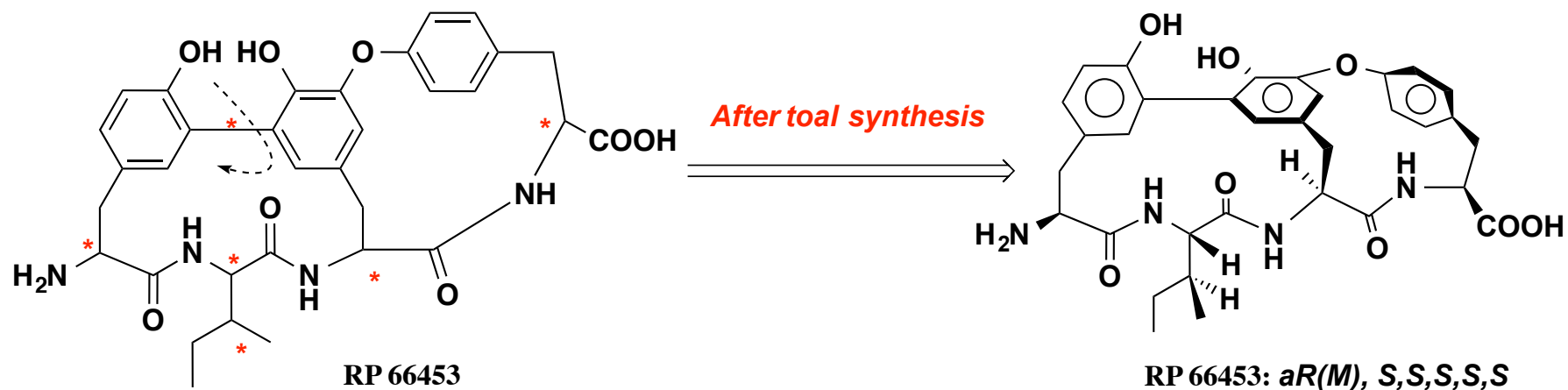
**Total synthesis of vitamin B12  
By Woodward and Eschenmoser**



**Woodward-Hoffman rules**

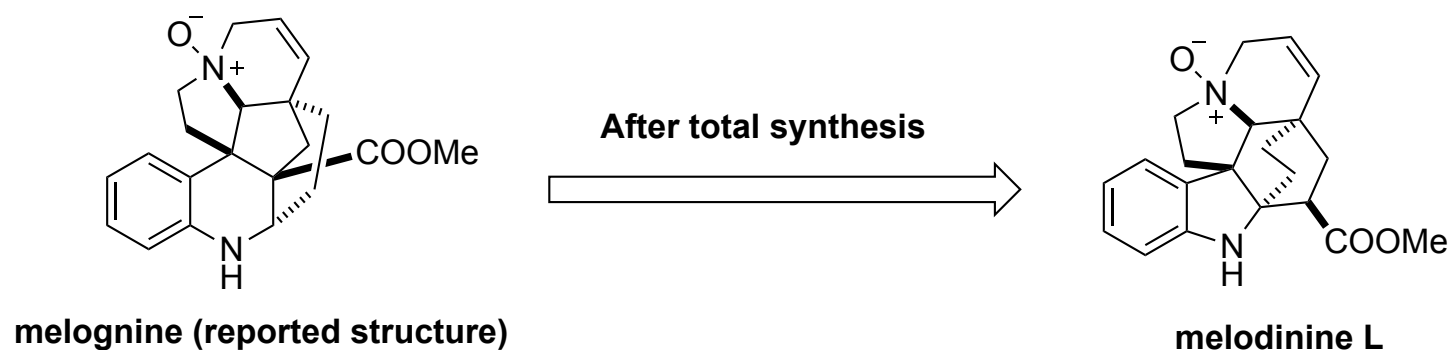
# Total Synthesis: For the Structural Determination

*Less prominent nowadays, but still valide*



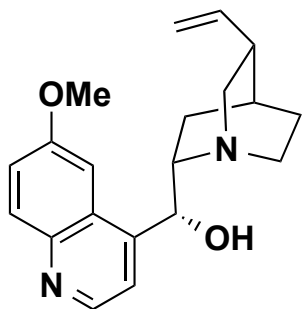
*5 stereocenters, 1 axial chirality*

Bois-Choussy, M.; Cristau, P.; Zhu, J. *Angew. Chem. Int. Ed.* **2003**, *42*, 4238-4241.



Yokoshima, S. *J. Am. Chem. Soc.* **2024**, *146*, 9526.

# Total Synthesis: Impact Science and Society

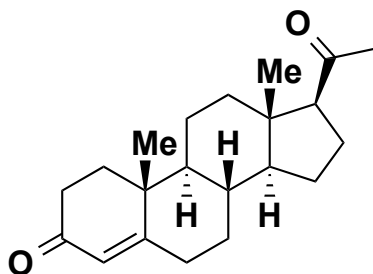


Quinine

**From earlier part of 20th century**

Chemistry of heteroaromatics, quinoline, pyridine  
Organocatalysts

Drug: New synthetic antimalaria agents

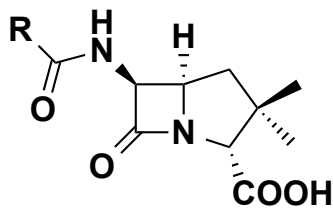


Progesterone

**From 1930's onward**

Conformational analysis  
Synthetic strategies

Drug: Birth control pill



Penicillin

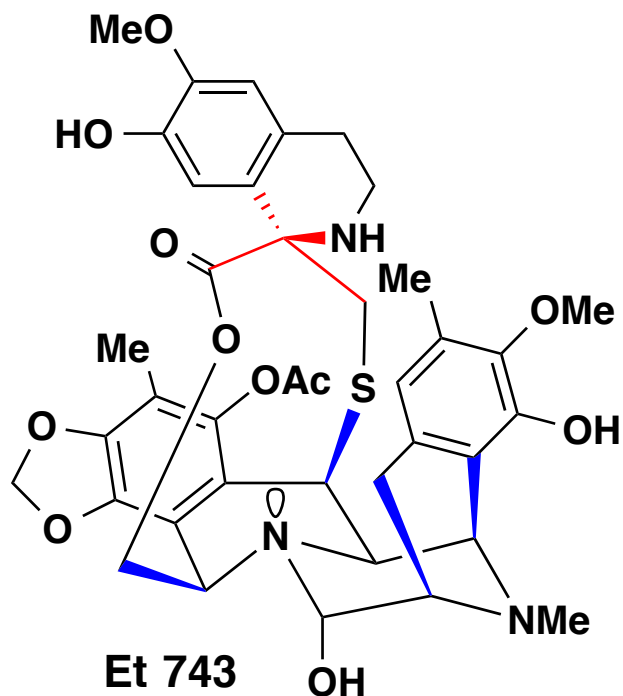
**From 1950's**

$\beta$ -lactam chemistry

Carbodiimide: Revolutionized peptide chemistry

Drug: New synthetic Antibiotics

# Total Synthesis: Impact Society by Solving Supply Issue



**Isolation:** From Caribbean Sea squirt

*Ecteinascidia Turbinate*

**Structural determination:** 1990

**Antitumor activities:** 10-100 times more active than Taxol, Camptothecin...

**Marketing Authorization:** 2007

**Mode of Action:**

DNA-Alkylator

Protein interaction...

Disrupting Protein-DNA interaction

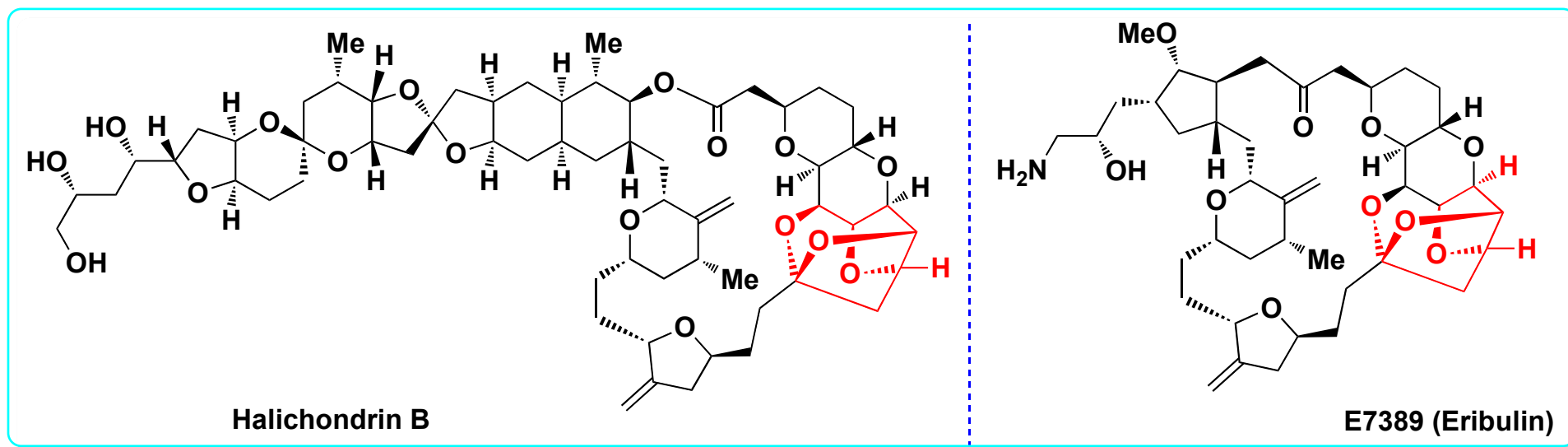
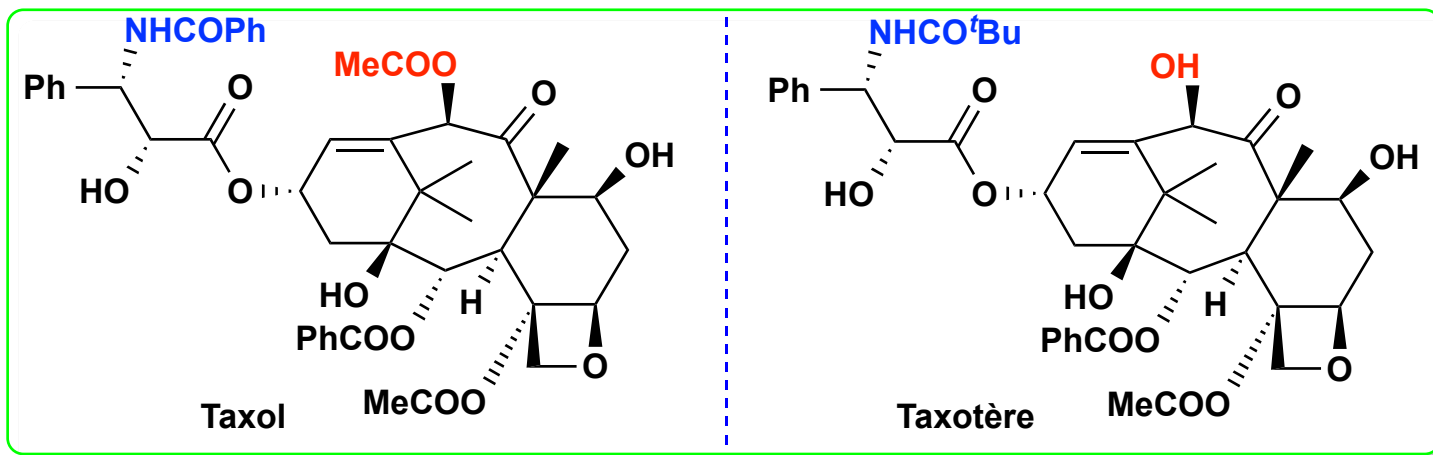
**Isolation yield:** 10<sup>-5</sup>%

**Acquaculture:** feasible but too expensive

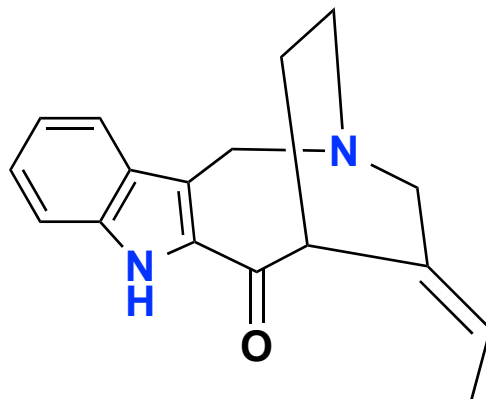
**SYNTHESIS is the only solution**

For a review: Cuevas, C. *Nat. Prod. Rep.* **2009**, 26, 322-337.

# Total Synthesis: Developing (Creating) New Natural Product-Based Drugs



## Total Synthesis: To deepen and Broaden biological studies of a given natural product



**Conolidine**

**Thanks to a total synthesis program, conolidine has been identified to be a potent non-opioid analgesic for tonic and persistent pain.**

**Micalizio, G. C. *Nat. Chem.* 2011, 3, 449-453.**

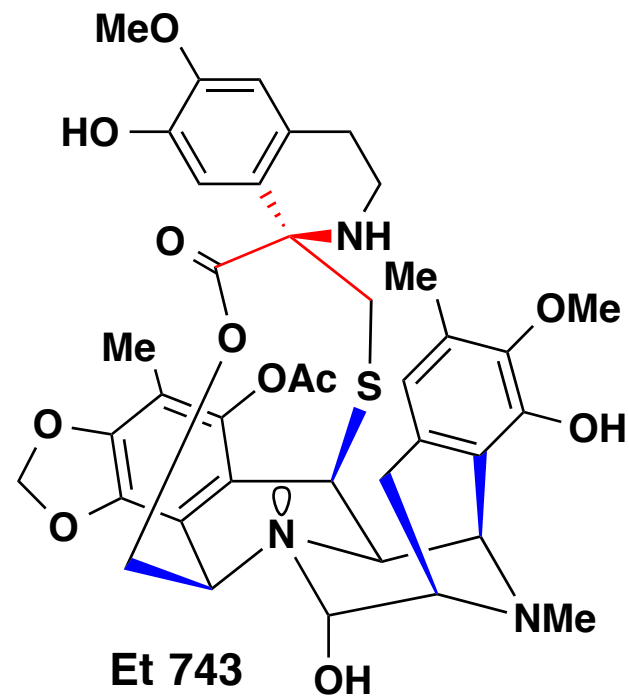
# Total Synthesis: A superb learning and creating process

*“The drug companies love to hire total synthesis people. They figure (correctly) that dealing with the adversity of that work is good training for drug discovery, where most things don't work, either...”*

*For the intellectual challenge and sheer excitement of the endeavor*

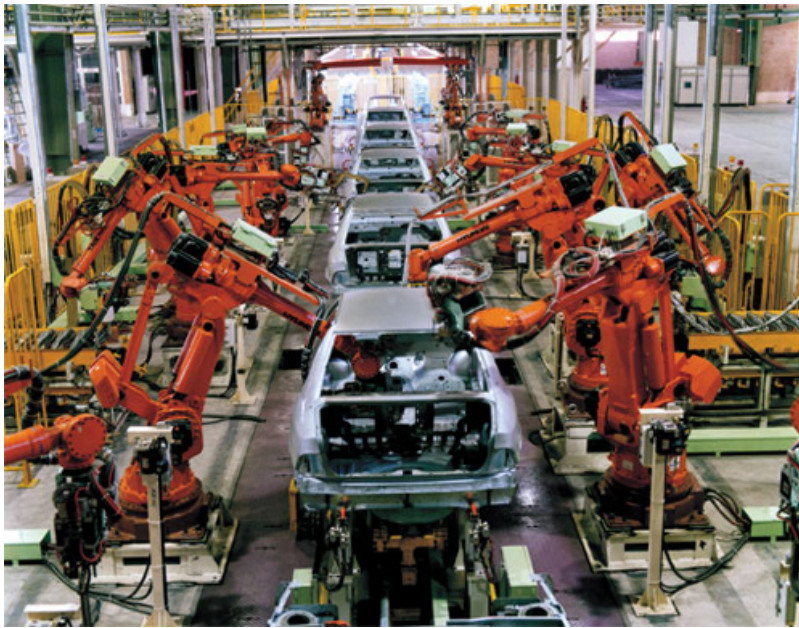
**Total Synthesis:**

**Science Behind Art  
or  
Art Behind Science**



**Yes, but the truth is**  $\implies$

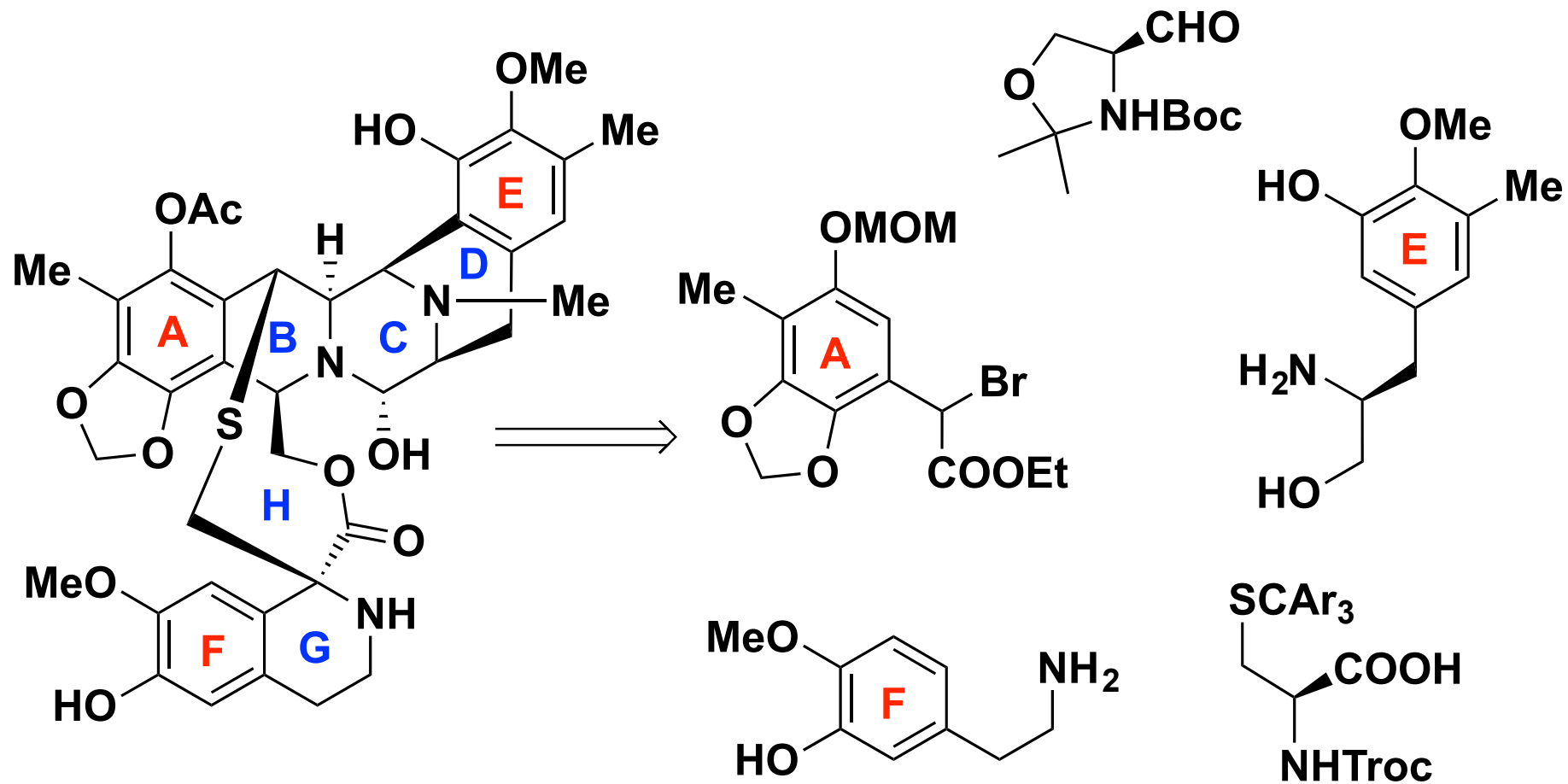
# Total Synthesis: How



**Taj Mahal**

Alper, J. *Science* **1994**, *264*, 1399-1401.

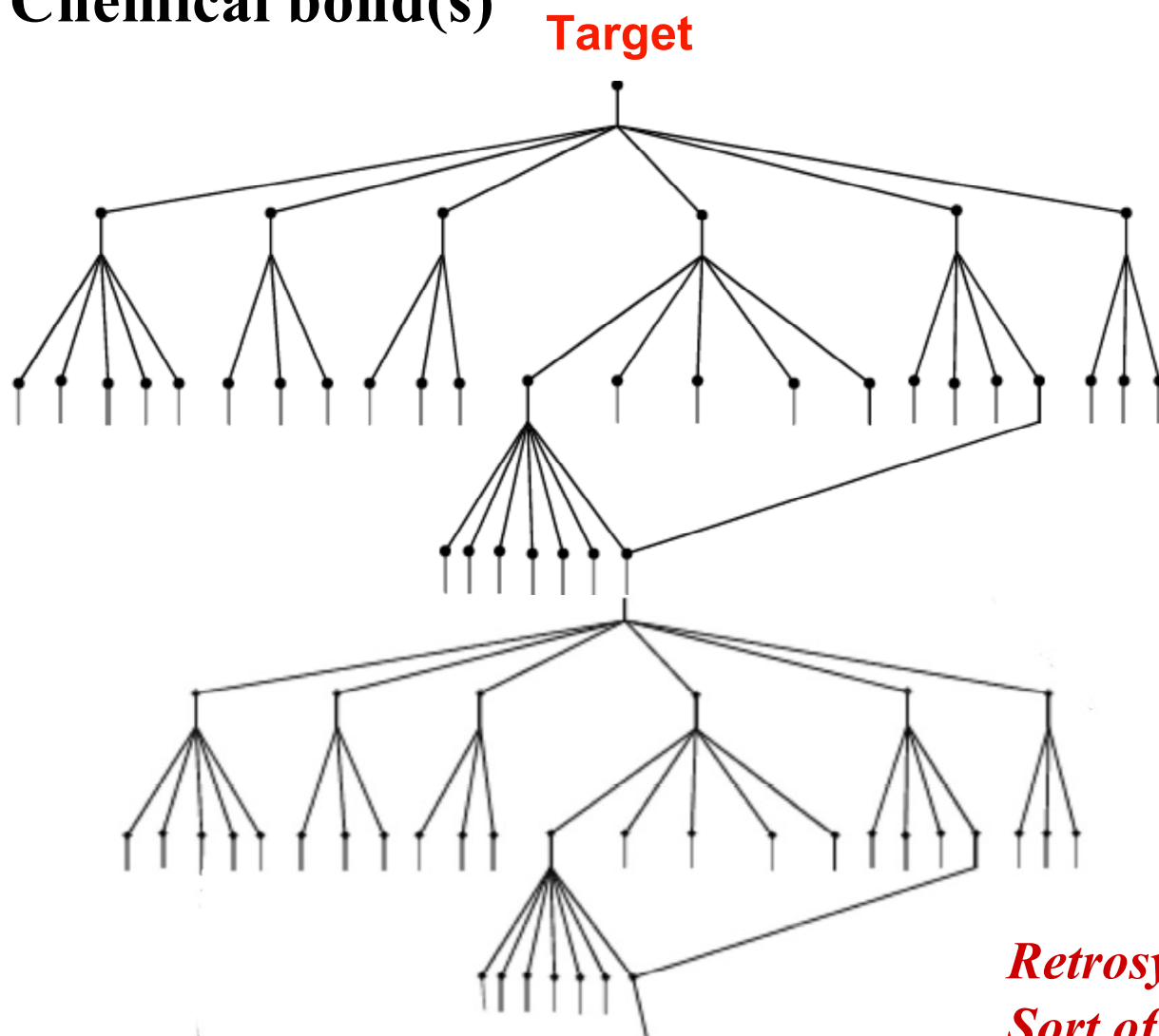
# Total Synthesis: How



J. Chen, X. Chen, M. Bois-Choussy, J. Zhu, *J. Am. Chem. Soc.* **2006**, *128*, 87.

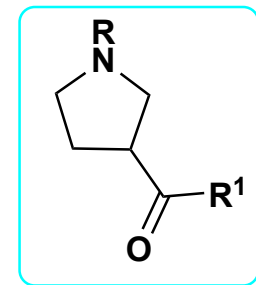
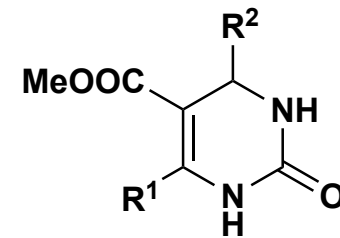
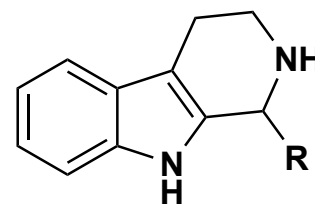
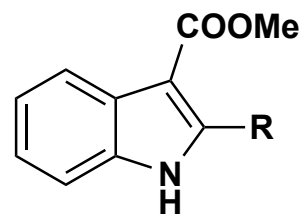
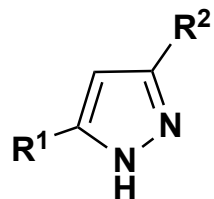
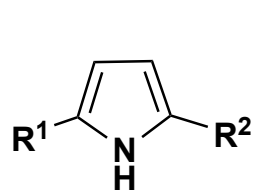
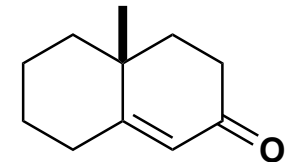
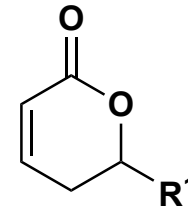
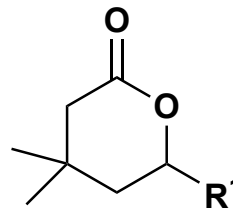
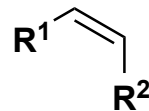
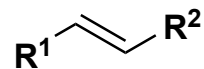
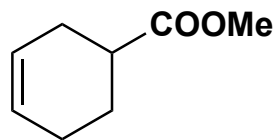
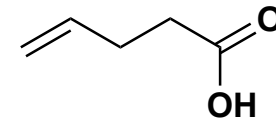
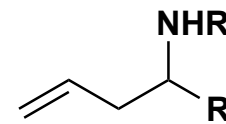
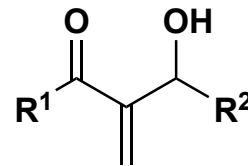
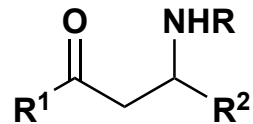
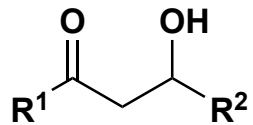
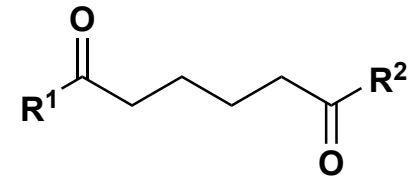
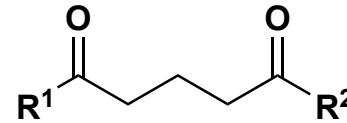
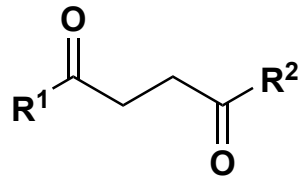
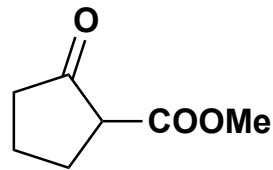
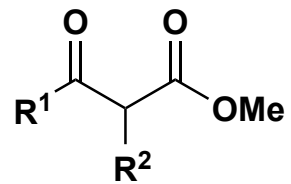
# Logic of Total Synthesis—Retro Synthetic Analysis

Reducing the molecular complexity by disconnecting the  
Chemical bond(s)



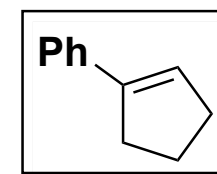
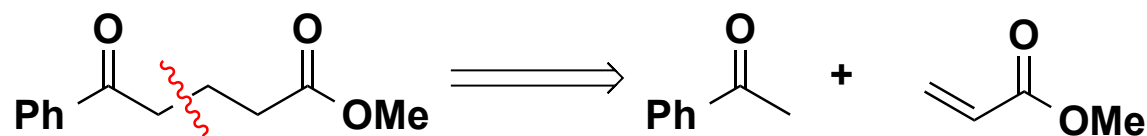
*Retrosynthesis:*  
*Sort of pattern recognition*

# Basic Structure Motifs: Simple examples

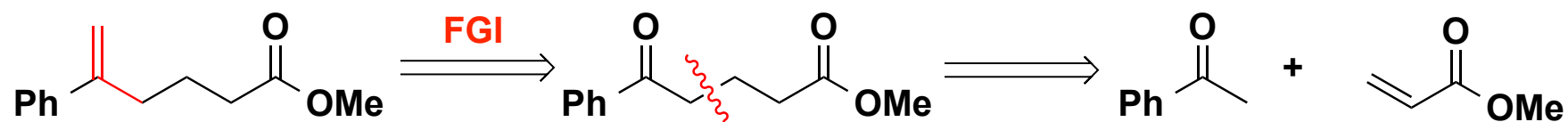


# Retro-Synthetic Analysis: Approaches

## 1. Direct structure simplification



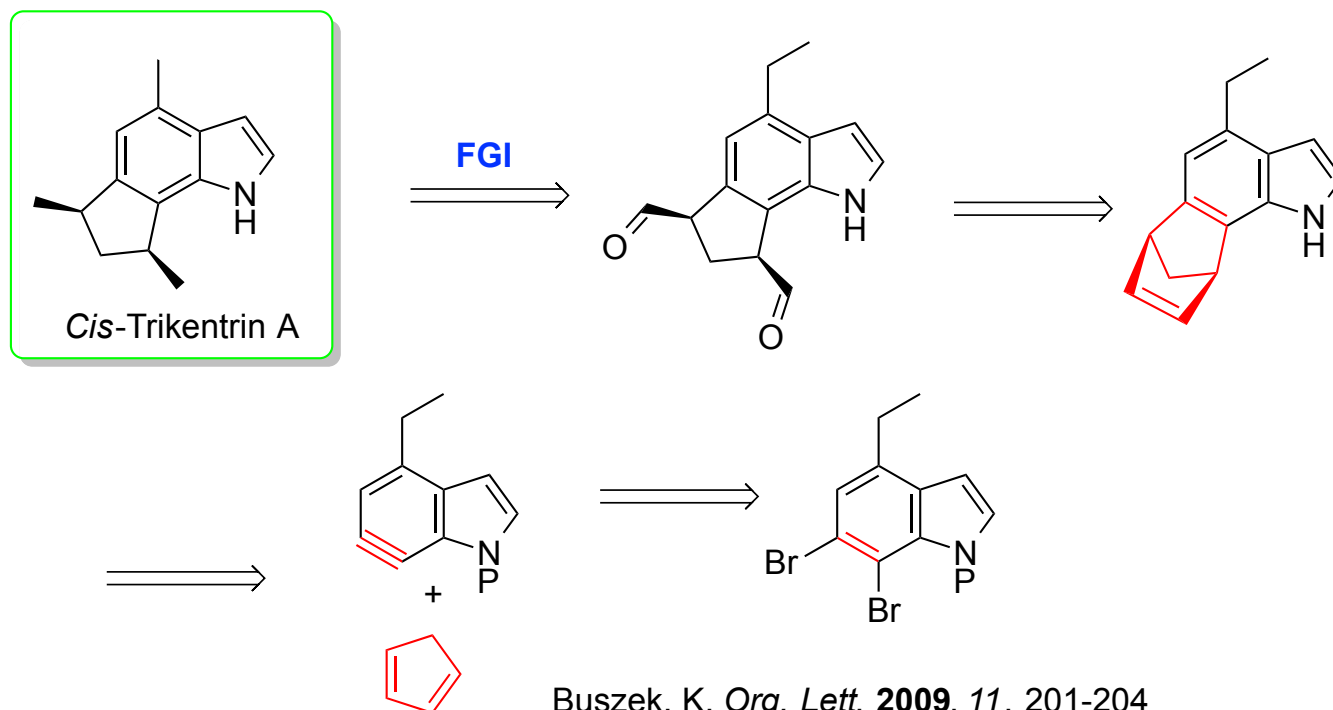
2. **Functional Group Interchange (FGI)** Transforms which bring about essentially no change in molecular complexity but allow the subsequent application of simplifying transforms:



3. Opposite to 1, **increasing structure complexity** includes addition of rings, functional groups (FGA), or stereocenters etc (Pattern recognition).

# Retro-Synthetic Analysis: Approaches

Increasing structure complexity includes addition of rings, functional groups (FGA), or stereocenters etc (Pattern recognition).



# Total Synthesis: Evolution of Criteria

Selectivity (Chemo-, Diastereo-, Enantioselectivity)  
Convergency, Divergency, Overall yields } "High"

## Economy

Atom-Economy, Step-Economy, Redox-Economy

## Free

Protective group-Free, Redox-Free

**Hendrickson's definition of ideal synthesis** : “ ...creates a complex molecule...in a sequence of only construction reactions involving no intermediary refunctionalizations, and leading directly to the target, not only its skeleton but also its correctly placed functionality.”

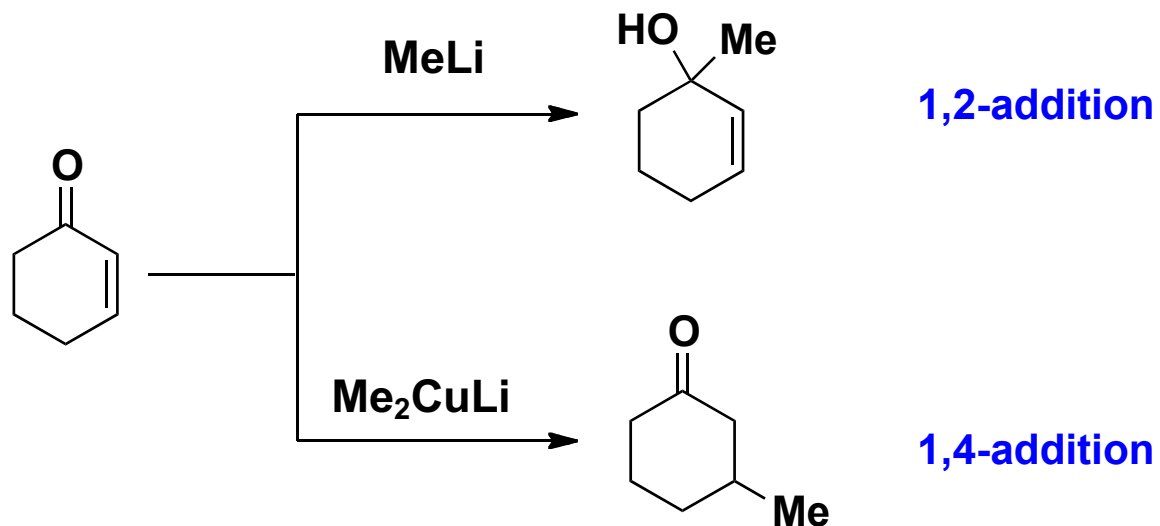
Hendrickson, J. B. *J. Am. Chem. Soc.* 1975, 97, 5784.

***Total synthesis: Be short, selective and high yielding***

# Chemo-, Regio-, and Stereo-selectivity

## Chemo-selectivity

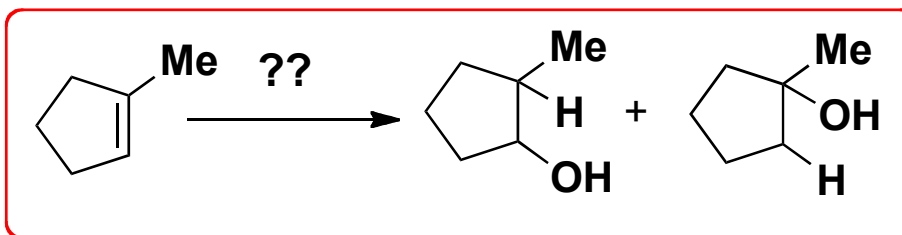
The preferential reaction of a given reagent with one of two or more functional groups



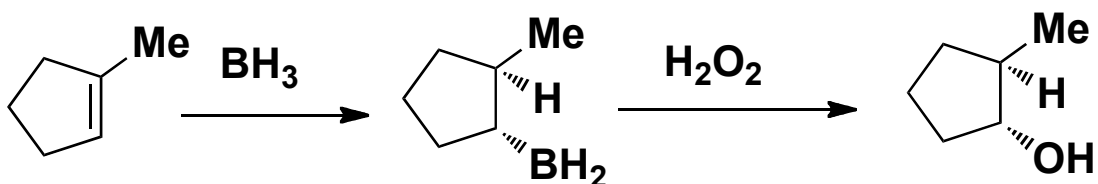
# Chemo-, Regio-, and Stereo-selectivity

## Regioselectivity

The directional preference of breaking or making a chemical bond



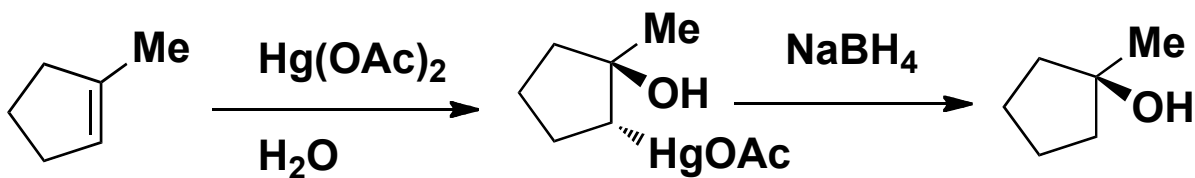
### Hydroboration



*Syn-addition*

*Retention of Stereochemistry*

### Oxymercuration/Reduction

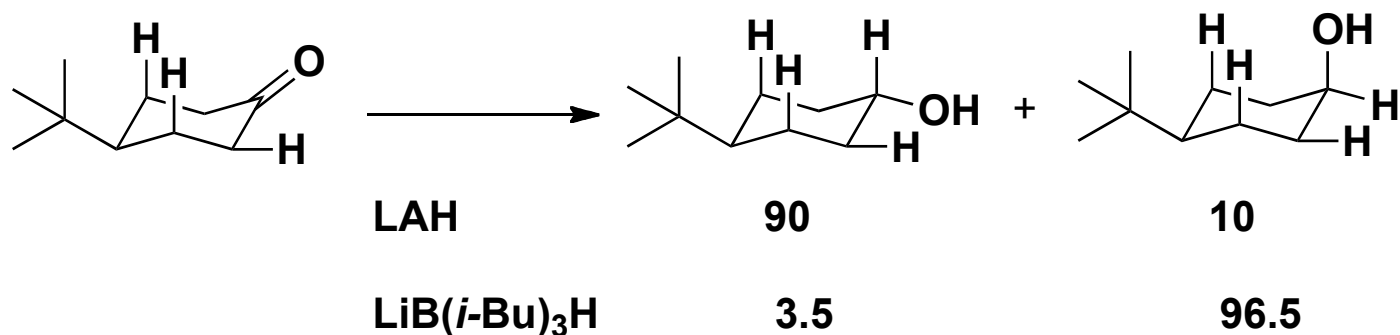


*Anti-addition*

# Chemo-, Regio-, and Stereo-selectivity

## Stereoselectivity

The preferential formation of one stereoisomer over another (diastereo-, enantio-selectivity)



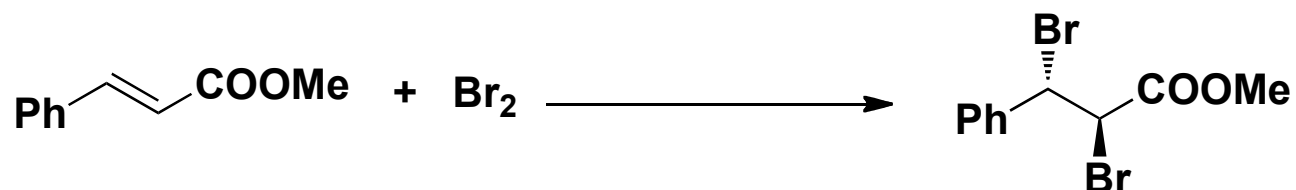
# Atom-Economy

$$\text{Atom-economy\%} = \frac{\text{Mass of Products}}{\text{Mass of Starting Materials}} \times 100$$

*Minimizing waste at the Molecular level*



Y: 100%  
But not atom-economic



Y: 100%  
atom-economic

**Good:** Addition, (cyclo)isomerizations, Pericyclic reactions  
using catalytic amount of reagent if needed. **Ideal:** « Mix and Go »

**Bad:** Substitution reaction, using stoichiometric reagents

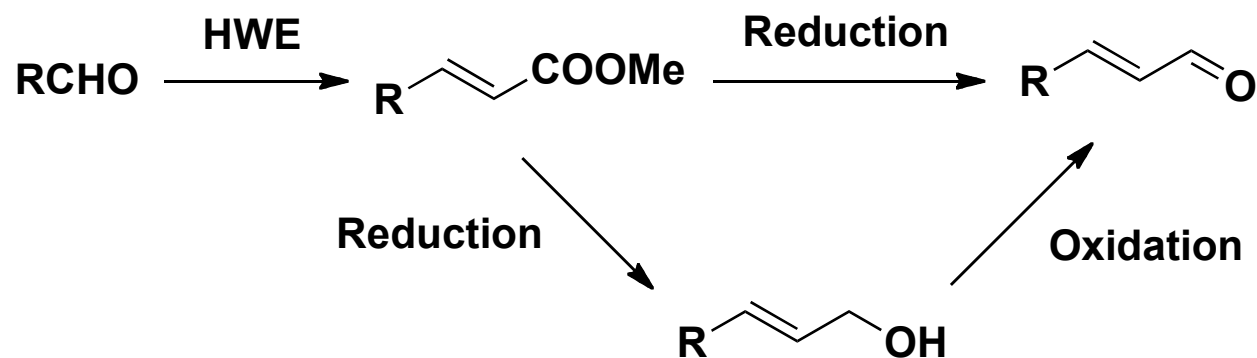
# Redox-Economy

- \* Reducing the number of nonstrategic or corrective Oxidation and Reduction steps
- \* Isohypsic = A synthesis sequence that is devoid of redox steps

## Nonstrategic or corrective Redox reactions:

- increasing the length of synthesis (bad in step economy)
- increasing the waster production
- generally difficult to scale up in industrial settings

**A sequence of low redox-economy (nevertheless “synthetically efficient”)**

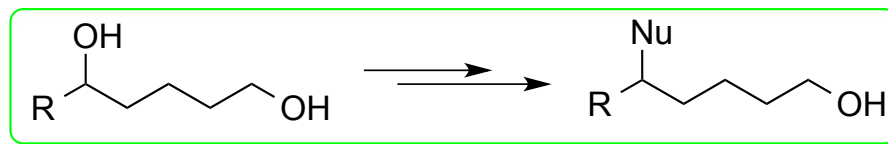


Burns, N. Z.; Baran, P. S.; Hoffmann, R. W. *Angew Chem. Int. Ed.* **2009**, 48, 2854-2867.

# Protecting-Group-Free Synthesis

Many creative synthesis would most likely be impossible without the Assistance of protective groups (PGs). The choice of PGs is part of the Synthesis Art.

**Orthogonality** of PGs is a key issue that one has to consider in planning a synthesis



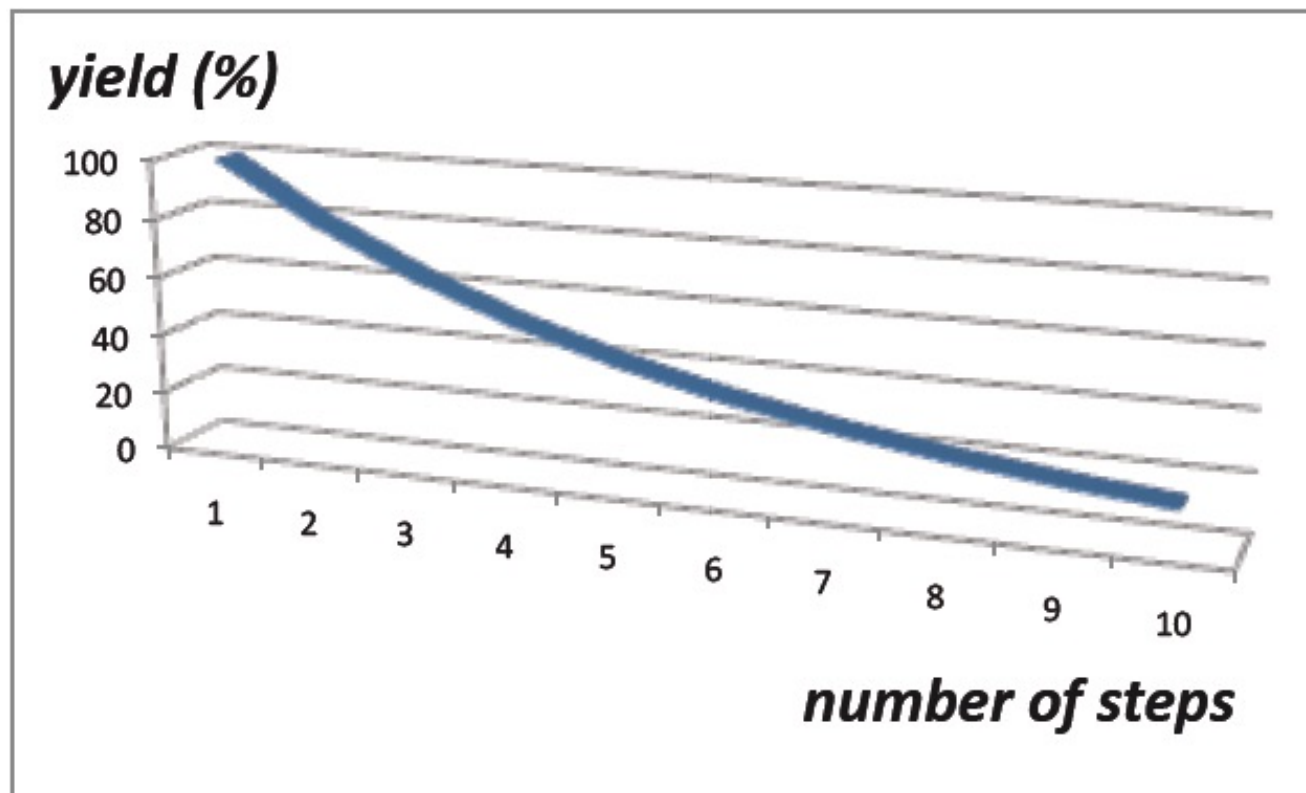
*However using PGs in a synthesis:*

- Add at least two steps (bad *step economy* and increasing waste production)
- Atoms corresponding to the protecting group is lost (bad *atom economy*)
- Could kill a synthesis due to difficulties in removing PGs (worst case but not rare).

The major challenges (in chemistry) are the construction of molecules without using protecting group chemistry and the ability to put molecules together in Fast and efficient ways (R. H. Grubbs, Nobel laureate 2005)

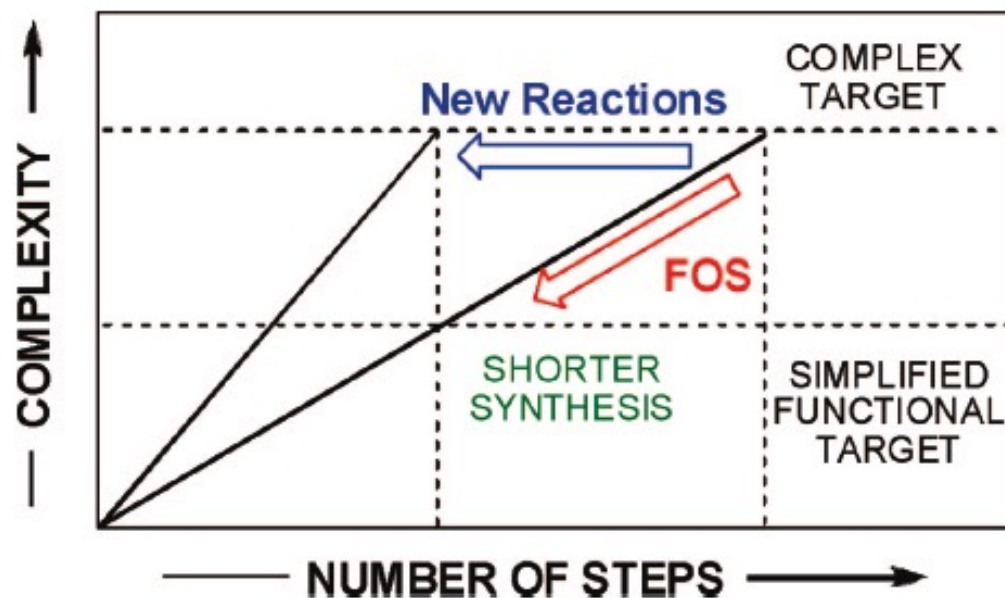
Young, I. S.; Baran, P. S. *Nature Chem.* **2009**, *1*, 193-205.

# Convergency, Divergency, Overall Yield

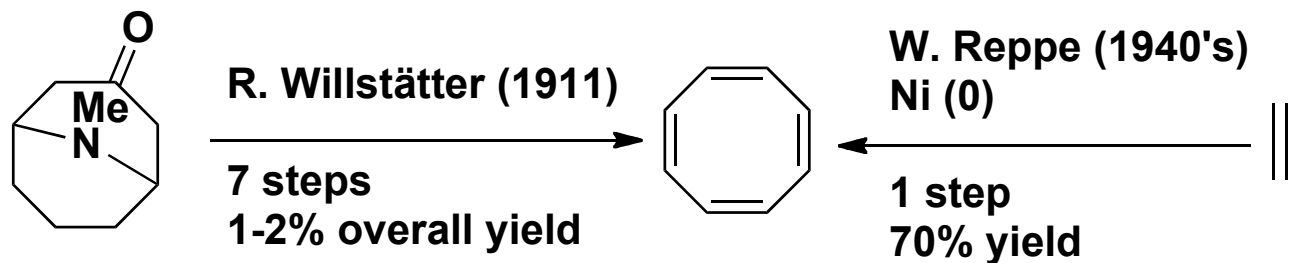


**Assuming each step proceeds in 80% yield, one will  
Lose about 90% of his materials after a 10-step synthesis**

# Step-Economy

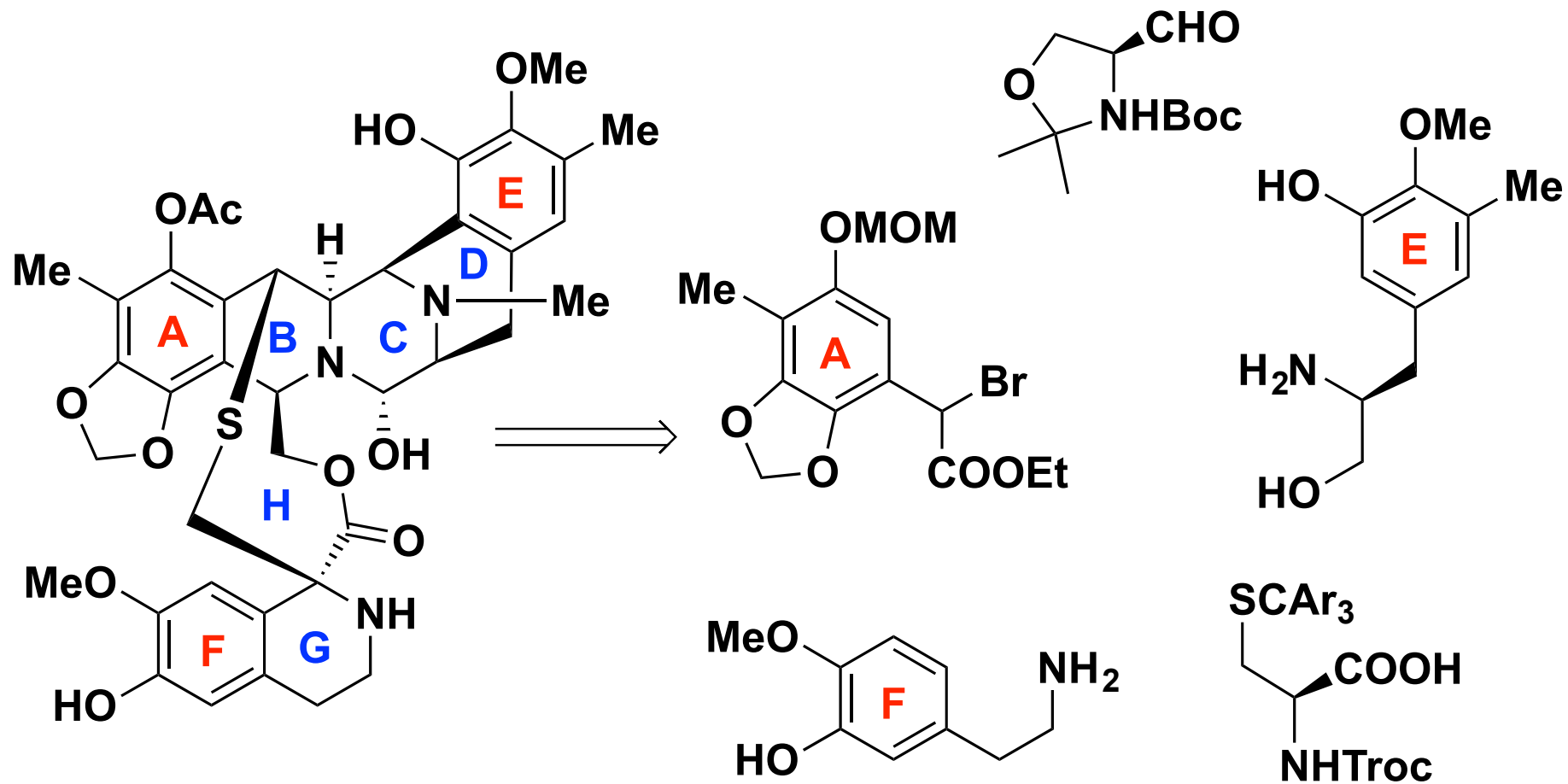


**Increasing Time, Cost efficiency; Reducing Waste production**  
*Need New Reactions, New Synthetic strategies, Tactics*



Wender, P. A. *et al. Acc Chem. Res.* **2008**, *41*, 40-49.

# Retrosynthetic Analysis: Any Guideline



J. Chen, X. Chen, M. Bois-Choussy, J. Zhu, *J. Am. Chem. Soc.* **2006**, *128*, 87.

## “Guidelines for Retrosynthetic Analysis”

- NO universal algorithm for synthesis design (**machine learning re-populized recently**)
- NO universal reactions (well, some are very reliable)
- Targets are too diverse

*Nevertheless some good practices could be helpful*

- Careful background check including biosynthesis pathway, degradation studies, what has been done, **especially what has failed to be productive.**
- **Draw the targets in as many ways as possible**
- Get a feeling for the three-dimensional structure
- **Cleave the easy-forming bonds (e.g. C-X bonds), maximize convergency**
- **Looking for hidden symmetry**
- Looking for multiple-bond forming processes (Dominio and Multicomponent reactions)
- Identify the thermodynamic sink
- Apply the atom-, redox-economy and protecting-group-free principle.
- Pattern recognition: identification of reactivity pattern within the target (moving backward could be rewarding in retrosynthesis)

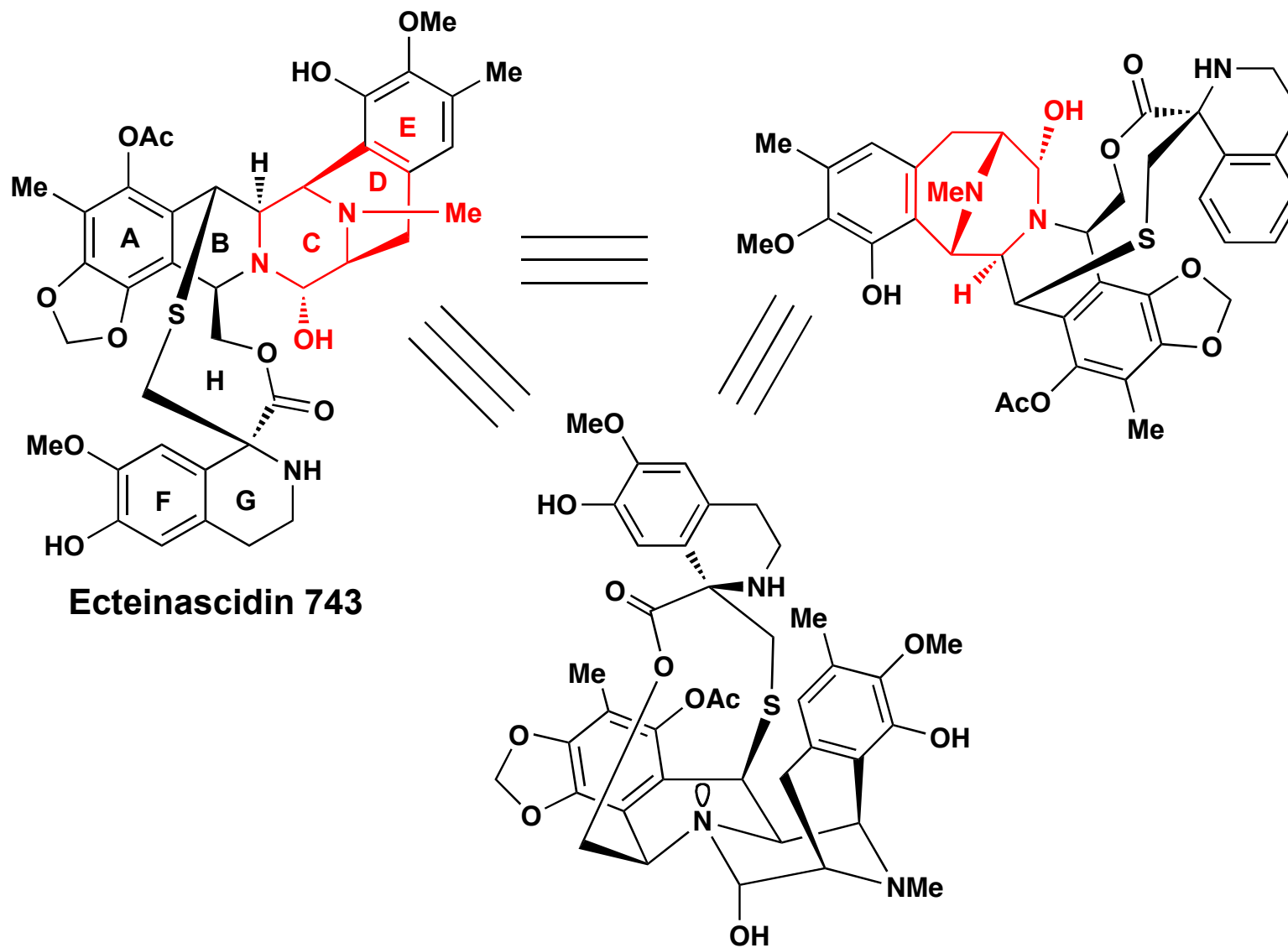
*When the retrosynthetic analysis converges with a feasible starting material, a synthesis plan is at hand.*

**“To study without thinking is futile, To think without studying is dangerous”  
---Confucius**

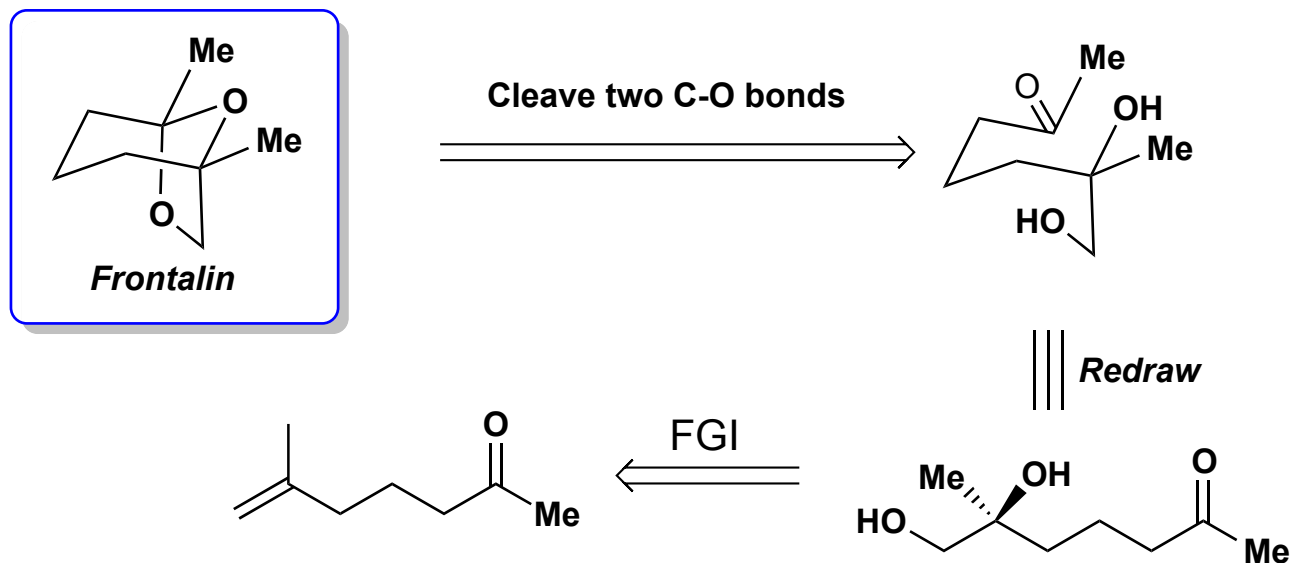
**“If you want to be creative, then you will have to get used to spending most of your time not being creative, to being becalmed on the ocean of scientific knowledge”** <sup>38</sup>

**---Steven Weinberg**

**Draw the target in as many ways as possible: Different drawings gave different impression and possibly different bond disconnection**

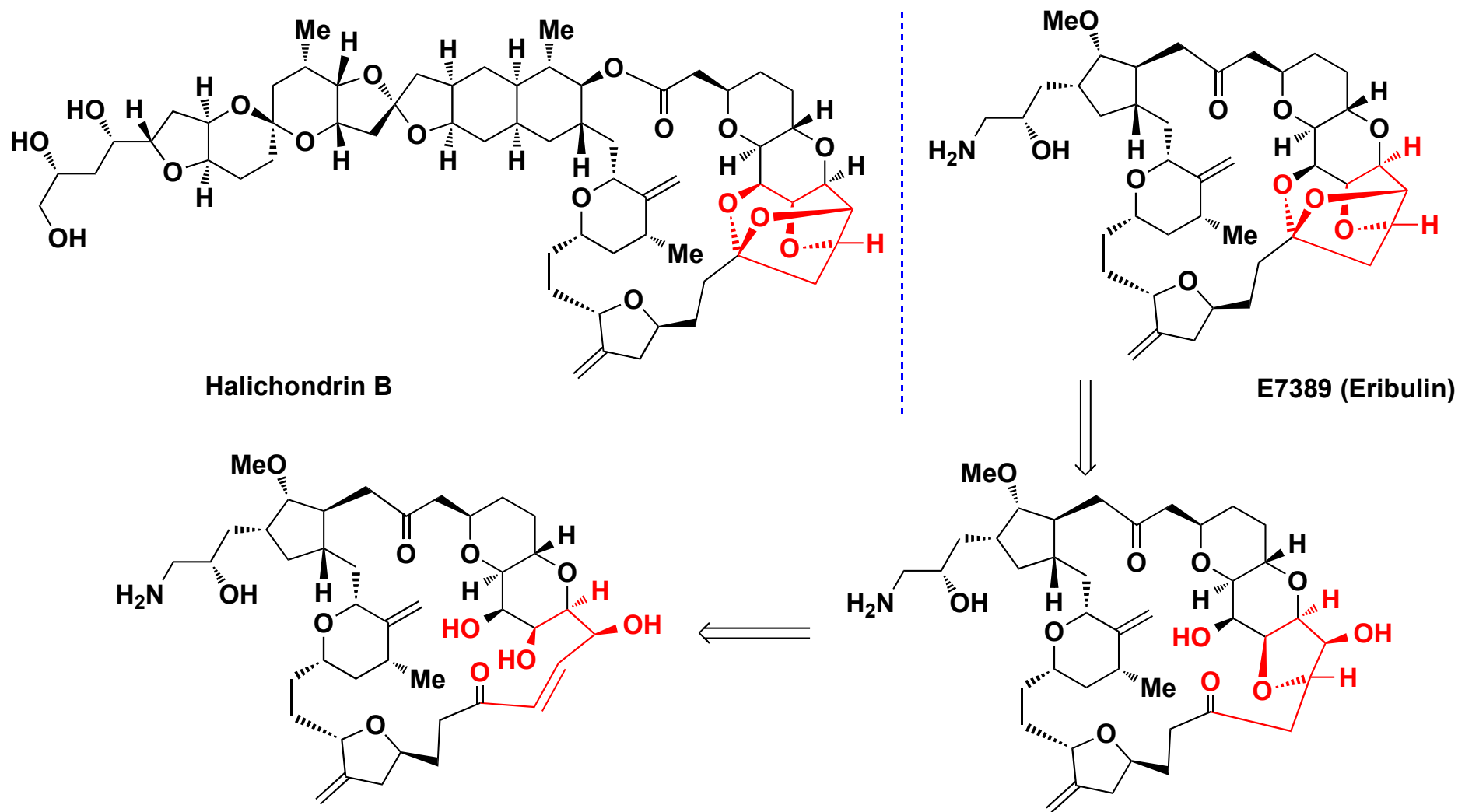


## Cleave the easy-forming bond to simplify the target



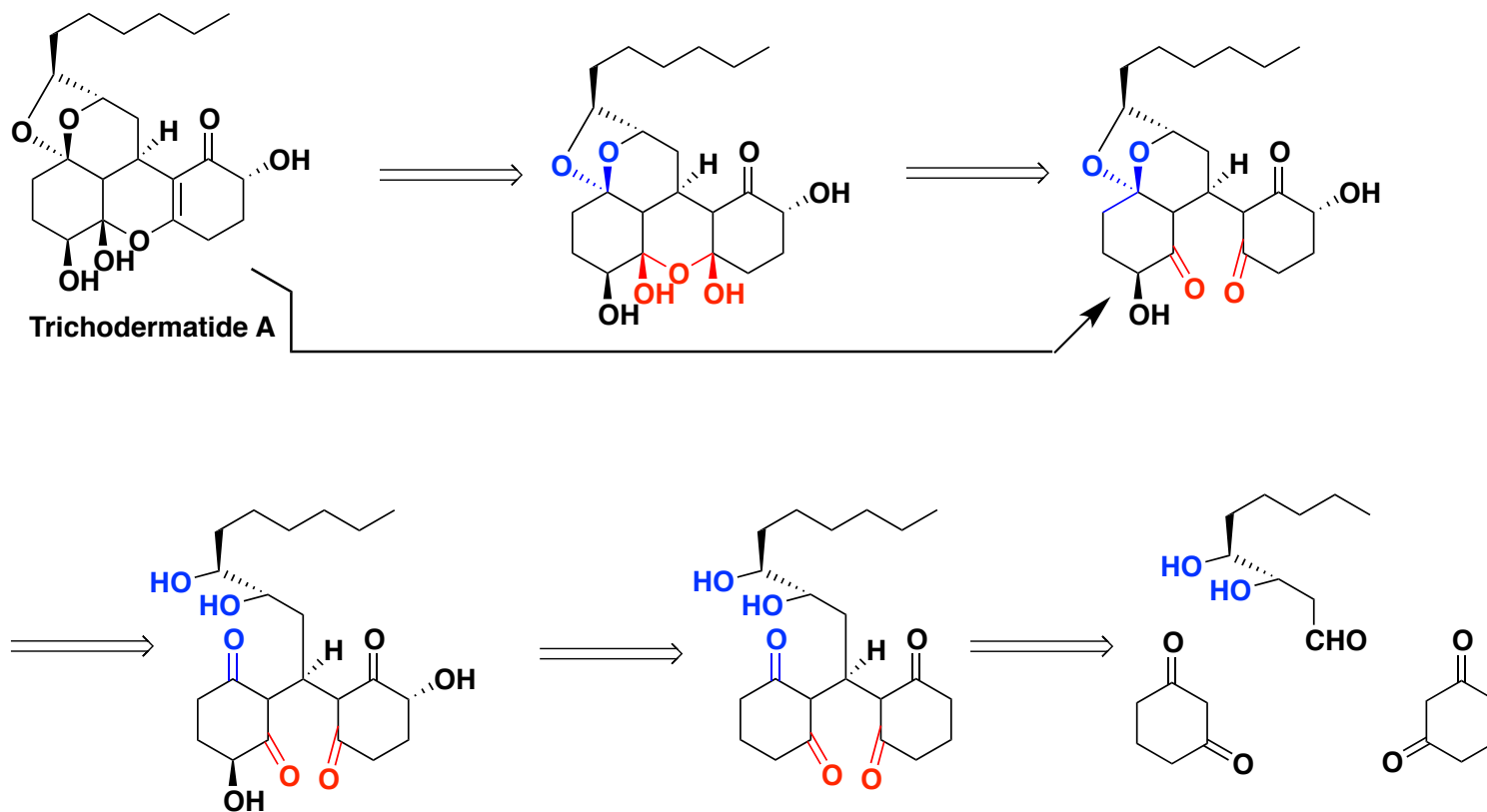
J. A. Turpin; L. O. Weige; *Tetrahedron Lett.* **1992**, 33, 6563.

# Cleave the easy-forming bond to simplify the target



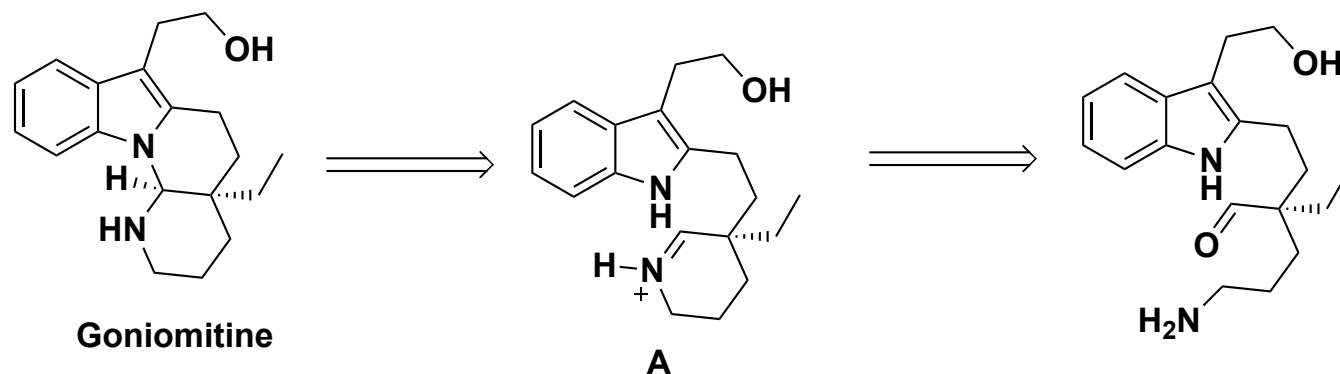
D.-S. Kim, C.-G. Dong, J. T. Kim, H. Guo, J. Huang, P. S. Tiseni, Y. Kishi,  
*J. Am. Chem. Soc.* **2009**, *131*, 15636-15641.

# Cleave the Easiest Bond and Find the Hidden Symmetry



Hiroya, K. *Angew. Chem. Int. Ed.* 2013. 52, 3646-3649.

# Identify the Thermodynamic Sink: Favorable Case



**Kinetic selectivity:** Cyclization of A will give goniomitine as a major product

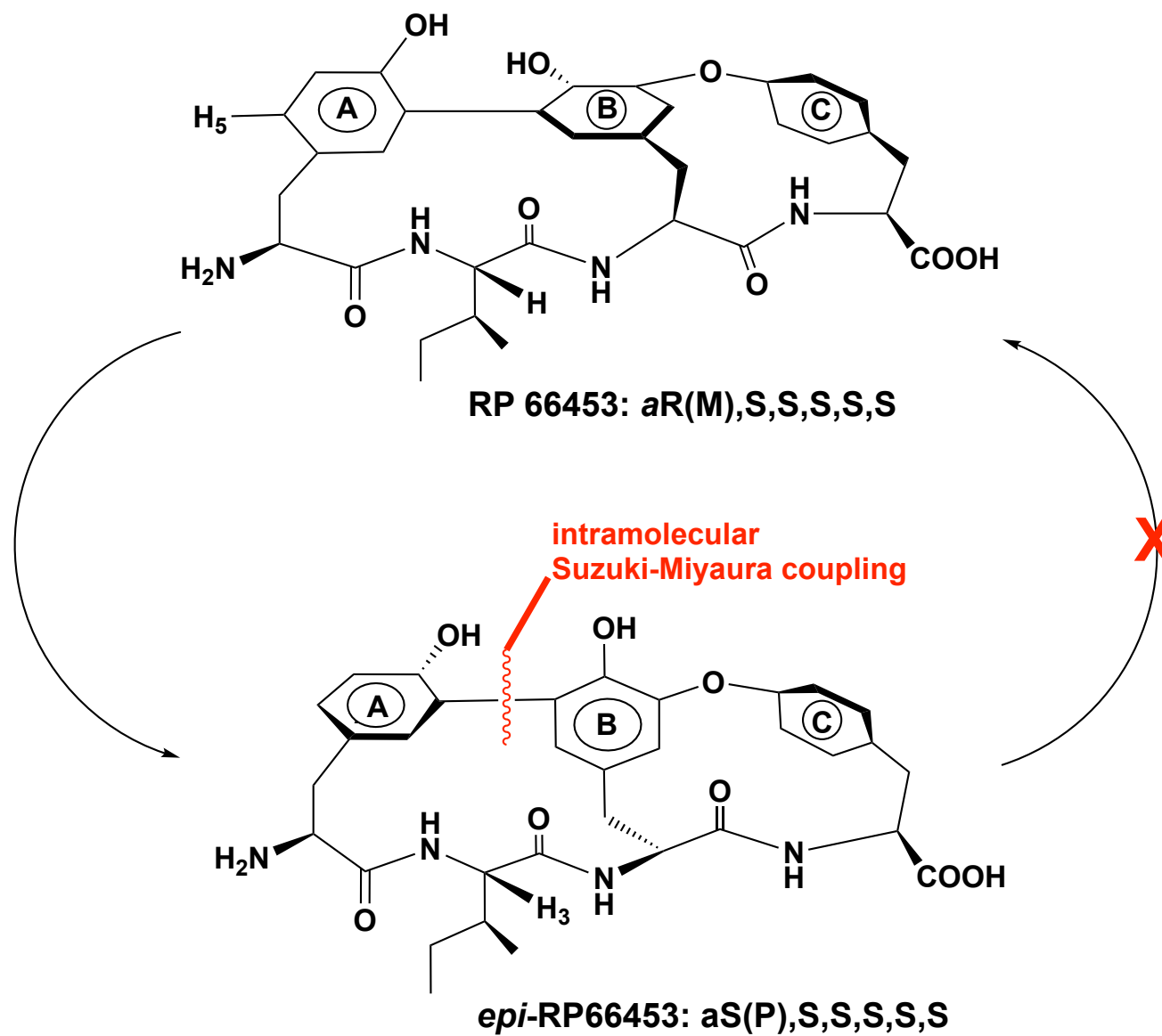
**Thermodynamic selectivity:** Goniomitine is thermodynamically more stable than its epimer

**Conclusion:** a favorable case for chemists...

Takano, S. *J. Chem. Soc. Chem. Commun.* **1991**, 462.

*But: in many case, a thermodynamic sink is difficult to identify though it many very often be one of the reasons that failed your synthesis plan.*

# Thermodynamic Sink: Unfavorable Case



# **“Wrong”, yet still popular terminologies**

## **“Wrong”**

**Asymmetric synthesis**

**Chiral center**

**Chiral HPLC**

**Chiral column**

**Optical purity**

## **“Correct”**

**Enantioselective synthesis**

**Diastereoselective synthesis**

**Stereocenter**

**Enantioselective HPLC**

**Enantioselective column**

**Enantiomeric excess**

a) Mislow, K, Siegel, J. *J. Am. Chem. Soc.* **1984**, *106*, 3319-3328.

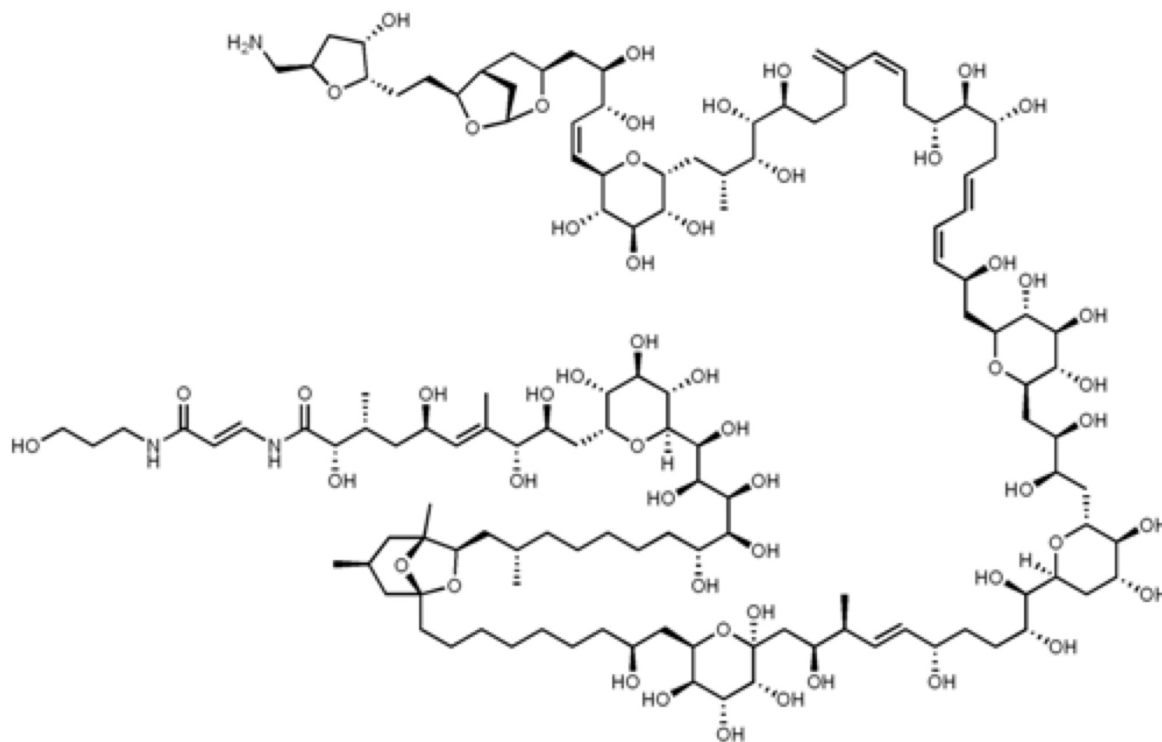
b) Eliel, E. L.: <http://www.uottawa.ca/publications/interscientia/inter.4/eliel/eliel.html>

# Total Synthesis: Where we stand

## Urea Synthesis by Friedrich Wöhler (1828)

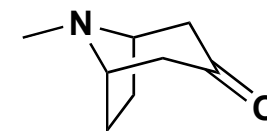


## Palytoxin Synthesis by Y. Kishi (JACS, 1994, 116, 11205)

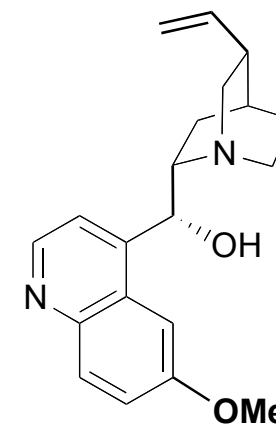


$\text{C}_{129}\text{H}_{223}\text{N}_3\text{O}_{54}$ , 64 Chiral centre,  $2^{64}$  enantiomers possible

*The question is not “Can we make it?” but “How well can we make everything” or “Can we impact science (society) on the way making molecules”*



**Tropinone**  
Robinson 1917



**Quinine**  
Woodward 1940's

# Total Synthesis: Where We Stand

**New Reaction**

**New Strategy**

**Short and Simple**

*The shorter the distance, the stronger the punch can be*

**By Bruce Lee**

# Total Synthesis: Molecular Complexity of Target

**Size**

**Topology**

**Stereochemistry**

**Functionality**

**Appendage**

# Reference books

## Basic:

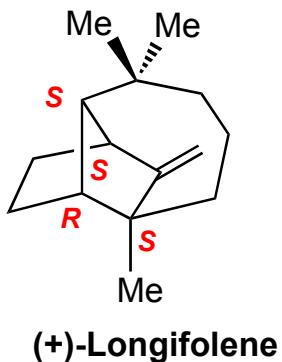
**“Organic Synthesis, The Disconnection Approach”**  
and **“Workbook for Organic Synthesis, The Disconnection Approach”**  
by Warren, S.; Wyatt, P., 2<sup>nd</sup> Ed. Wiley 2008

## Advanced:

**“The Logic of Chemical Synthesis”**  
by Corey, E. J.; Cheng, X-M. Wiley and Son 1989  
**“Classics in Total Synthesis: Targets, Strategies, Methods”**, Wiley  
by Nicolaou, K. C. Sorensen, E. J., Wiley VCH 1996  
**“Classics in Total Synthesis II: Targets, Strategies, Methods”**, Wiley  
by Nicolaou, K. C. Snyder, S., Wiley VCH 2003  
**“Classics in Total Synthesis III”**  
by Nicolaou, K. C.; Chen, J. S., Wiley VCH 2011

And more importantly: *Original publications of each total synthesis*

# Longifolene

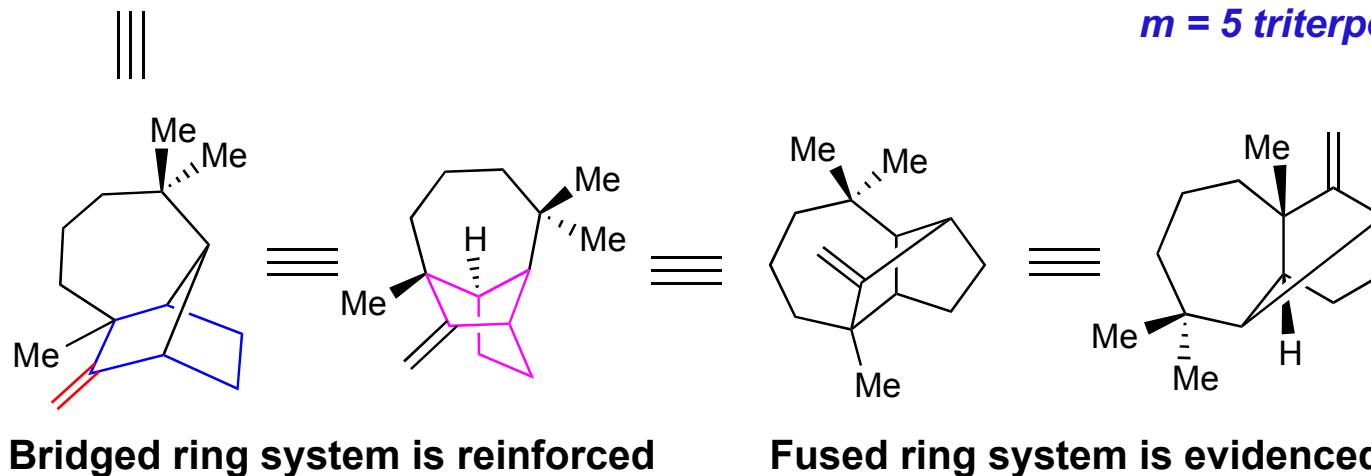


Isolation: G. Ourisson, *Bull. Soc. Chim. Fr.*, 1955, 895  
from "*pinus longifolia*", a **sesquiterpene**

Used as fragrance

(-)-logifolene also exist in Nature  
from Fungi and liverworts

*Formular of terpenes:  $(C_5H_8)_m$*   
*m = 2, monoterpene*  
*m = 3 sesquiterpene*  
*m = 4 diterpene*  
*m = 5 triterpene*



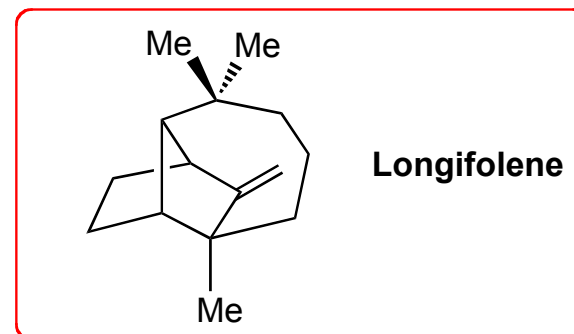
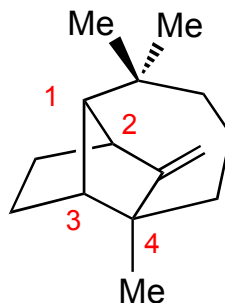
**Using Molecular Models!!!**

***Corey's 1961 synthesis of longifolene « officially » introduced the principles of retrosynthetic analysis!***

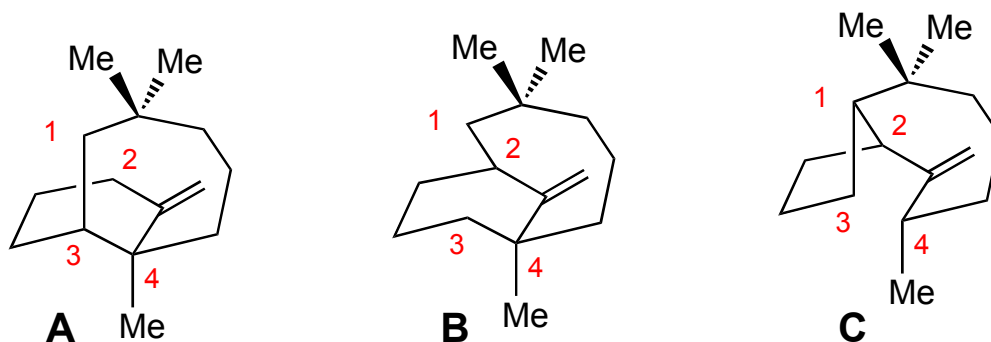
# Longifolene—Corey 1961

## Step 1:

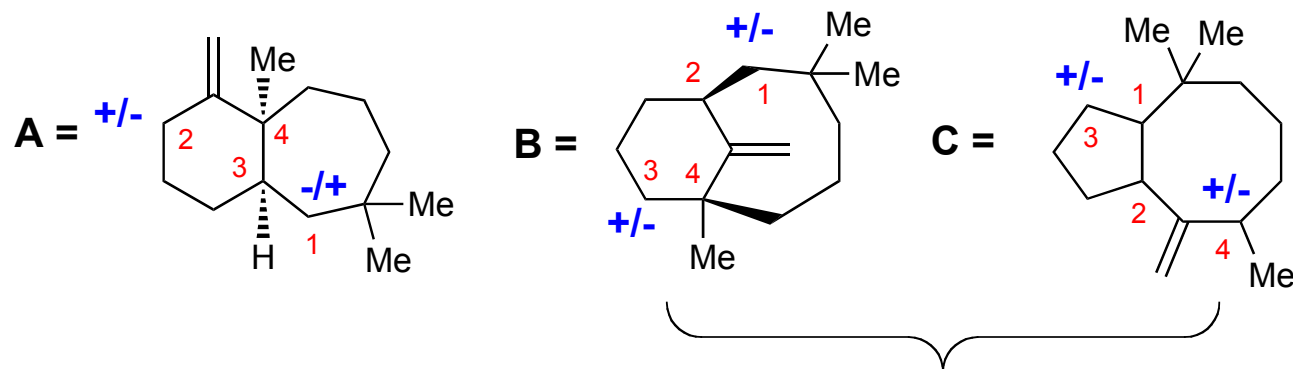
Number the bridge atom



Step 2: Disconnect the chemical bond between the bridge atoms

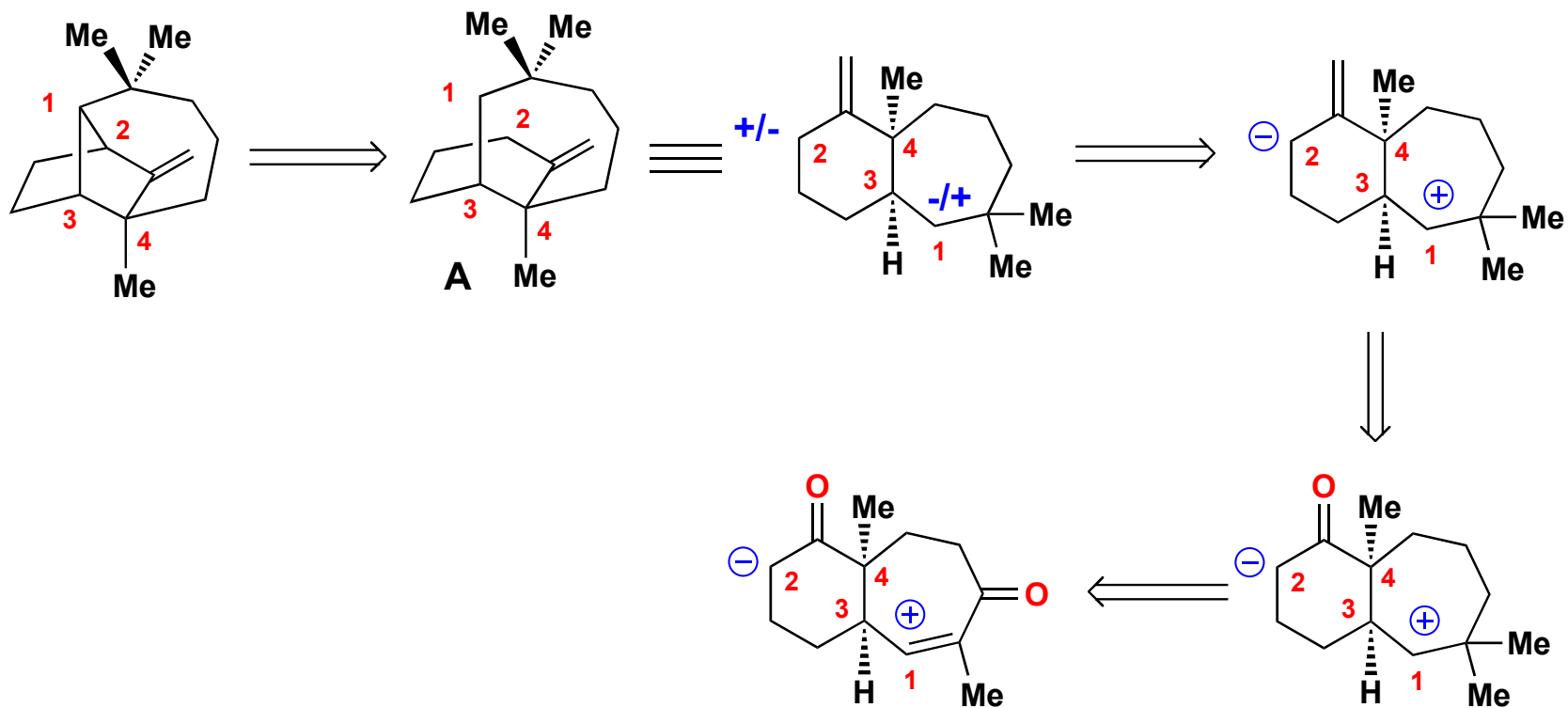


Step 3: Redraw the synthons to make them more familiar to you

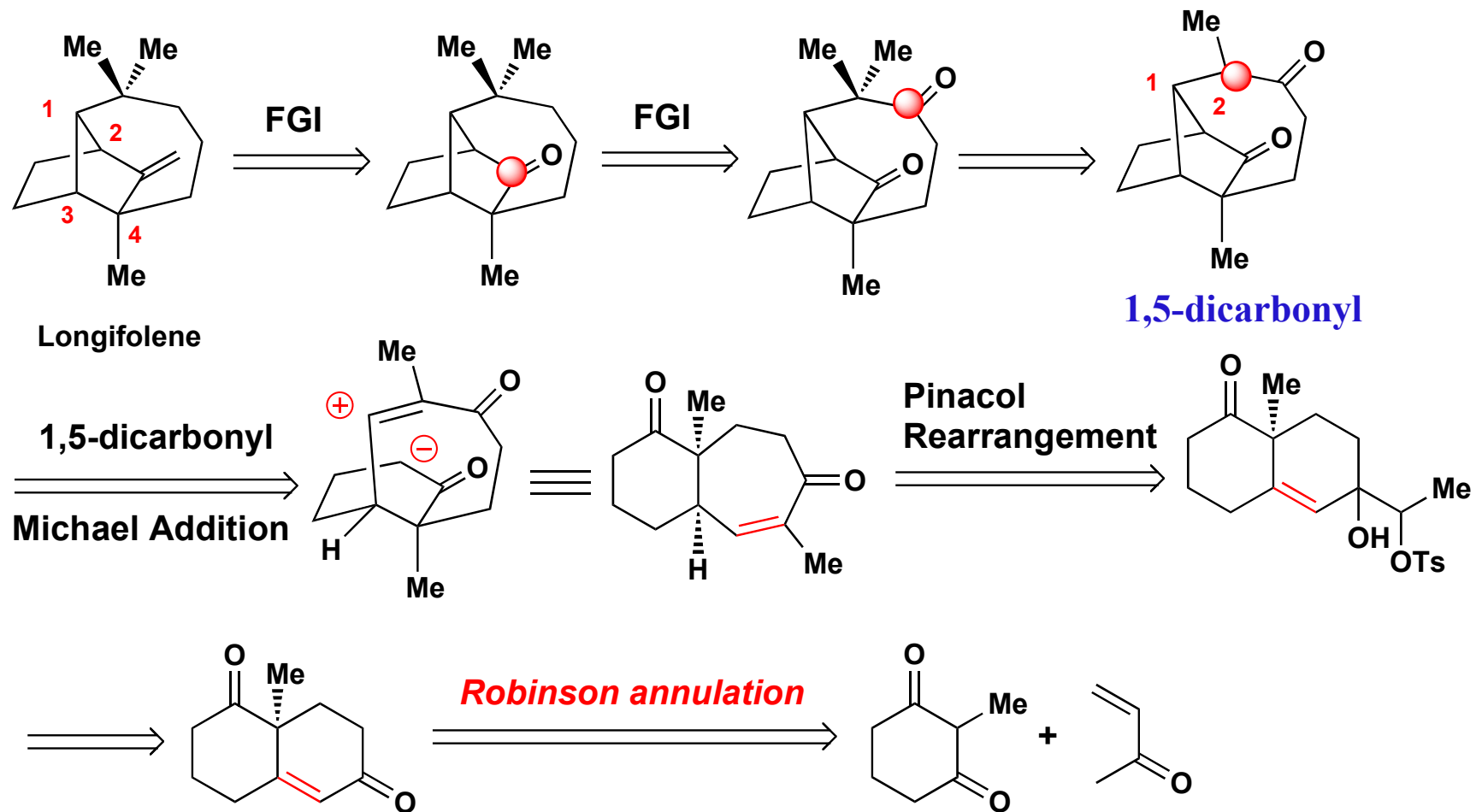


8-membered ring, acces difficult

# Longifolene—Corey 1961



# Longifolene—Corey 1961



## Wieland-Miescher ketone

Wieland, P.; Miescher, K. *Helv. Chim. Acta* **1950**, 33, 2215.

Corey, E. J. et al *J. Am. Chem. Soc.* **1961**, 83, 1251-1231; **1964**, 86, 478-485.

Rev on Wieland-Miescher Ketone: Bradshaw, B.; Bonjoch, J. *Synlett* **2012**, 23, 337-356.

## **Reactions Needed to Know**

**Robinson Annulation (Michael/aldol cascade)**

**Wittig reaction**

**Michael addition: For the synthesis of 1,5-dicarbonyls**

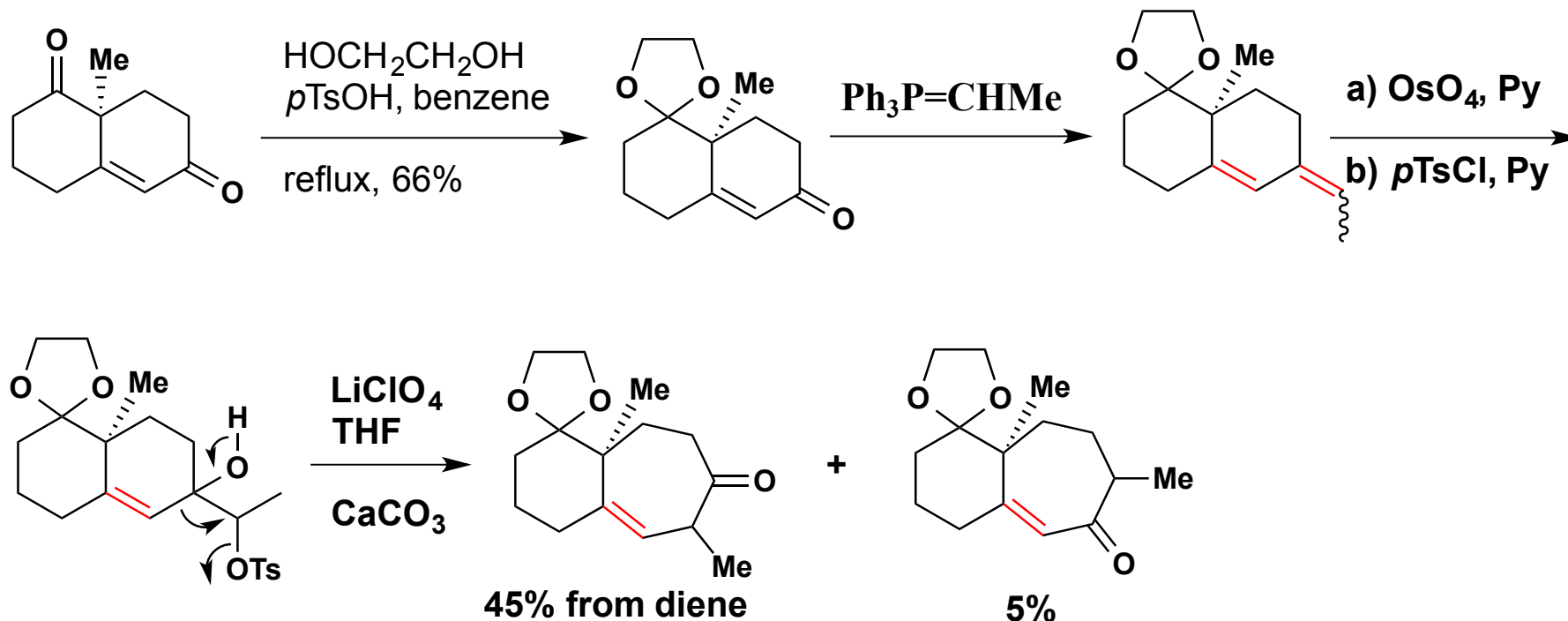
**Dihydroxylation of Alkene with OsO<sub>4</sub>**

**Pinacol Rearrangement: Chemo- and Regio-selectivities**

**Enolate Alkylation: Regioselectivity**

**Reactivity of Carbonyl Group**

# Longifolene—Corey 1961

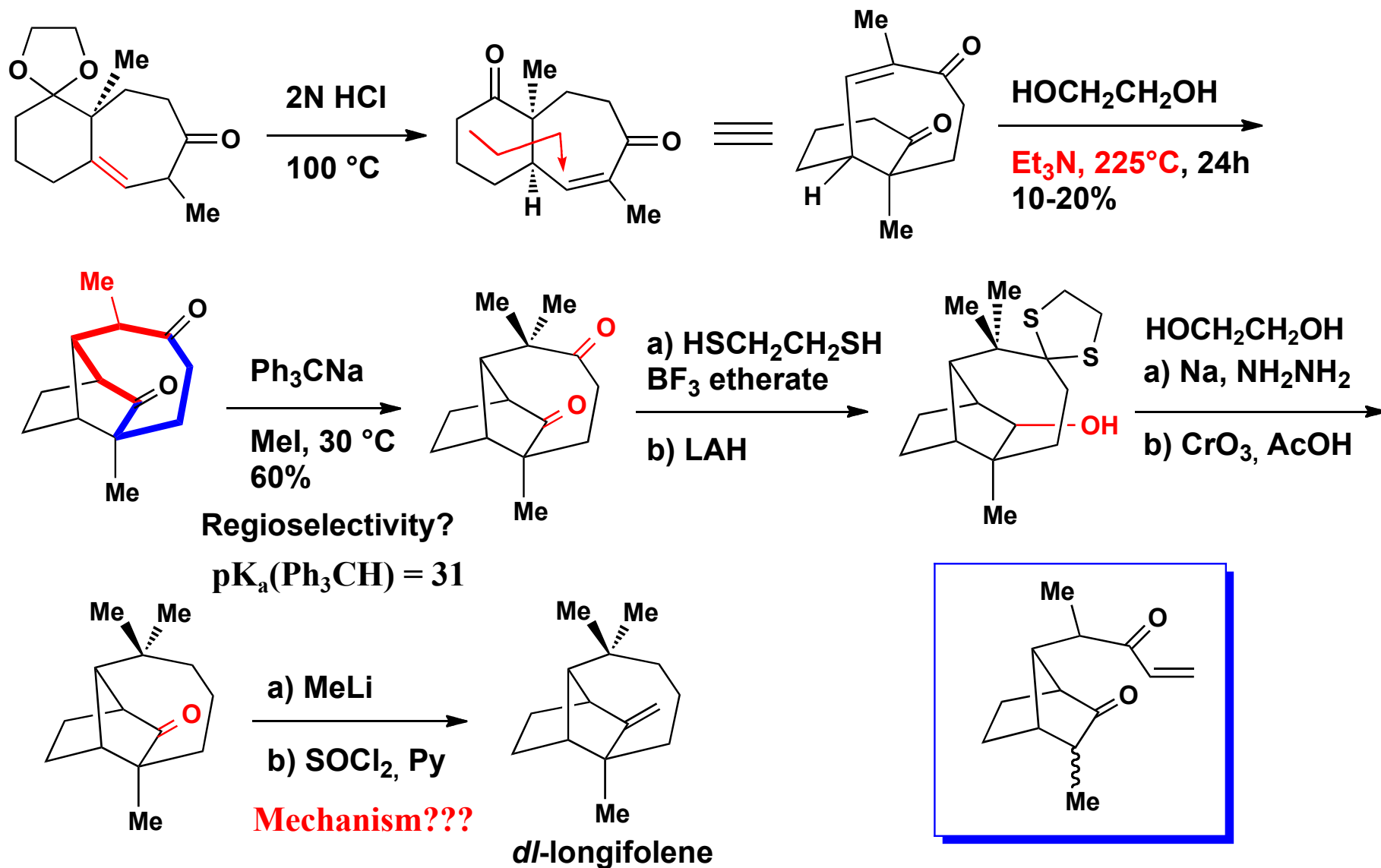


\* Regio-selective dihydroxylation of diene

\* Pinacol rearrangement:

- migration aptitude: vinyl vs  $\text{Csp}_3$  carbon
- solvolysis of tosylsulfonate vs tertiary allylic alcohol

# Longifolene—Corey 1961



Corey, E. J. et al *J. Am. Chem. Soc.* **1961**, 83, 1251-1253; **1964**, 86, 478-485.

# Summary of the Synthesis: Reactions and Tactics

**Robinson Annulation (Michael/aldol cascade)**

**Wittig reaction**

**Michael addition: For the synthesis of 1,5-dicarbonyls**

**Dihydroxylation of Alkene with OsO<sub>4</sub>**

**Pinacol Rearrangement: Chemo- and Regio-selectivities**

**Enolate Alkylation: Regioselectivity**

**Protective group: Heavily used**

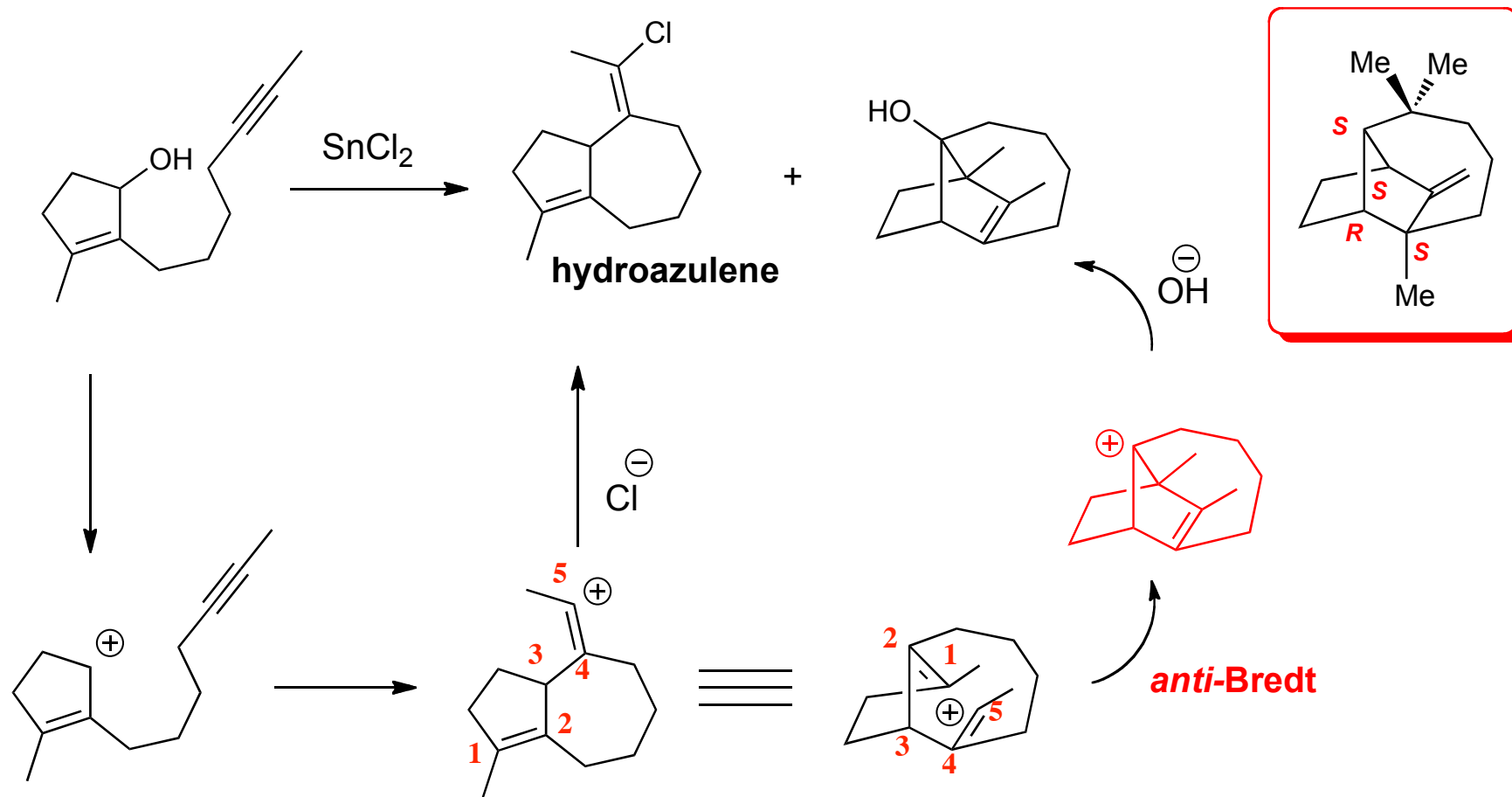
**Protection of Carbonyl: Acetal, thioacetal**

**How the 7-membered ring was synthesized?**

**Ring expansion via pinacol rearrangement (Strategic)**

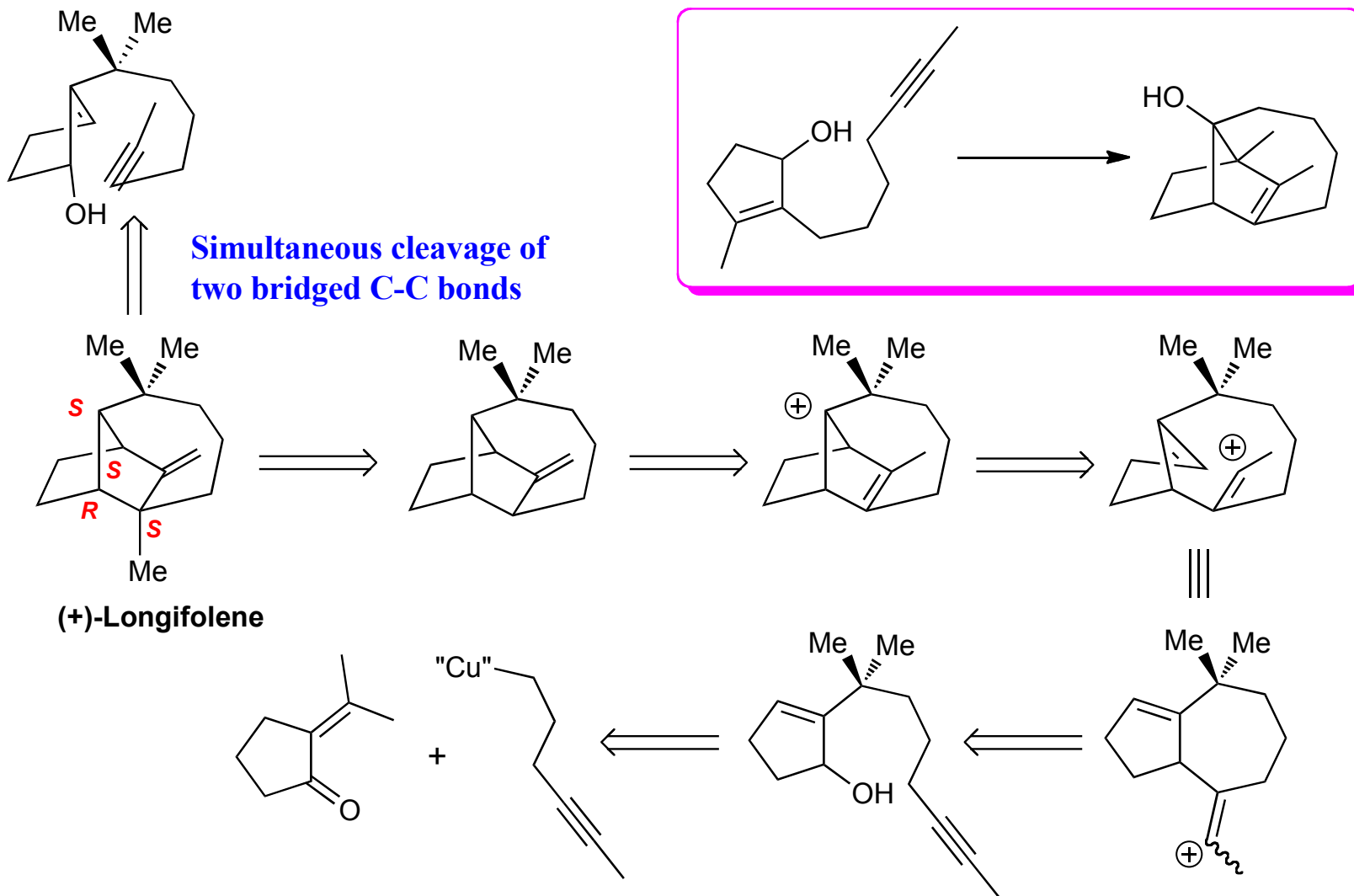
# Longifolene (Johnson 1975): Unexpected Results

From hydroazulene to longifolene



Johnson, W. S. et al *J. Am. Chem. Soc.* **1975**, *97*, 4777-4779.

# Longifolene—Johnson 1975



Johnson, W. S. et al *J. Am. Chem. Soc.* **1975**, *97*, 4777-4779

## Reactions Needed to Know

**Organocuprate: 1,4-addition to enone**

**Cationic cyclizations: Alkenes and alkynes are  $\pi$ -nucleophiles**

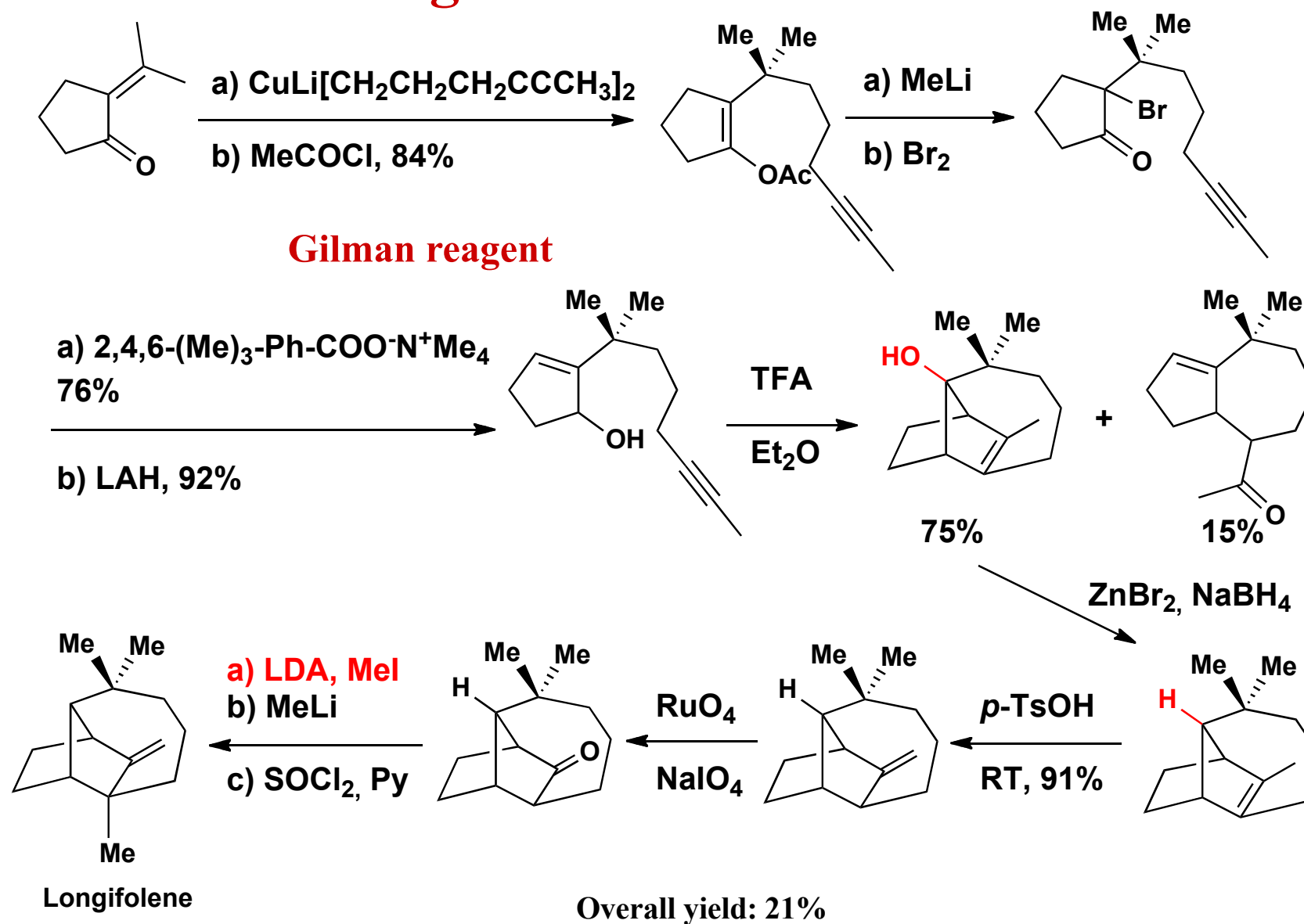
**Oxidation of ketone to enone: Control of regioselectivity**

**Dihydroxylation of alkenes with  $\text{RuO}_4/\text{NaIO}_4$**

**Non-Classic carbocation**

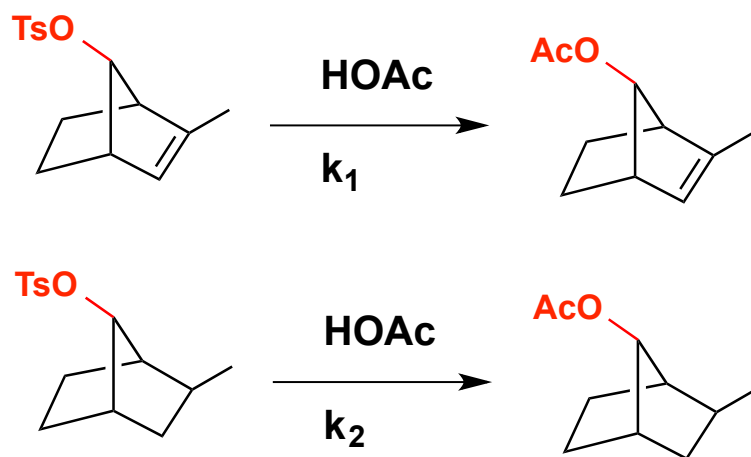
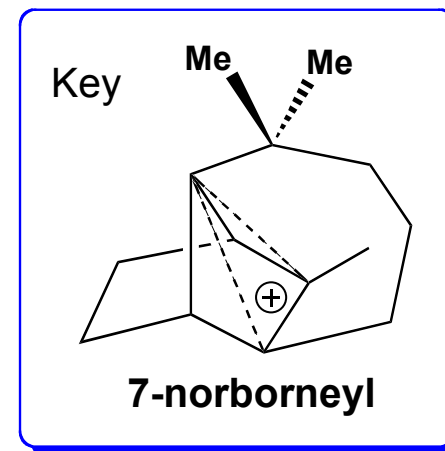
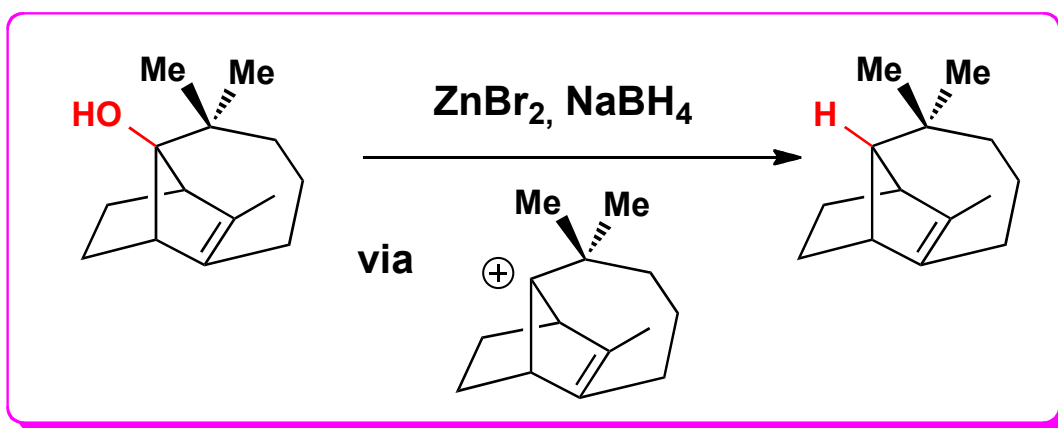
**Bredt Rule, Anti-Bredt**

# Longifolene—Johnson 1975



Johnson, W. S. et al *J. Am. Chem. Soc.* **1975**, *97*, 4777-4779

# Non-Classical Carbocation



$$k_1/k_2 > 10^{11}$$

*anti* 7-TsO-Norbornene solvolysis occurs with complete retention of stereochemistry

Weinstein, S. *J. Am. Chem. Soc.* **1956**, 78, 592; **1963**, 85, 2324-2326.  
Brown, H. C. *J. Am. Chem. Soc.* **1963**, 85, 2324.

Computational studies on Johnson's synthesis: Tantillo, D. J. *J. Org. Chem.* **2005**, 70, 5139-43.

# Summary of the Synthesis: Reactions and Tactics

**Organocuprate: 1,4-addition to enone**

**Cationic cyclizations: Alkenes and alkynes are  $\pi$ -nucleophiles**

**Oxidation of ketone to enone: Control of regioselectivity**

**Dihydroxylation of alkenes with  $\text{RuO}_4/\text{NaIO}_4$**

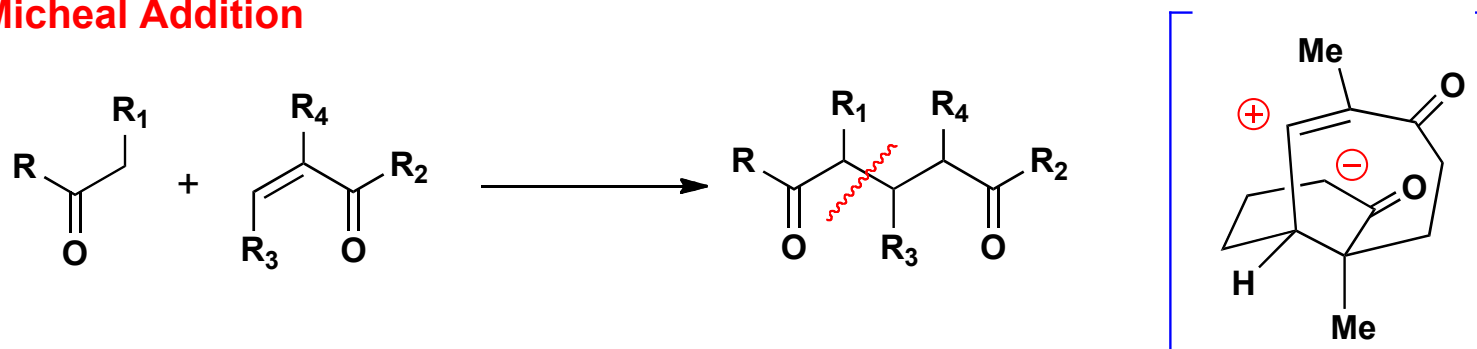
**Non-Classic carbocation**

## How to cleave a C=C double bond?

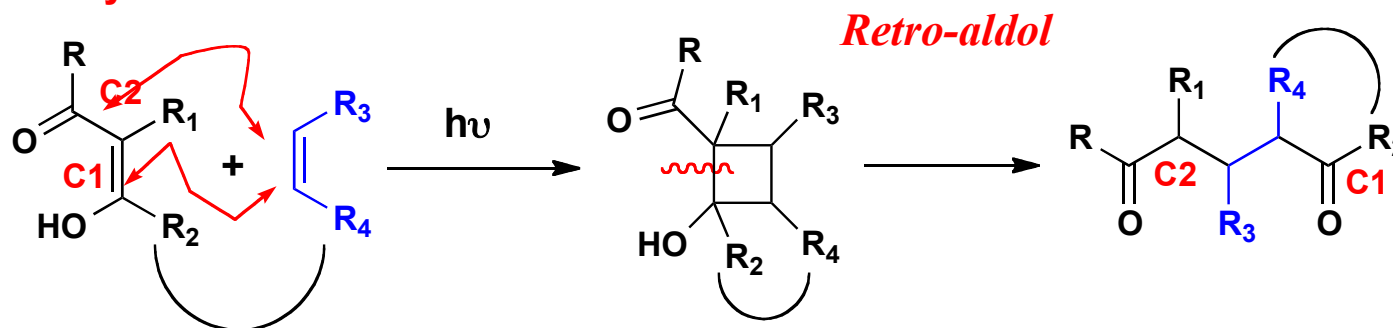
- a) Ozonolysis
- b) Dihydroxylation followed by oxidative cleavage of the resulting 1,2-diol

# Synthesis of 1,5-Dicarbonyl Compounds

## Micheal Addition



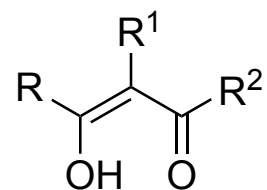
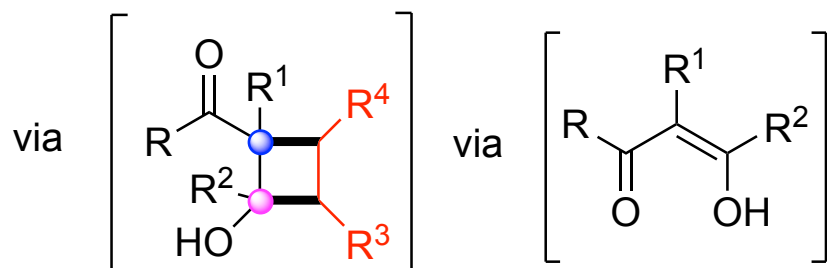
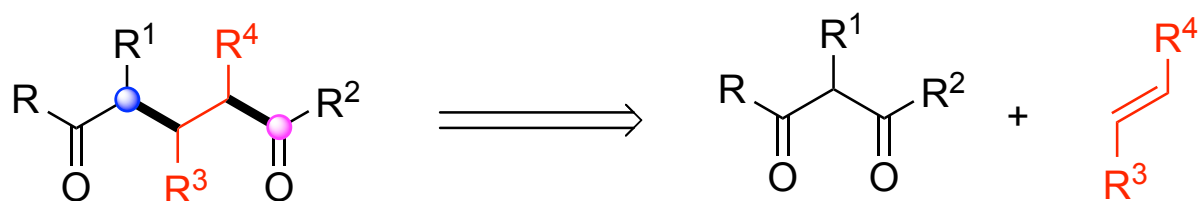
## De Mayo reaction



Olefin inserted into the C1-C2 bond of the enol ether

P. de Mayo, *Acc. Chem. Res.*, **1971**, *4*, 41;  
W. Oppolzer, *Acc. Chem. Res.*, **1982**, *15*, 135

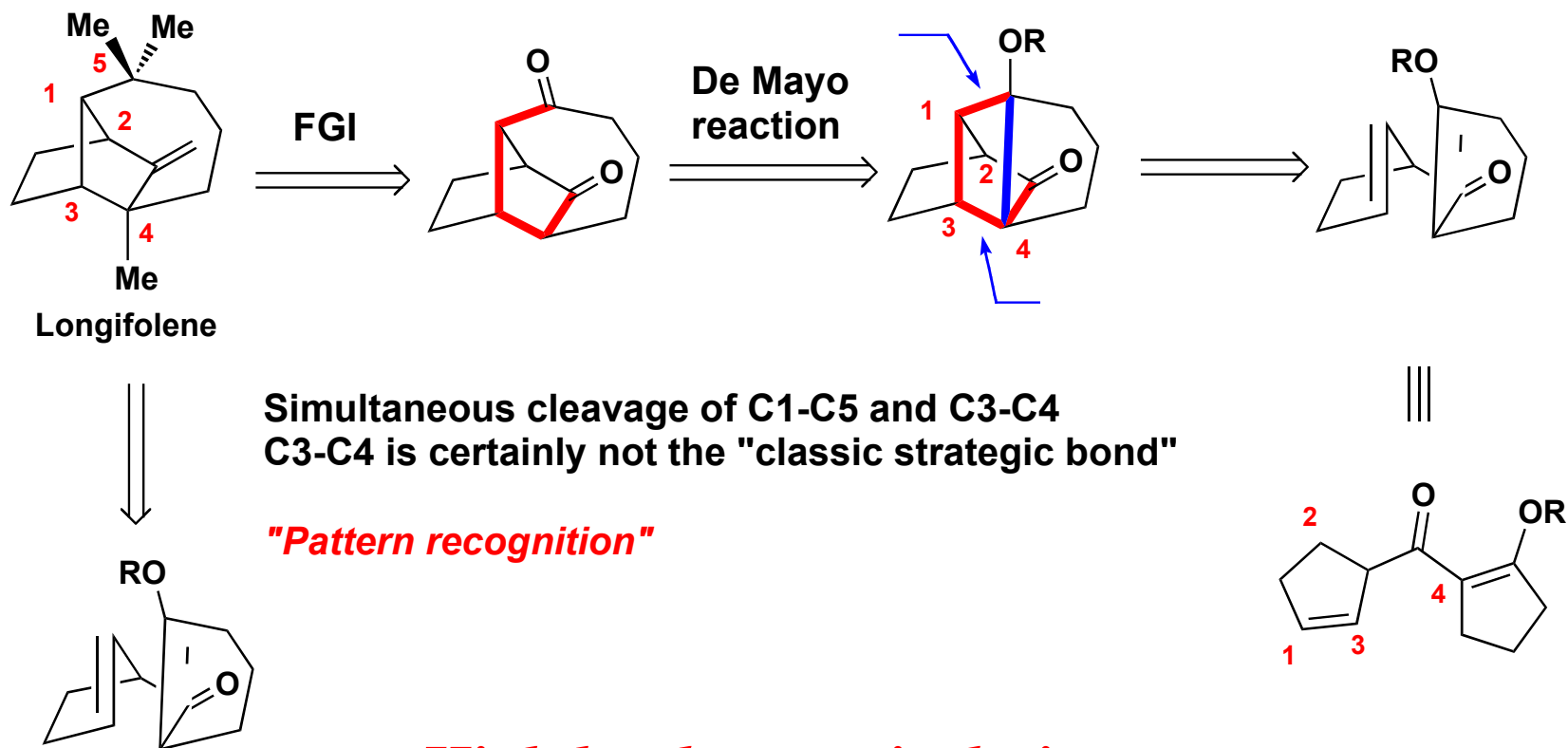
# Synthesis of 1,5-dicarbonyls by De Mayo Reaction



**Will give another regioisomer**

P. de Mayo, *Acc. Chem. Res.*, **1971**, *4*, 41;  
W. Oppolzer, *Acc. Chem. Res.*, **1982**, *15*, 135

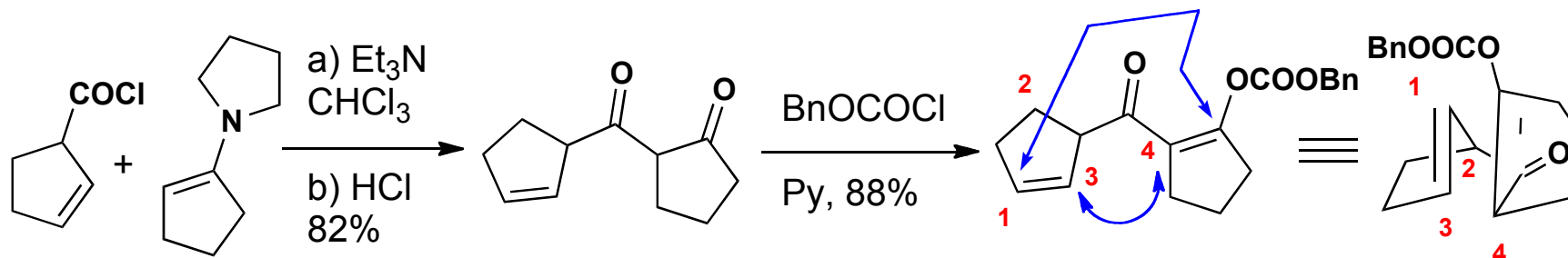
# Longifolene—Oppolzer 1978



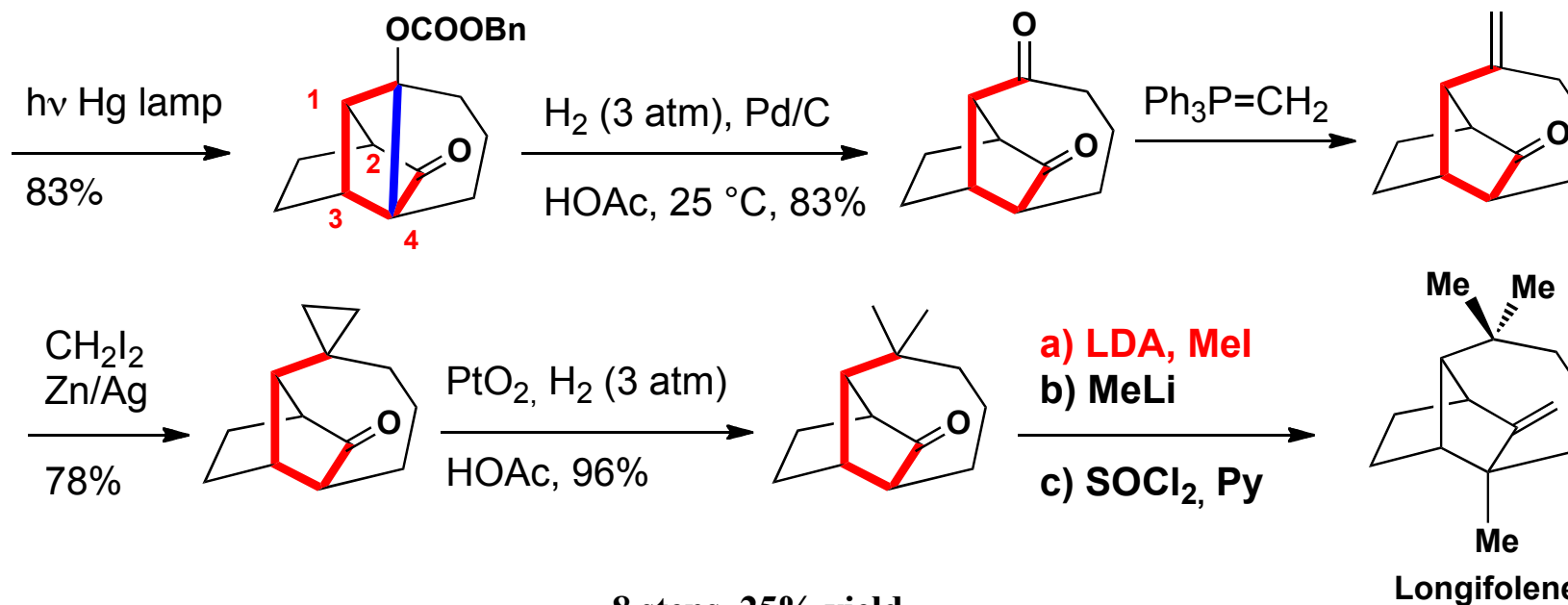
*High level strategic design*

“Pattern recognition”, termed by Danishefsky, seeks to discover an exploitable substructural motif around which to organize the thought process and then the synthesis. The most fascinating cases arise when the target itself must actually be modified leading to a more complex compound before a staple pattern is revealed. M. Wilson, S. J. Danishefsky, *J. Org. Chem.* **2007**, 72, 4293-4305.

# Longifolene—Oppolzer 1978



*Acylation of enolate, potential problem?*



8 steps, 25% yield.

Oppolzer, W.; Godel, T. *J. Am. Chem. Soc.* **1978**, *100*, 2583-2584.

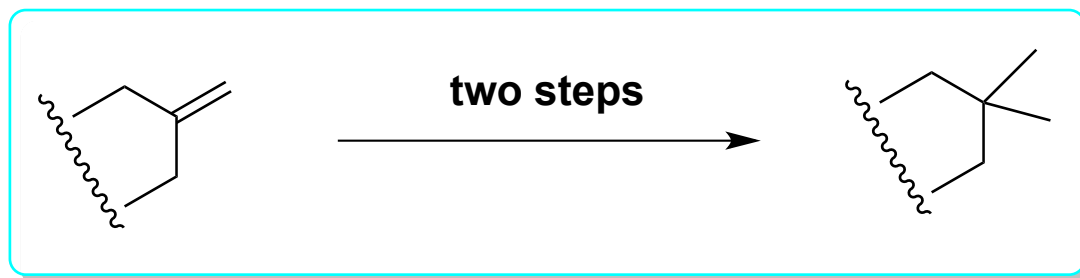
# Summary of the Synthesis: Reactions and Tactics

Stork enamine acylation: Synthesis of **1,3-dicarbonyls**

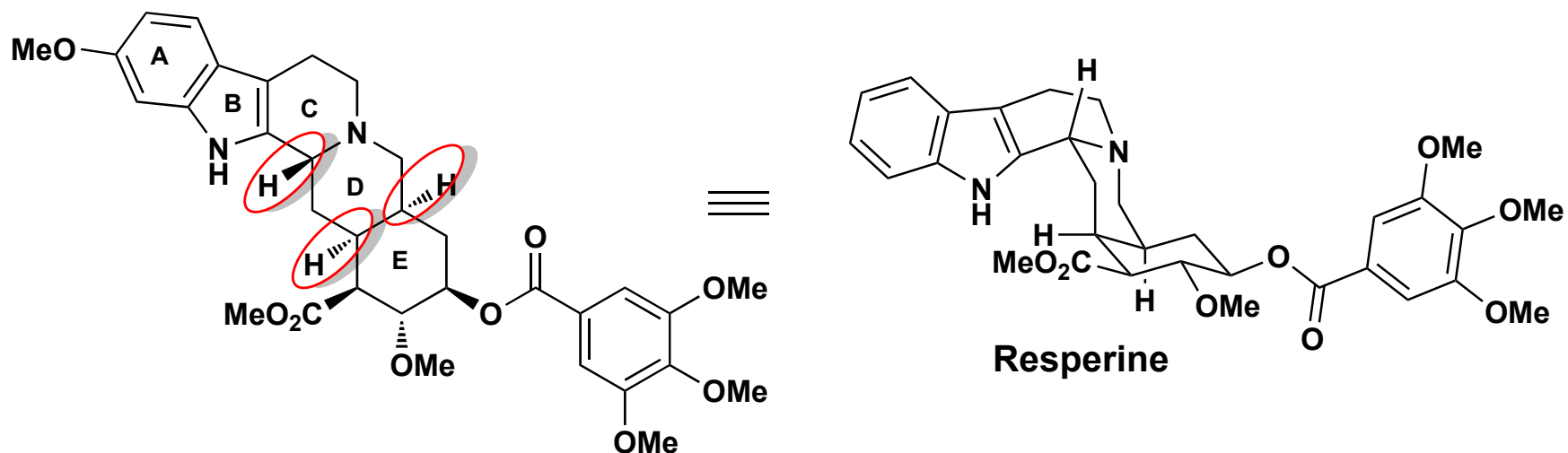
De Mayo reaction: [2+2] cycloaddition/retro-aldol for the Synthesis of **1,5-dicarbonyls**

Simmons-Smith cyclopropanation

**How to realize the following transformation**



# Reserpine



**Isolation: 1952,**

**Structural determination: Plane structure 1953**

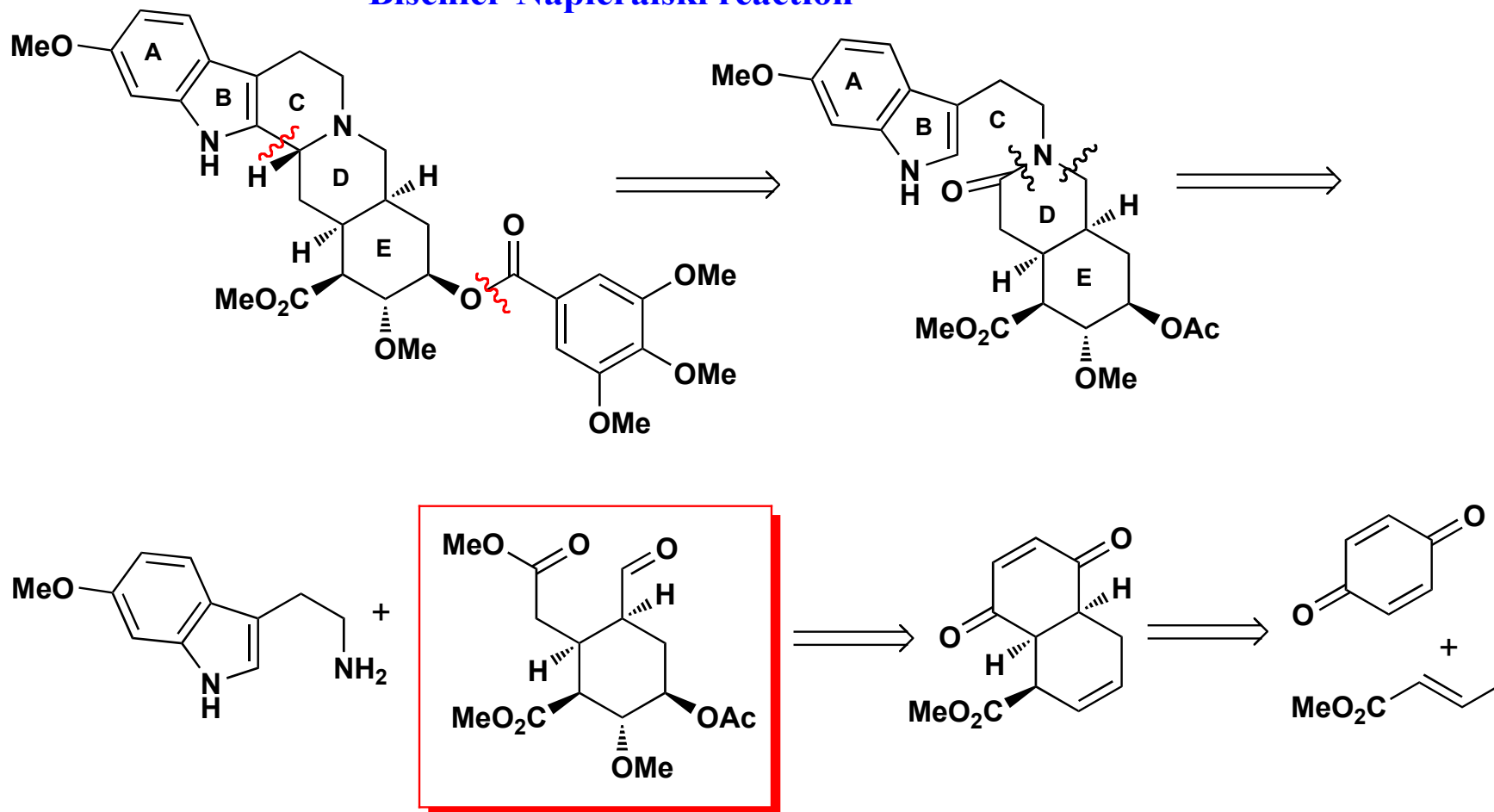
**Stereochemistry: 1955**

**Bioactivities: Used for the treatment of hypertensive, nervous and mental disorder, discontinued due to side-effects including Depression.**

**First total synthesis by Woodward 1956**

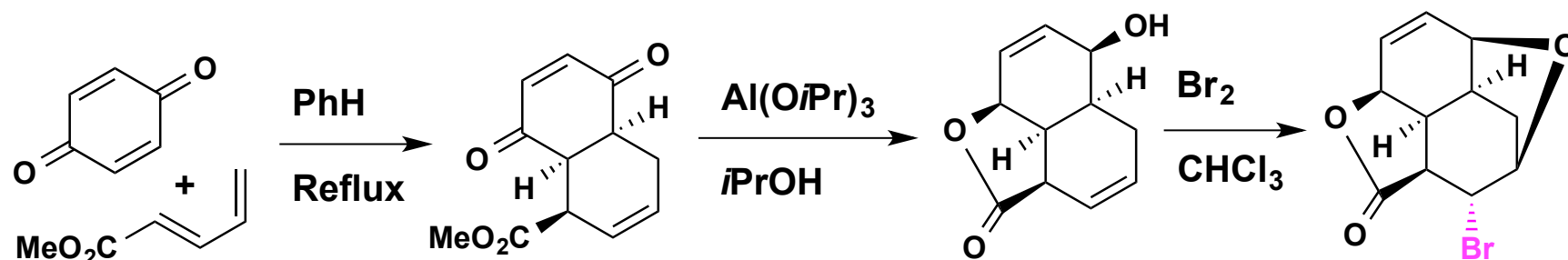
# Reserpine-Woodward

## Bischler-Napieralski reaction

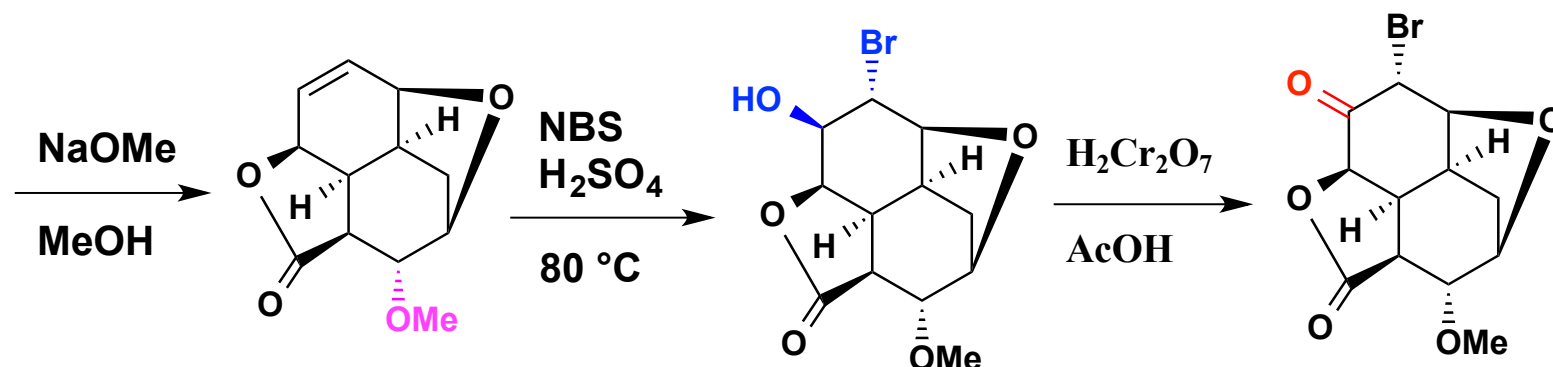


R. B. Woodward, F. E. Bader, H. Bickel, A. J. Frey, R. W. Kierstead, *J. Am. Chem. Soc.* **1956**, 78, 2023-2025.

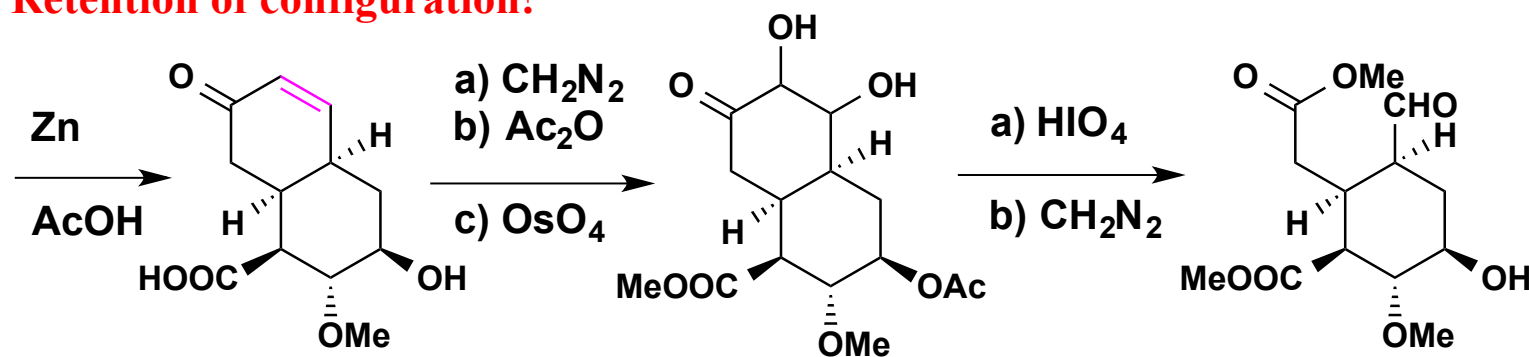
# Reserpine-Woodward



## Meerwein-Ponndorf-Verley



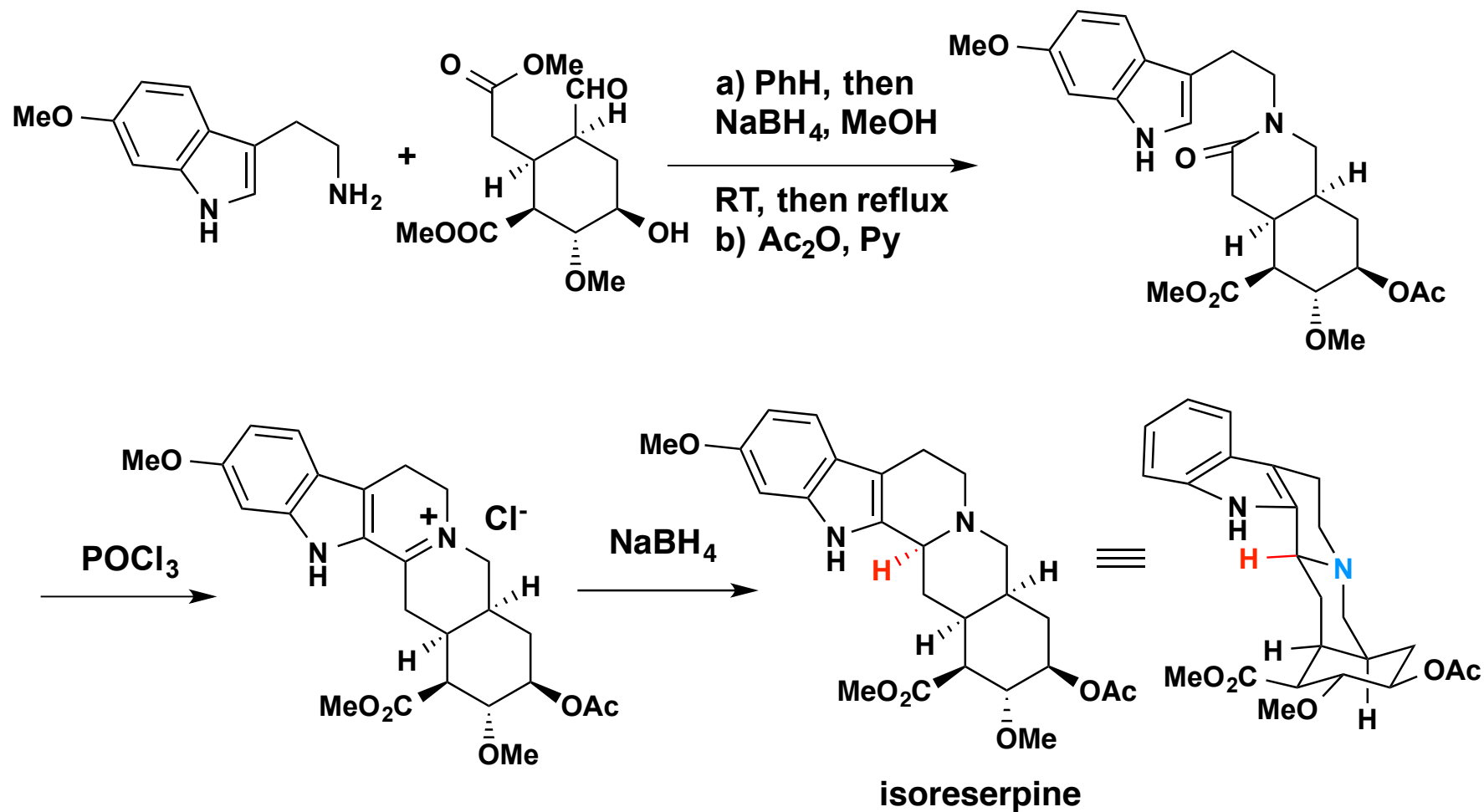
## Retention of configuration!



## Control of relative stereochemistry: "A Cyclic trick"

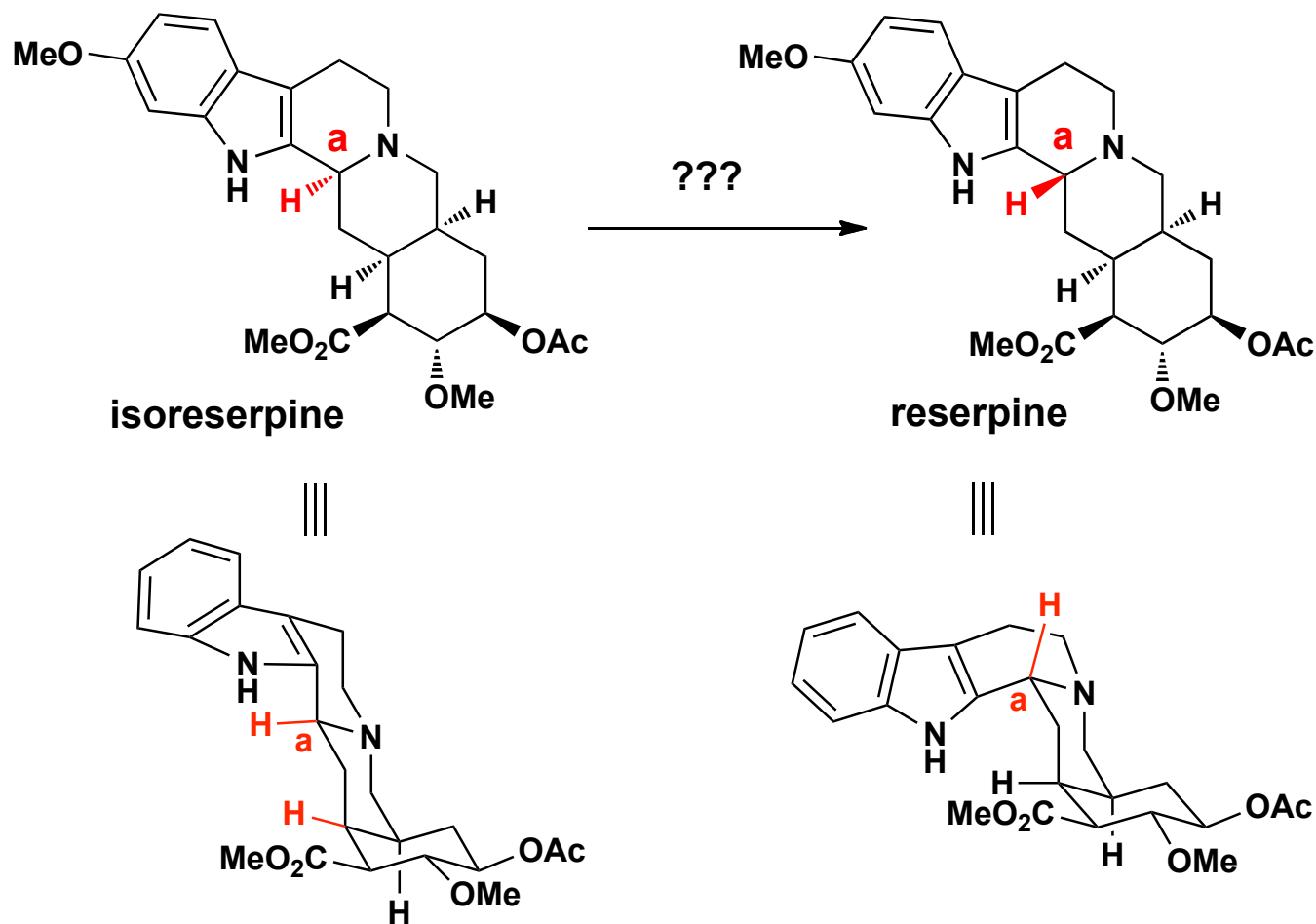
R. B. Woodward, *J. Am. Chem. Soc.* **1956**, *78*, 2023-2025 and *Tetrahedron* **1958**, *2*, 1-57.

# Reserpine-Woodward



*Hydride attacks on the convex face affording the epimer of natural product*

# Reserpine-Woodward

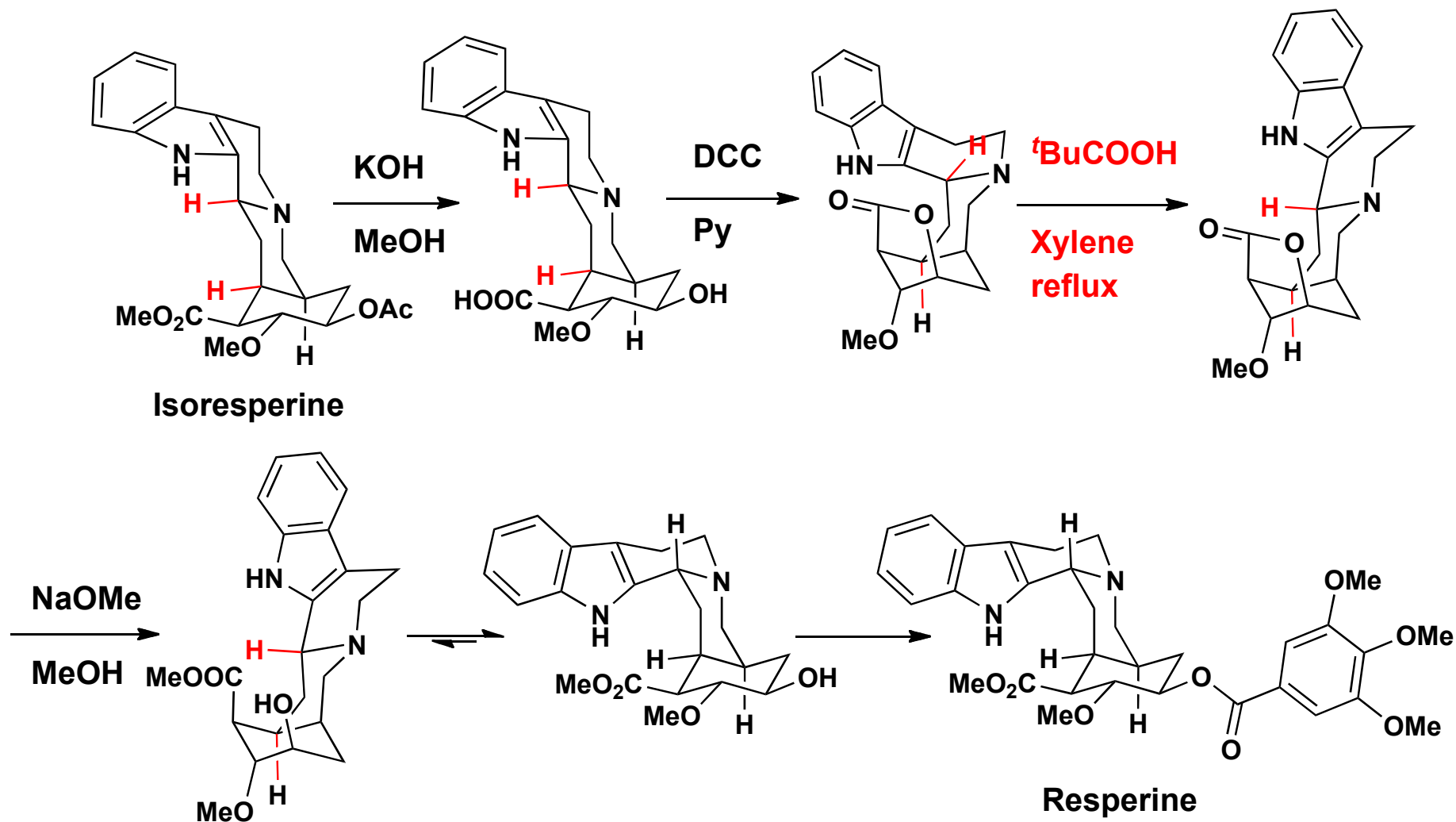


**Isomerization:** Mechanistically possible However,  $C_a$  of iso-reserpine is configurationally quite stable and is kinetically and thermodynamically favored

**Solution:** Change the conformation of iso-reserpine!!!

**Conformational analysis:** Barton, D. H. R. *Experientia* 1950, 316-320.

# Reserpine-Woodward

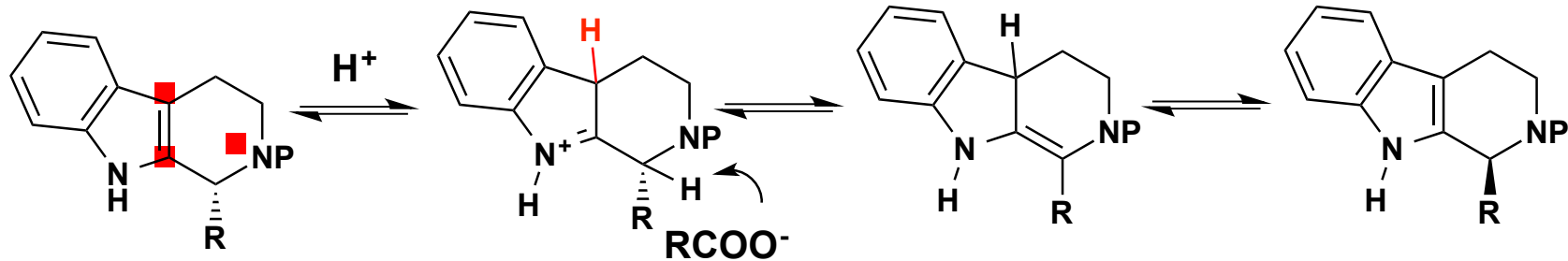


R. B. Woodward, F. E. Bader, H. Bickel, A. J. Frey, R. W. Kierstead, *J. Am. Chem. Soc.* **1956**, 78, 2023-2025.  
*Tetrahedron* **1958**, 2, 1-57.

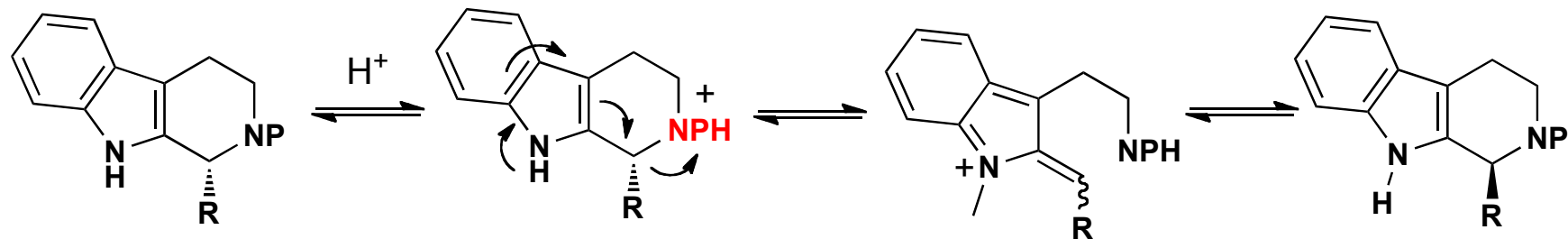
# Reserpine-Woodward

## Possible Mechanism of Epimerization of Carbazole

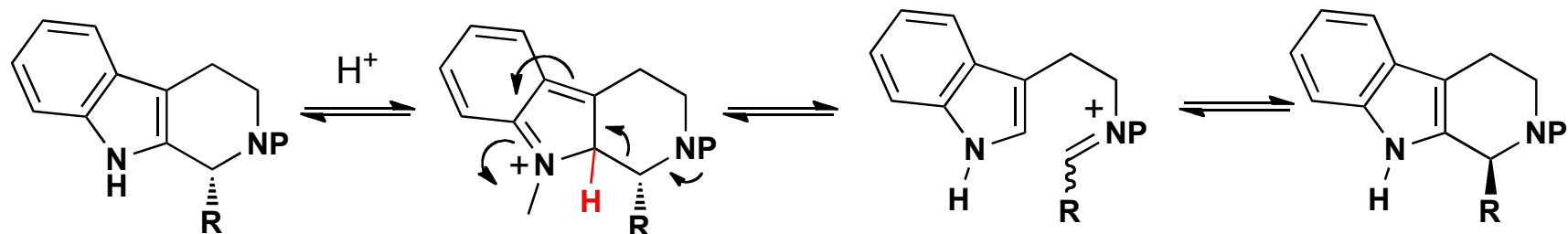
### Mechanism 1



### Mechanism 2

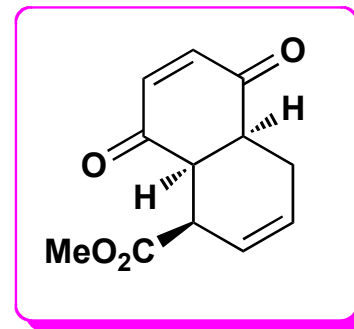
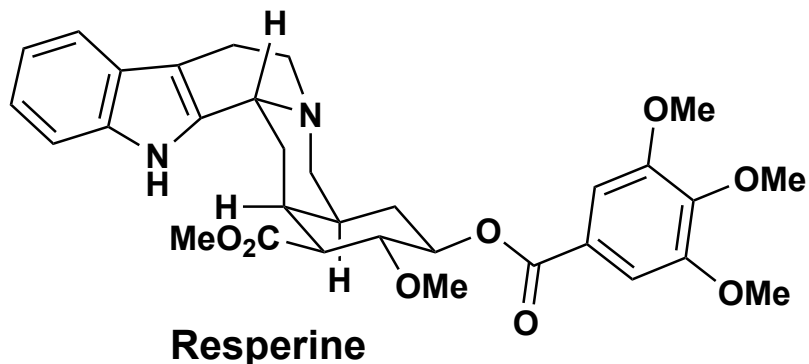


### Mechanism 3



*Which mechanism is the most probable one*

# Reserpine-Woodward



Considered as one of Woodward's greatest contributions to synthesis

- Using Diels-Alder to generate *cis*-fused bicyclic ring system.
- Conceptualized the use of *cis*-fused bicyclic ring system as stereocontrol element.
- Pioneering conformational analysis in total synthesis

R. B. Woodward, F. E. Bader, H. Bickel, A. J. Frey, R. W. Kierstead, *J. Am. Chem. Soc.* **1956**, 78, 2023-2025.  
*Tetrahedron* **1958**, 2, 1-57.

For a recent total synthesis, see: R. Sarpong, *Nat. Chem.* **2013**, 5, 126-131.

*Home work: Read this paper and understand the design principle*

# Summary of the Synthesis: Reactions and Tactics

**Diels-Alder cycloaddition**

**Meerwein-Ponndorf-Verley (MPV) reaction**

**Bromoetherification**

**Beta-Elimination**

**Dihydroxylation of alkene/cleavage of diol**

**Bischler-Napiralski reaction**

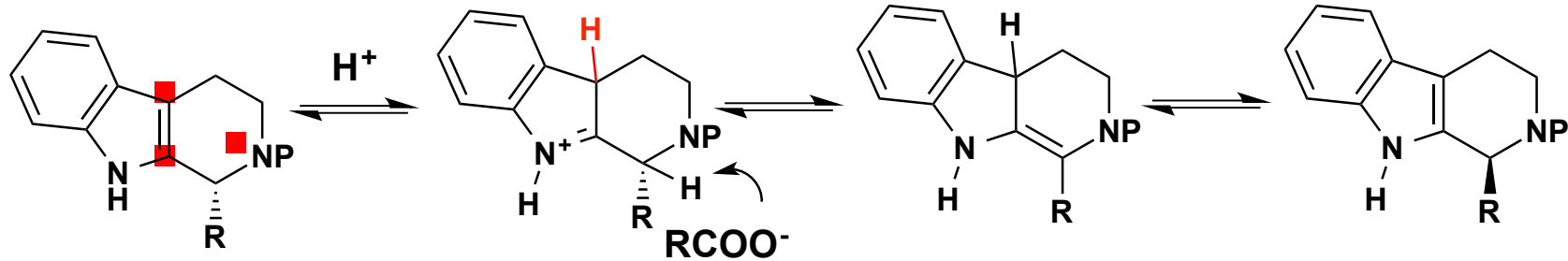
**Cyclic trick to control the relative stereochemistry**

**Conformation analysis solve the stereochemical problem**

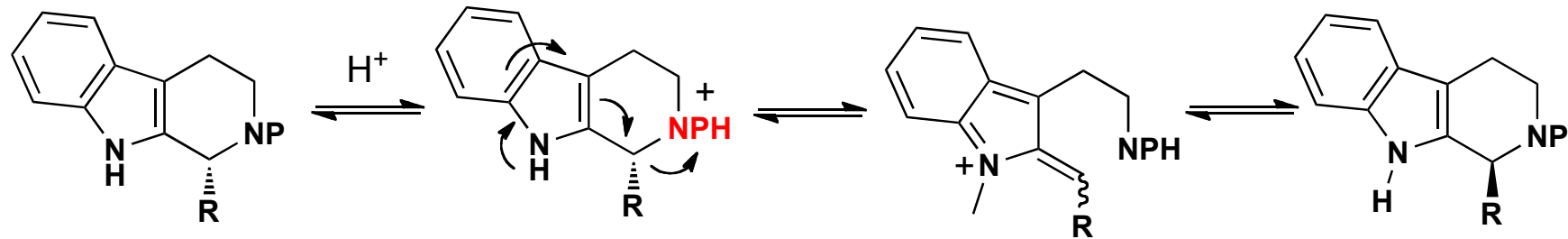
# Reserpine-Woodward

## Possible Mechanism of Epimerization of Carbazole

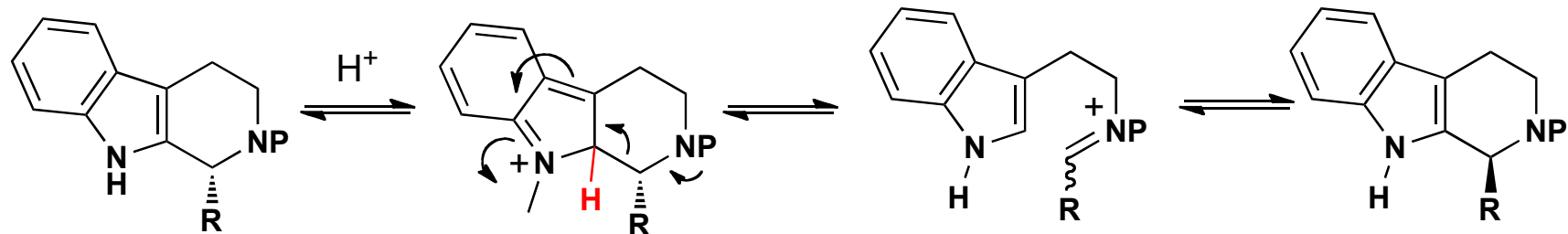
### Mechanism 1



### Mechanism 2

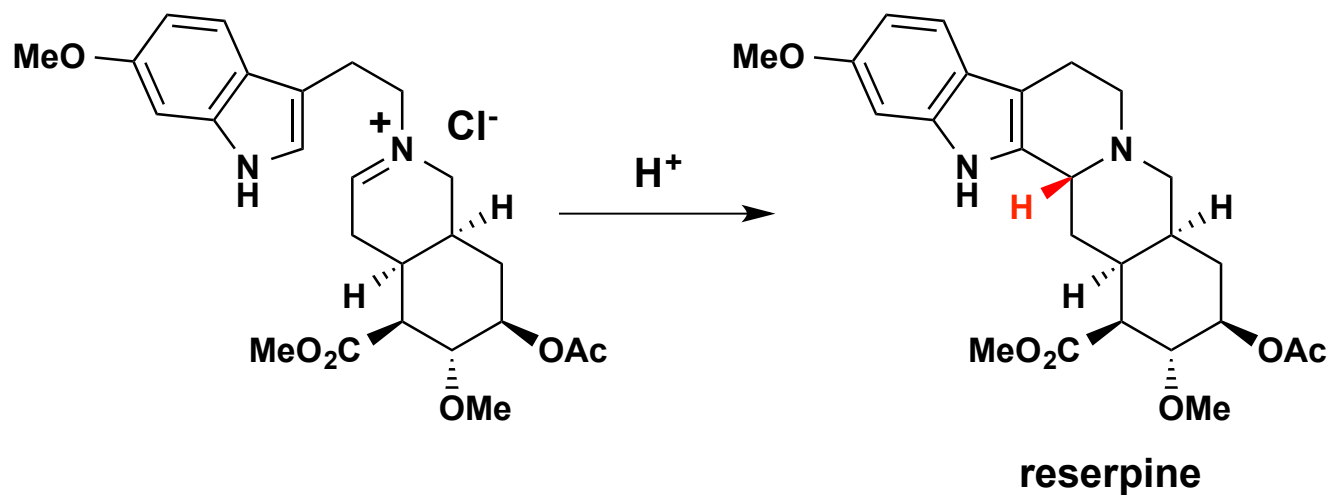
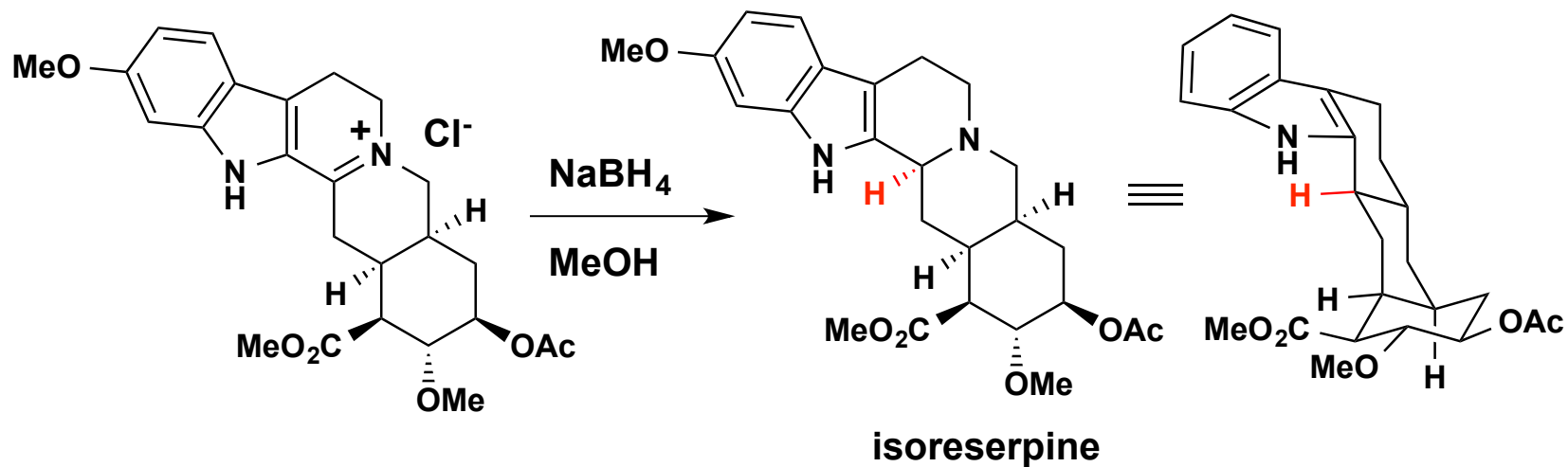


### Mechanism 3



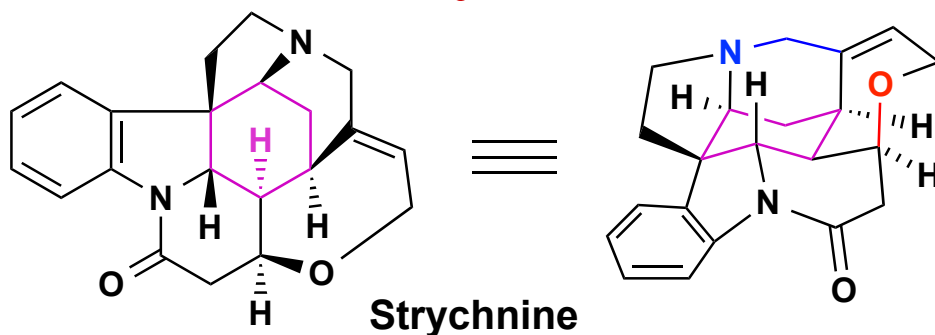
*Which mechanism is the most probable one*

# Reserpine Synthesis through Pictet-Spengler Reaction



For a recent total synthesis, see: R. Sarpong, *Nat. Chem.* **2013**, *5*, 126-131.

# Strychnine



**1818:** Isolated by Pelletier & Caventou, *Ann. Chim. Phys.* **1818**, 8, 323.

**1838:** Molecular formule determined by Regnault, *Ann.* **1838**, 26, 35.

**1946:** Structure determined by Robinson, *Nature* **1946**, 157, 438.

and in **1947** by Woodward, *J. Am. Chem. Soc.* **1948**, 70, 2107-2115.

**1950:** Relative stereochemistry determined by X-ray analysis

Robertson & Beevers, *Nature* **1950**, 165, 690-691.

Bijvoet *et al.* *Acta Crystallogr.* **1951**, 4, 275-280.

**1956:** Absolute configuration defined by X-ray crystal by Peerdeman,

*Acta Crystallogr.* **1956**, 9, 824.

**1963:** Absolute configuration confirmed by chemical methods by Schmid

*Helv. Chim. Acta.* **1963**, 46, 1212-1231.

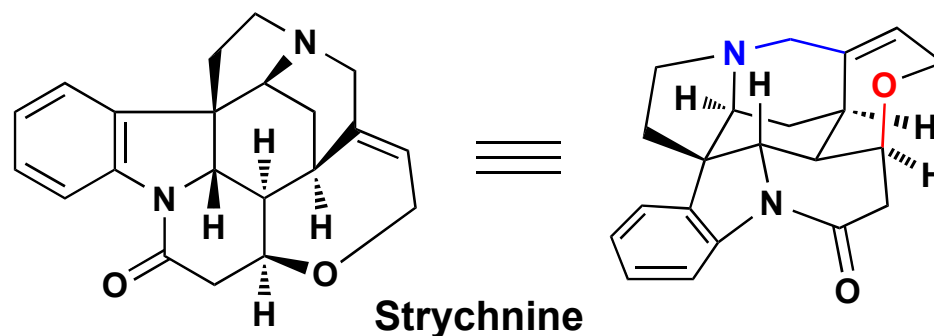
Strong poison ( $\approx 50$  mg is lethal for an adult human).

***Between the isolation and structure elucidation, Sir Robert Robinson's group alone published over 250 communications on the subject. Hermann Leuchs contributed another 125 papers.***

Review, see: Bonjoch, J.; D. Solé, *Chem. Rev.* **2000**, 100, 3455-3482.

Mori, M. *Heterocycles* **2010**, 81, 259-292.

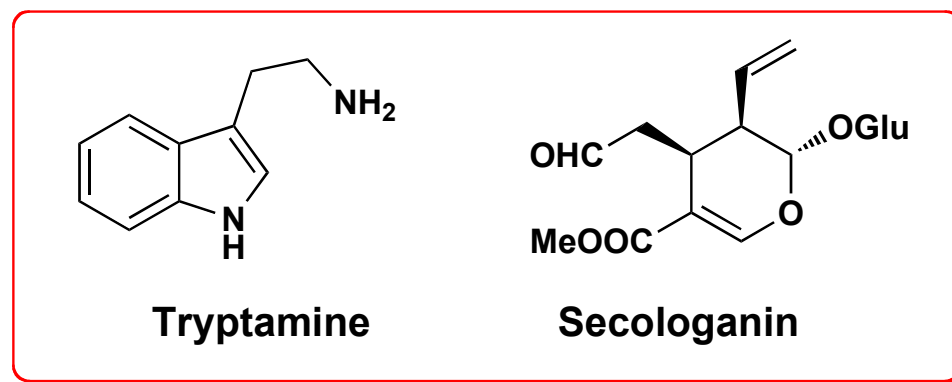
# Strychnine



Molecular formula:  $C_{21}H_{22}N_2O_2$

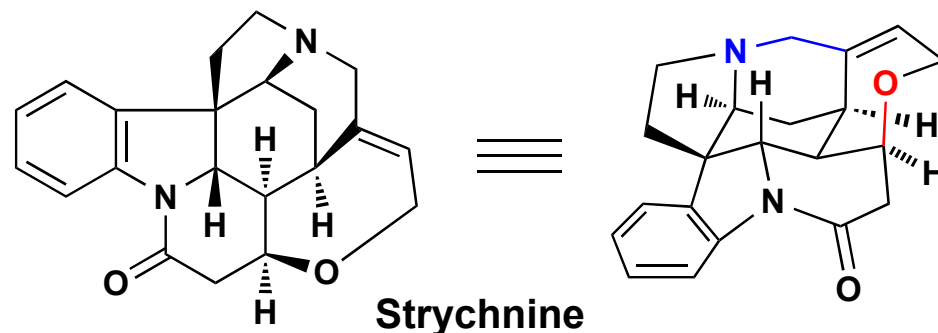
**24** skeletal atoms, **7** Rings, **6** Contiguous stereocentres, **5** in the cyclohexane ring.

A monoterpene indole alkaloid biogenetically derived from tryptamine and secologanin



Biosynthesis of monoterpene indole alkaloids: T. M. Kutchan, *Phytochem.* **1993**, 32, 493-506.  
S. E. O' Connor, J. J. Maresh, *Nat. Prod. Rep.* **2006**, 23, 532-547

# Total Synthesis of Strychnine



**R. B. Woodward** - Harvard University (1954)

**28 steps**

**Philip Magnus** - University of Texas (1992)

**Gilbert Stork** - Columbia University (1992)

**Larry E. Overman** - University of California, Irvine (1993)

**Martin E. Kuehne** - University of Vermont (1993)

**Viresh H. Rawal** - The Ohio State University (1994)

**Josep Bonjoch & Joan Bosch** - University of Barcelona (1999)

**Stephen F. Martin** - University of Texas (1996-2001)

**Michael J. Eichberg & K. Peter C. Vollhardt** - University of California, Berkeley (2000)

**Graham J. Bodwell** - Memorial University of Newfoundland (2002)

**Miwako Mori** - Hokkaido University (2002)

**Masakatsu Shibasaki** - University of Tokyo (2002)

**Tohru Fukuyama** - University of Tokyo (2004)

**Albert Padwa** - Emory University (2007)

**Rodrigo, B. Andrada**, Temple University (2010)

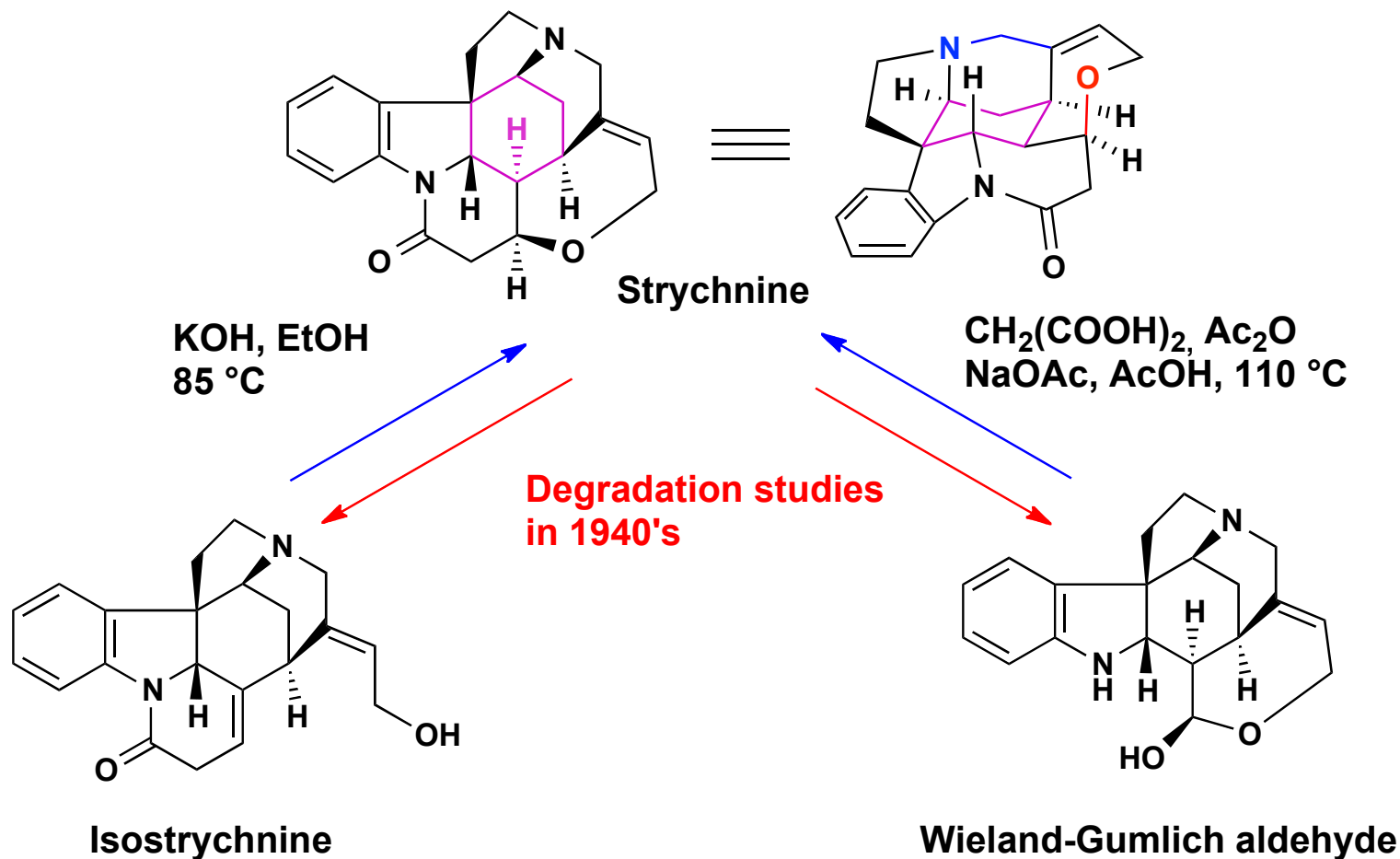
**Hans-Ulrich, Reissig**, Freie University Berlin (2010)

**Christopher, D. Vanderwal**, University of California, Irvine (2011)

**6 Steps**

**38 years**

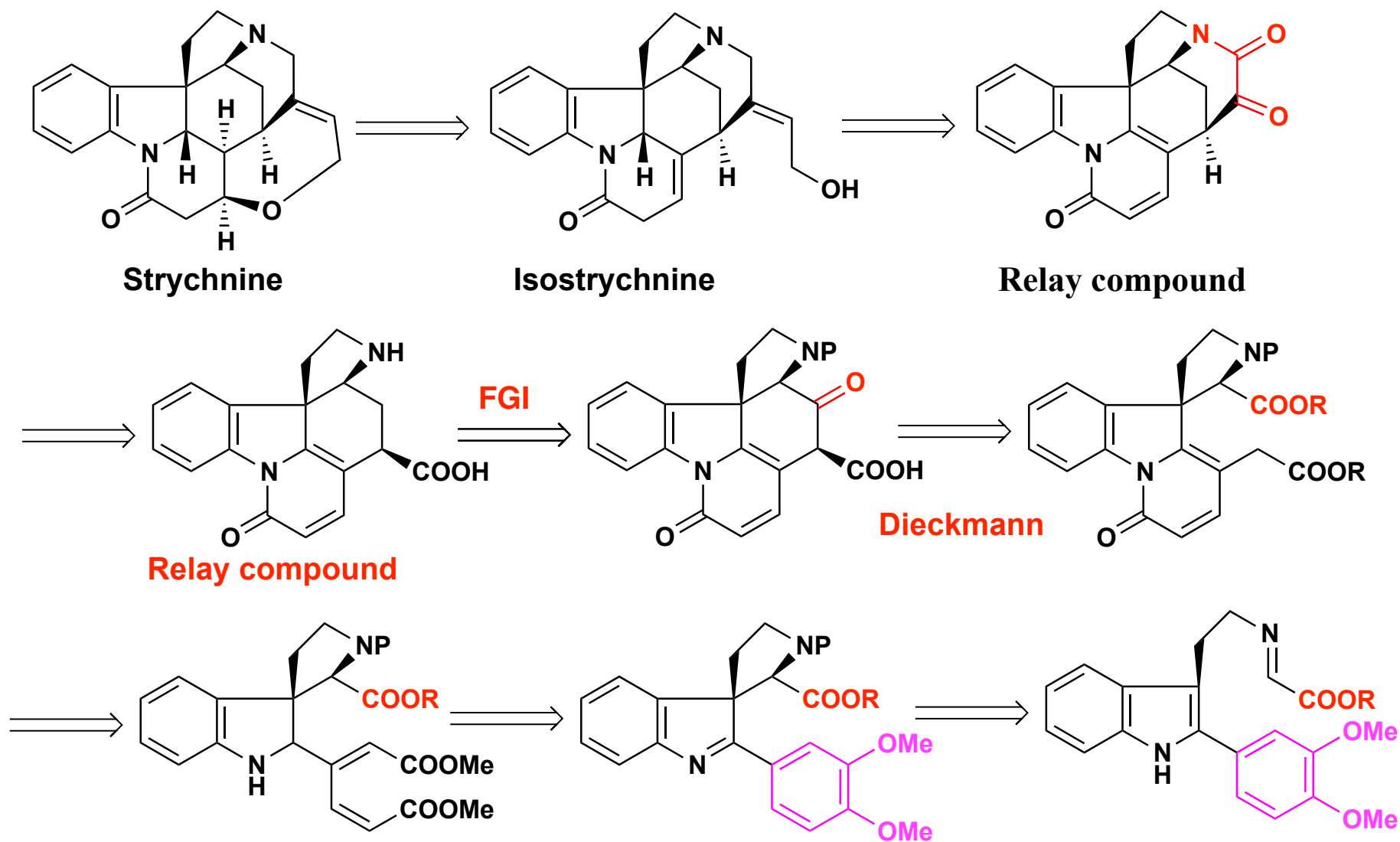
# Strychnine: Key background



In 1970's, both Isostrychnine and Wieland-Gumlich aldehyde were found in Nature.

Heinrich Wieland: Nobel Prize in 1927.

## Woodward's Total Synthesis of Strychnine: *A Relay Synthesis*



One of the key strategic considerations: **Relay Synthesis**

Woodward, R. B. *J. Am. Chem. Soc.* **1954**, *76*, 4749-4751; *Tetrahedron* **1963**, *19*, 247-288.

## **Reactions Needed to Know**

**Fischer indole synthesis**

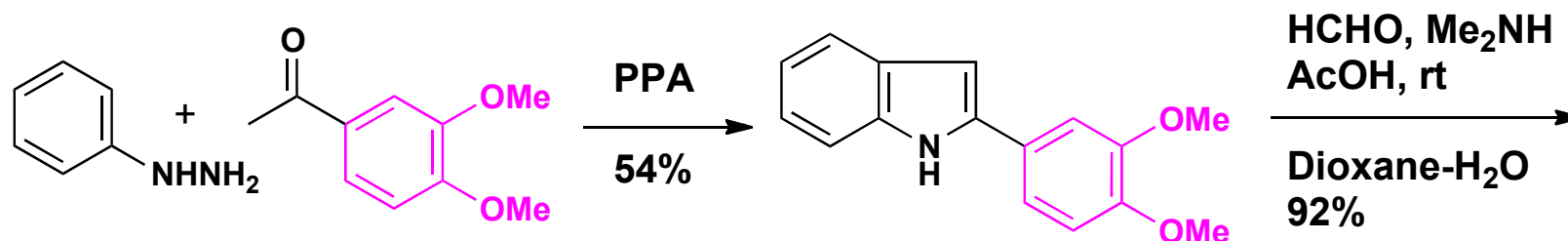
**Dieckman condensation**

**Pictet-Spengler reaction and Interrupted Pictet-Spengler reaction**

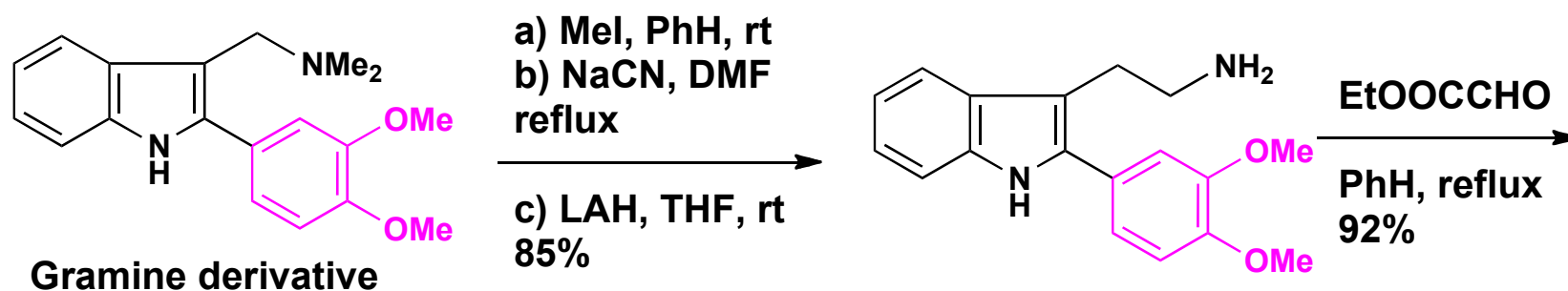
**Azalacton formation**

**Alpha-hydroxylation of ketone (SeO<sub>2</sub>, Ene reaction/2,3-σ rearrangement)**

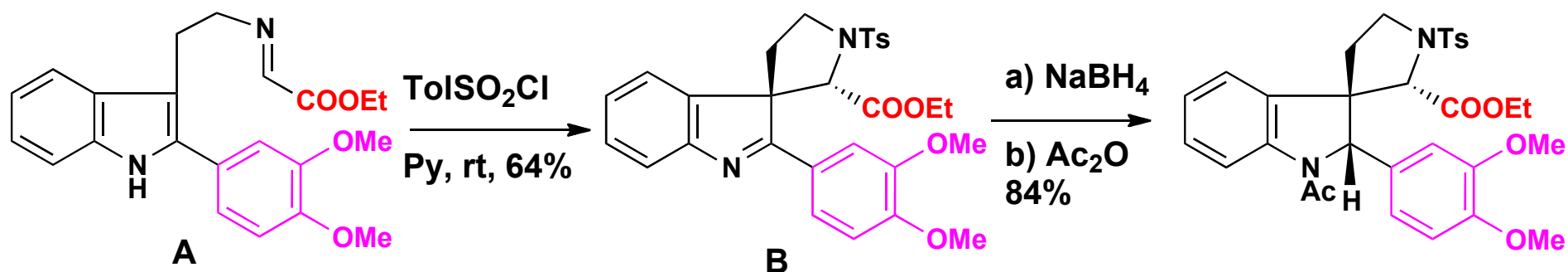
# Woodward's Total Synthesis of Strychnine



## Fischer Indole Synthesis

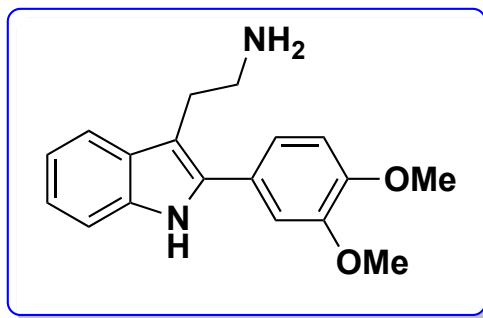


Gramine derivative

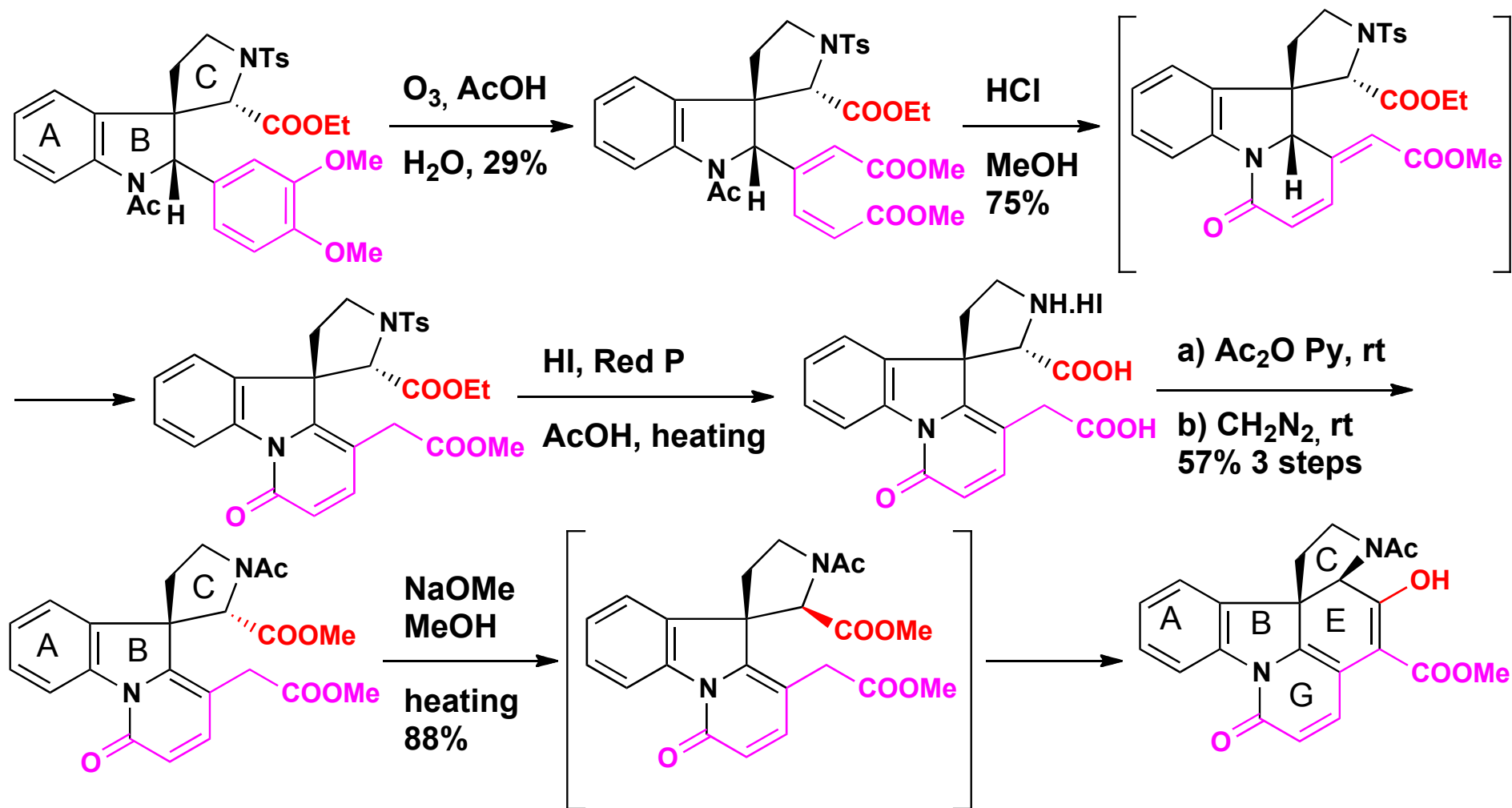


The C-2 position of Indole is blocked by the aromatic ring. Only spiroindolene was produced in the Pictet-Spengler reaction (*Interrupted Pictet-Spengler reaction*).

# Alternative Syntheses of 2,3-Disubstituted Indoles

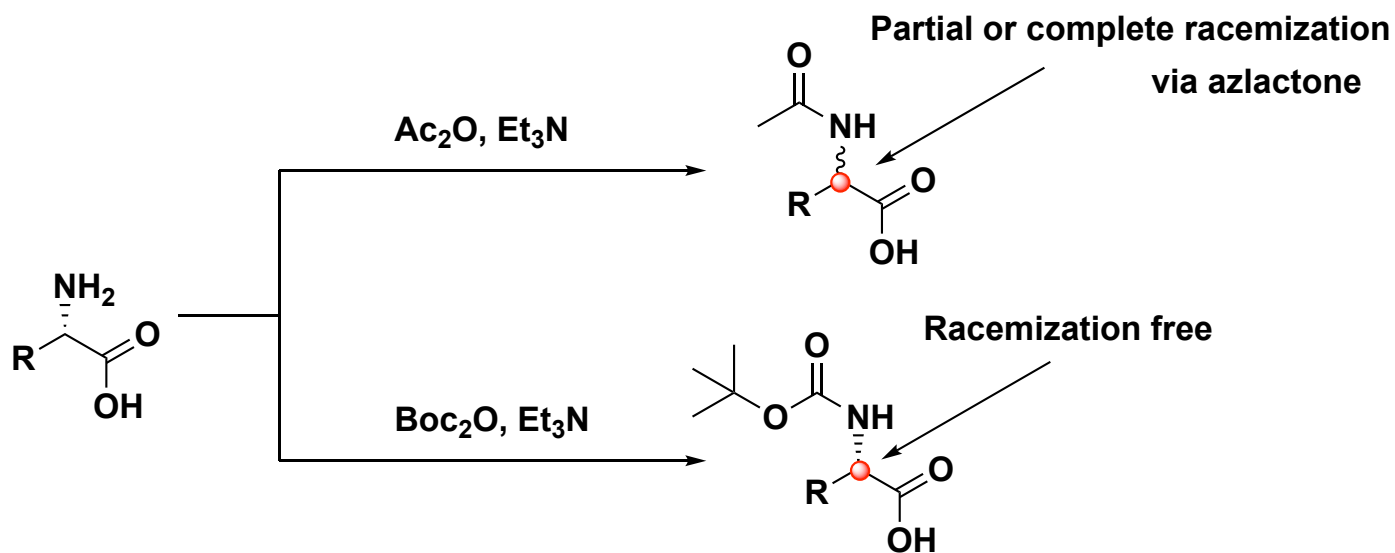


# Woodward's Total Synthesis of Strychnine

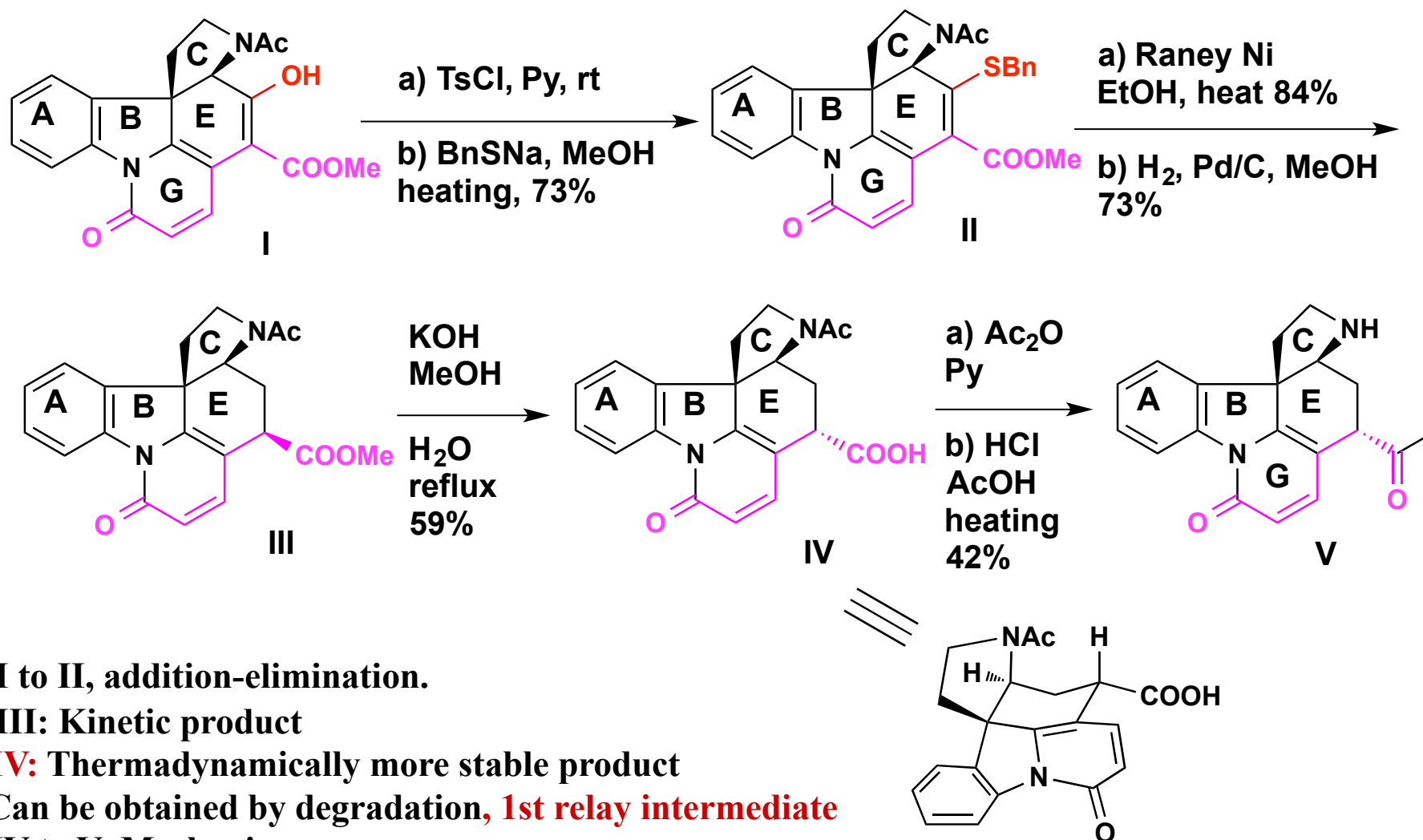


- \* Conversion of NTs to NAc at the C-ring is required to avoid the decomposition.
- \* Epimerization at the C- ring is required before Dieckman condensation (draw the conformer)
- \* The original aromatic ring became integral part of the G, E rings (beauty of the design, based on author's biosynthesis hypothesis, thought late it was proved to be wrong).

# Avoiding Acetylation of Free $\alpha$ -Aminoacid

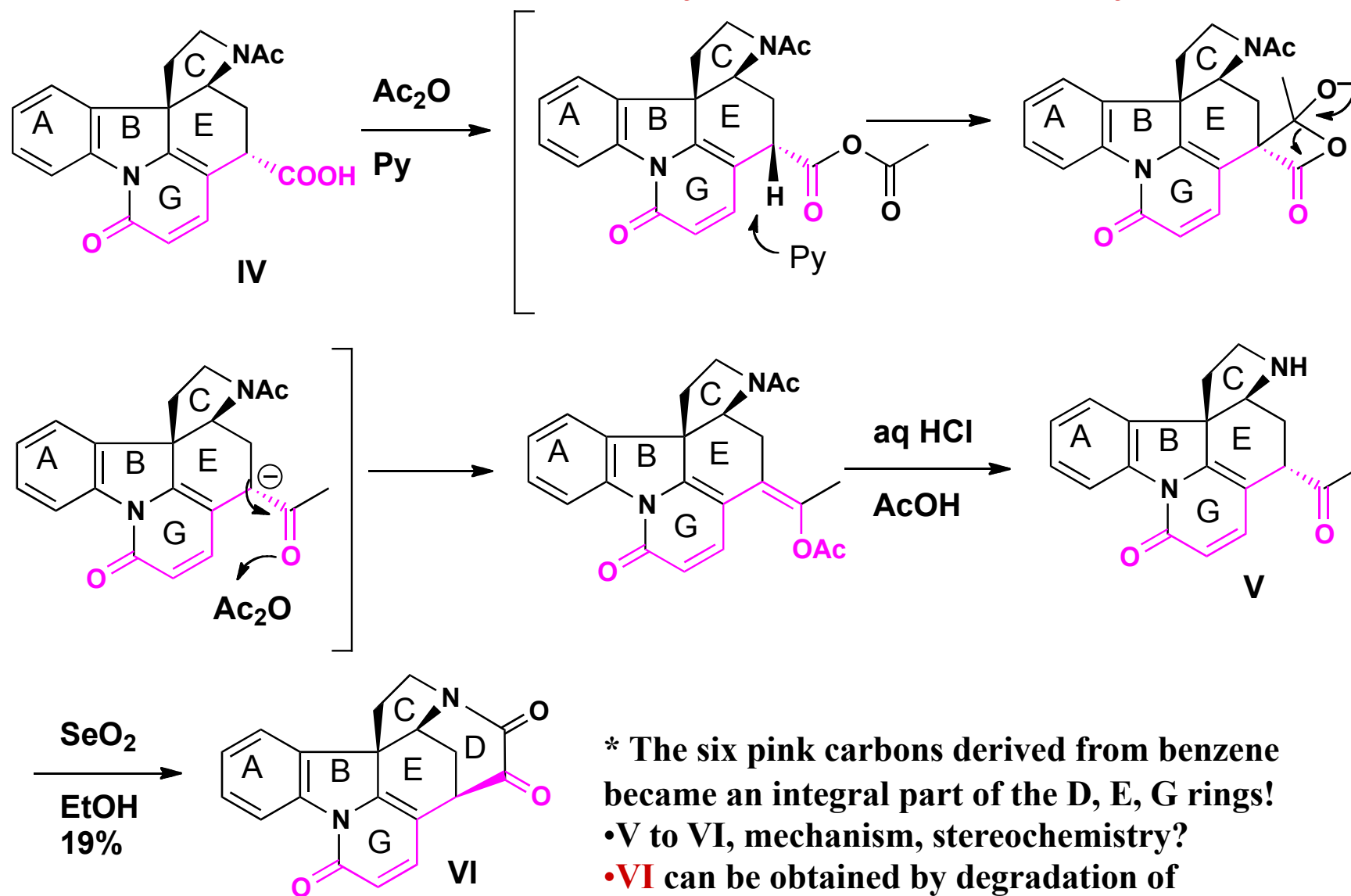


# Woodward's Total Synthesis of Strychnine



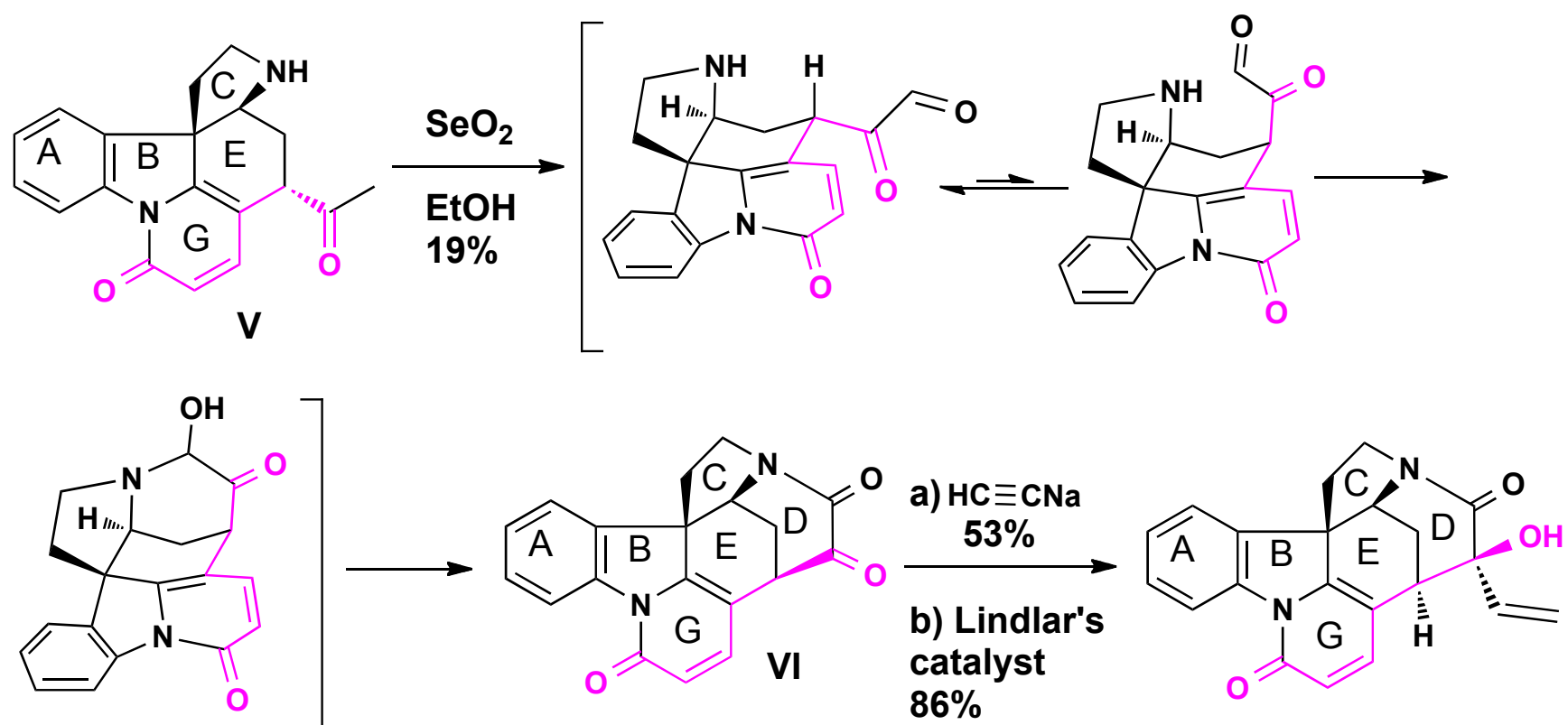
- \* I to II, addition-elimination.
- \* III: Kinetic product
- IV: Thermodynamically more stable product  
Can be obtained by degradation, **1st relay intermediate**
- IV to V: Mechanism

# Woodward's Total Synthesis of Strychnine

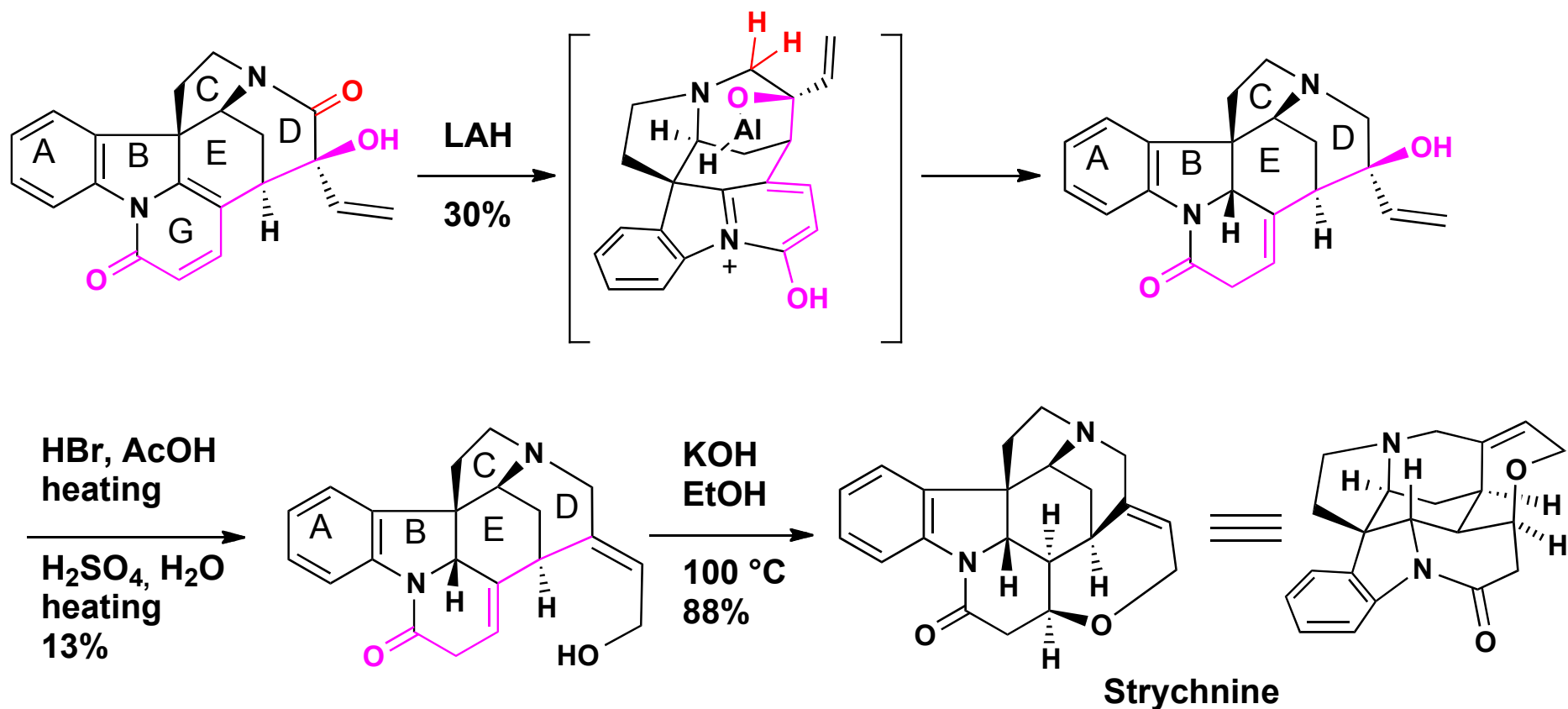


- \* The six pink carbons derived from benzene became an integral part of the D, E, G rings!
- V to VI, mechanism, stereochemistry?
- VI can be obtained by degradation of strychnine (8 steps), **2nd relay intermediate**

# Woodward's Total Synthesis of Strychnine



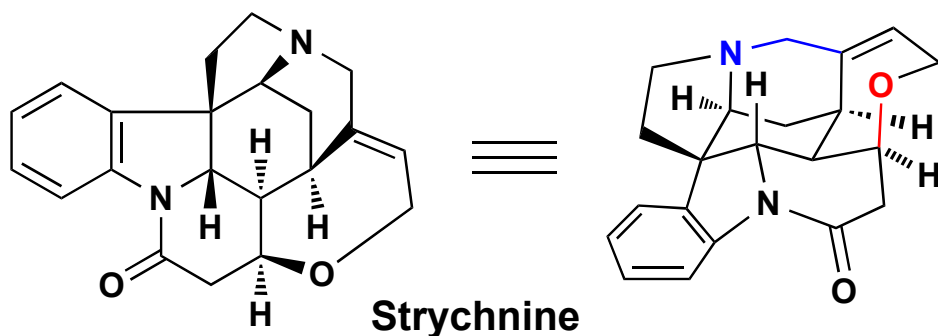
# Woodward's Total Synthesis of Strychnine



## *Neighboring group-directed Diastereoselective reduction*

Woodward, R. B.; Cava, M. P.; Ollis, W. D.; Hunger, A.; Daeniker, H. U.; Schenker, K.  
*J. Am. Chem. Soc.* **1954**, *76*, 4749-4751; *Tetrahedron* **1963**, *19*, 247-288.

# Woodward's Total Synthesis of Strychnine



"For its molecular size  
it is the most complex  
substance known."  
Robert Robinson (1952)

## Key points

- Using simple reagents available at that time.
- A classic example of relay synthesis

**28 Steps, 0.00006% overall yield, using relay intermediates**

**Truly spectacular, remained as one of the most significant landmarks in total synthesis.**

**Commercially available at 273 dollar/Kg (2011)**

Review, see:

Bonjoch, J.; D. Solé, *Chem. Rev.* **2000**, *100*, 3455-3482.

Mori, M. *Heterocycles* **2010**, *81*, 259-292.

Overman, L. E. *Angew. Chem. Int. Ed.* **2012**, *51*, 4288-4311.

# Summary of Woodward's Synthesis: Reactions and Tactics

Fischer indole synthesis

Dieckman condensation

Pictet-Spengler reaction and Interrupted Pictet-Spengler reaction

Azalacton formation

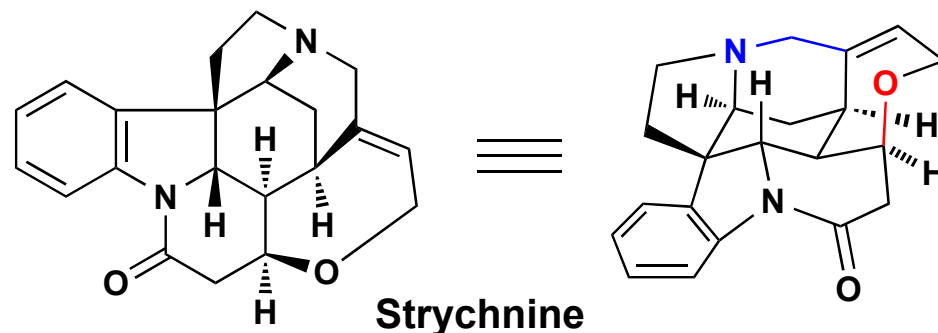
Alpha-hydroxylation of ketone (SeO<sub>2</sub>, Ene reaction/2,3- $\sigma$  rearrangement)

## *Key Features:*

*Stepwise construction of ring system*

*Neighboring group-directed Diastereoselective reduction*

# Total Synthesis of Strychnine



R. B. Woodward - Harvard University (1954)

28 steps

Philip Magnus - University of Texas (1992)

Gilbert Stork - Columbia University (1992)

Larry E. Overman - University of California, Irvine (1993)

Martin E. Kuehne - University of Vermont (1993)

**Viresh H. Rawal - The Ohio State University (1994)**

14 Steps

Josep Bonjoch & Joan Bosch - University of Barcelona (1999)

Stephen F. Martin - University of Texas (1996-2001)

Michael J. Eichberg & K. Peter C. Vollhardt - University of California, Berkeley (2000)

Graham J. Bodwell - Memorial University of Newfoundland (2002)

Miwako Mori - Hokkaido University (2002)

Masakatsu Shibasaki - University of Tokyo (2002)

Tohru Fukuyama - University of Tokyo (2004)

Albert Padwa - Emory University (2007)

Rodrigo, B. Andrada, Temple University (2010)

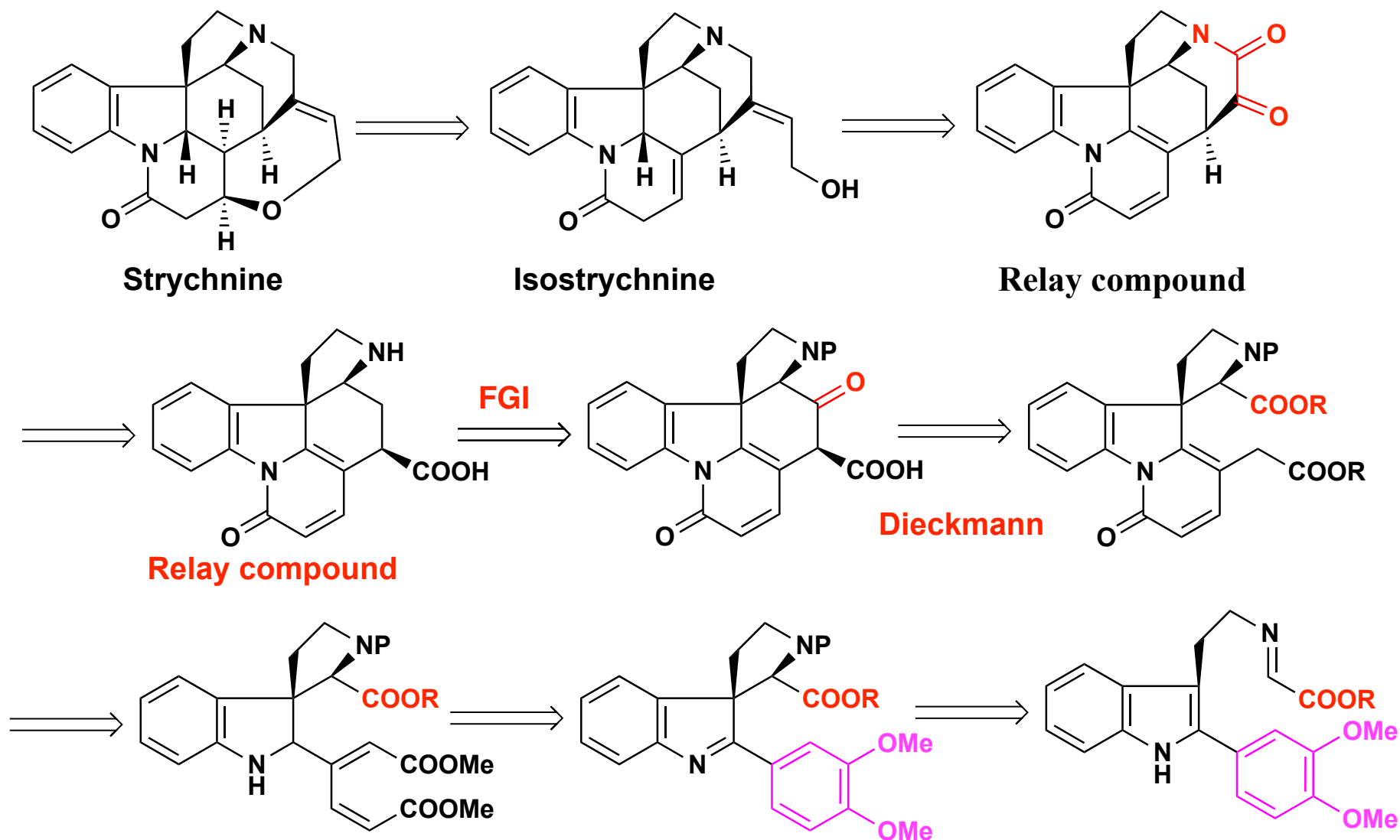
Hans-Ulrich, Reissig, Freie University Berlin (2010)

Christopher, D. Vanderwal, University of California, Irvine (2011)

6 Steps

38 years

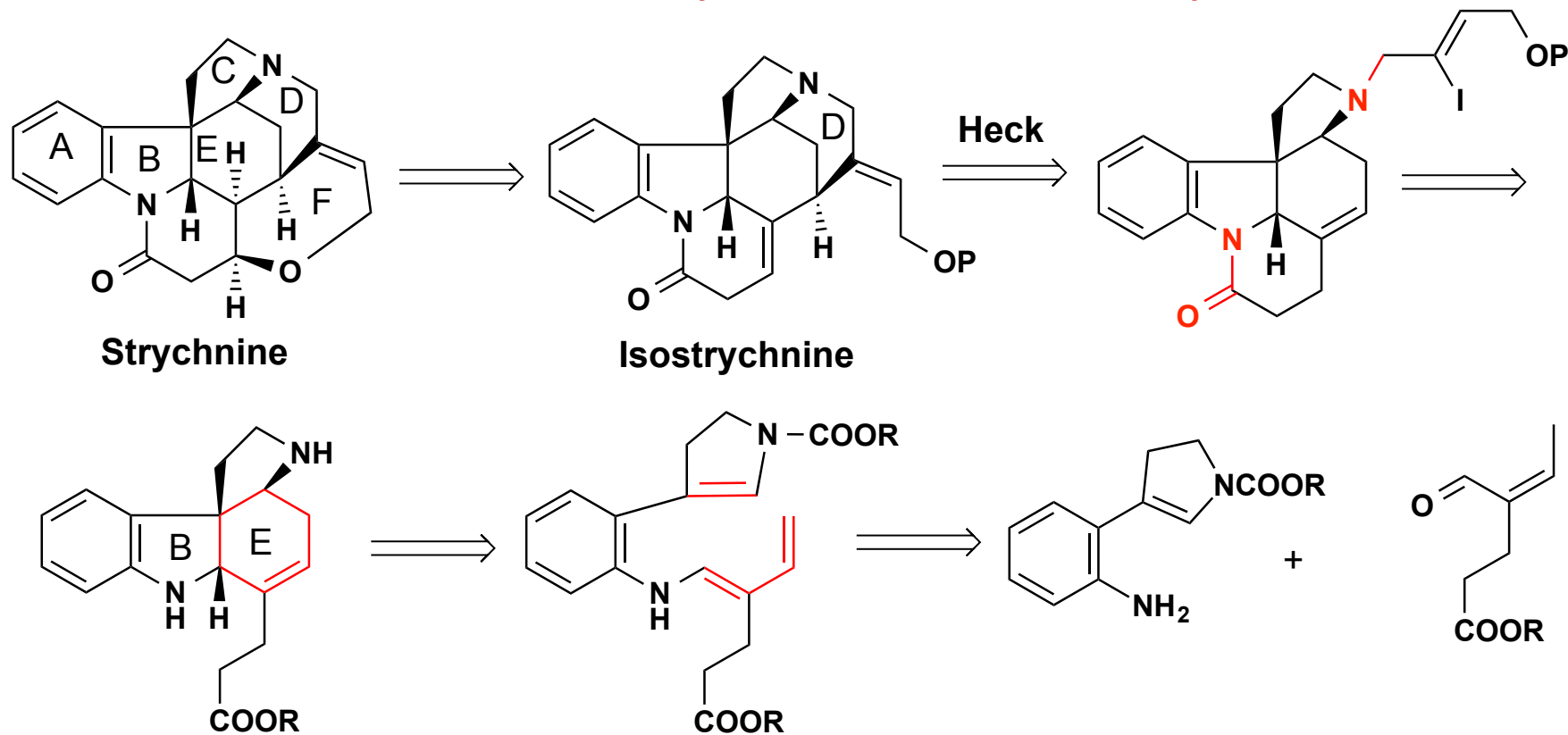
## Woodward's Total Synthesis of Strychnine: *A Relay Synthesis*



One of the key strategic considerations: **Relay Synthesis**

Woodward, R. B. *J. Am. Chem. Soc.* **1954**, *76*, 4749-4751; *Tetrahedron* **1963**, *19*, 247-288.

# Rawal's Total Synthesis of Strychnine



## Key disconnections

- \* Isostrychnine as a late stage intermediate
- \* Intramolecular Heck reaction for the construction of D-ring
- \* Intramolecular Diels-Alder cycloaddition for the simultaneous construction of B and E ring

Rawal, V. H.; Iwasa, S. *J. Org. Chem.* **1994**, *59*, 2685-2686.

## **Reactions Needed to Know**

**Cyclopropane to cyclopentene rearrangement**

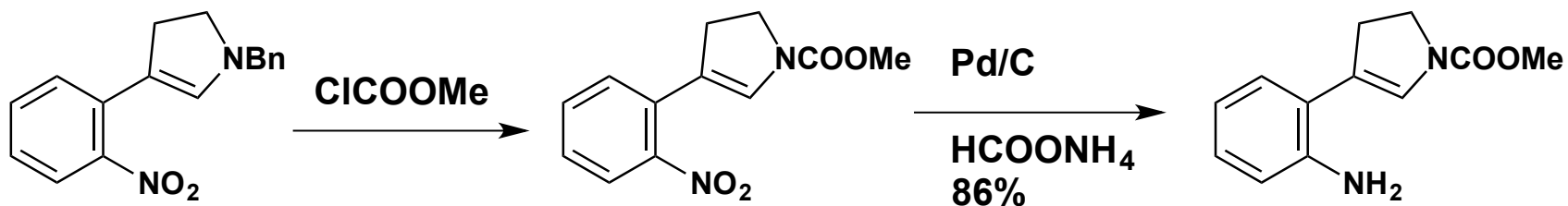
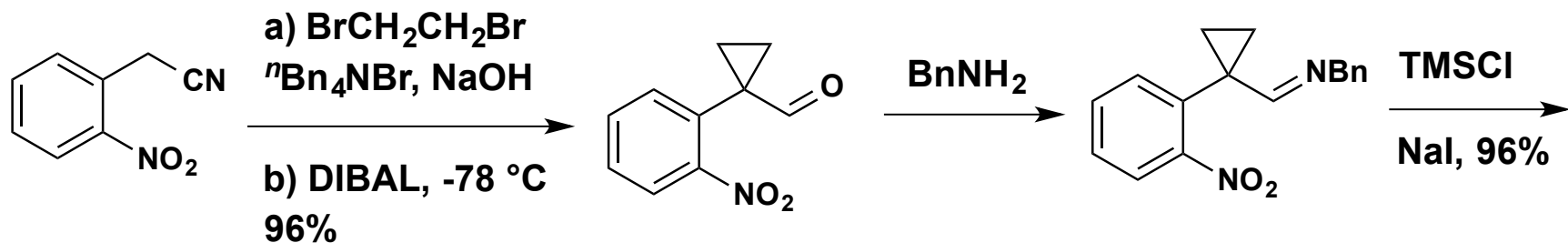
**Cyclopropylimine to pyrrole rearrangement**

**Von Braun Reaction**

**Diels-Alder reaction**

**Heck reaction**

# Rawal's Total Synthesis of Strychnine

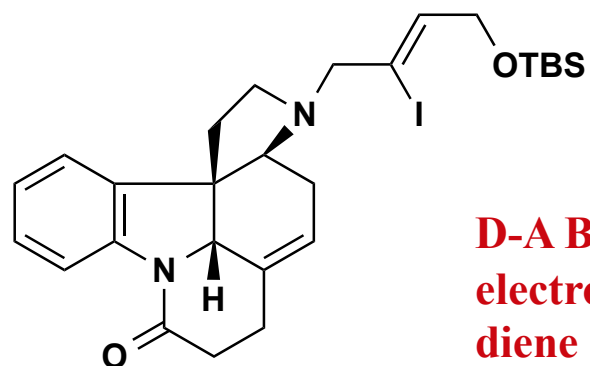
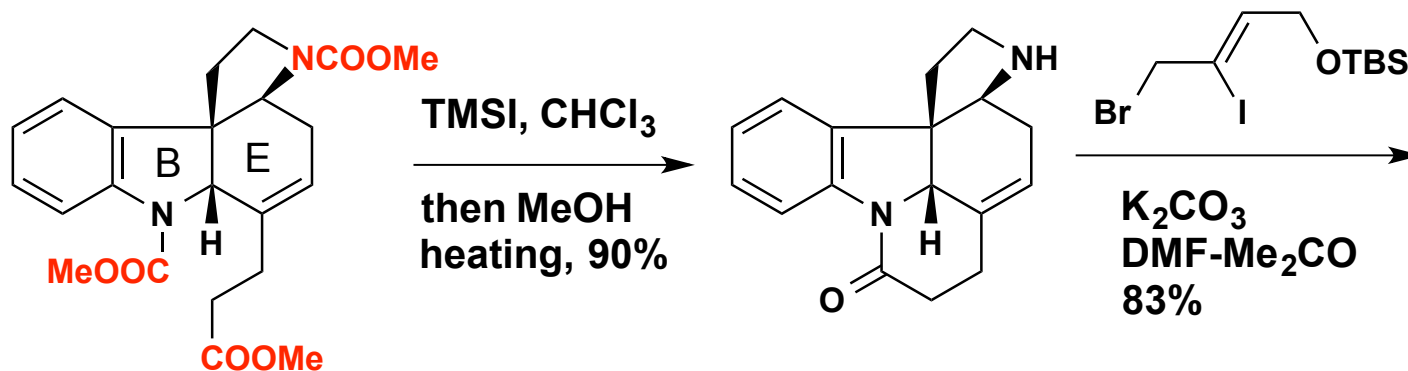
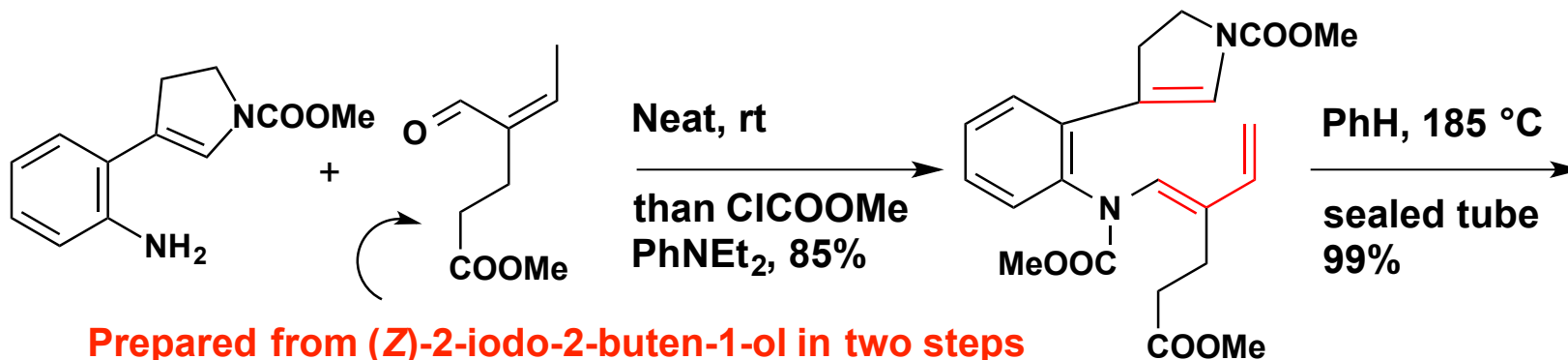


## Von Braun Reaction

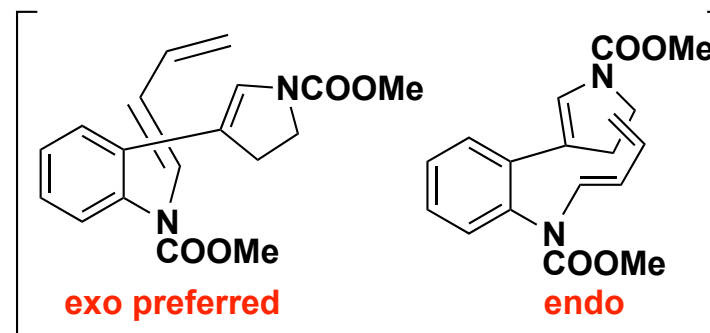
Mechanism of:  
Cyclopropyliminium ion rearrangement  
Von Braun reaction

Rawal, V. H.; Iwasa, S. *J. Org. Chem.* **1994**, *59*, 2685-2686.

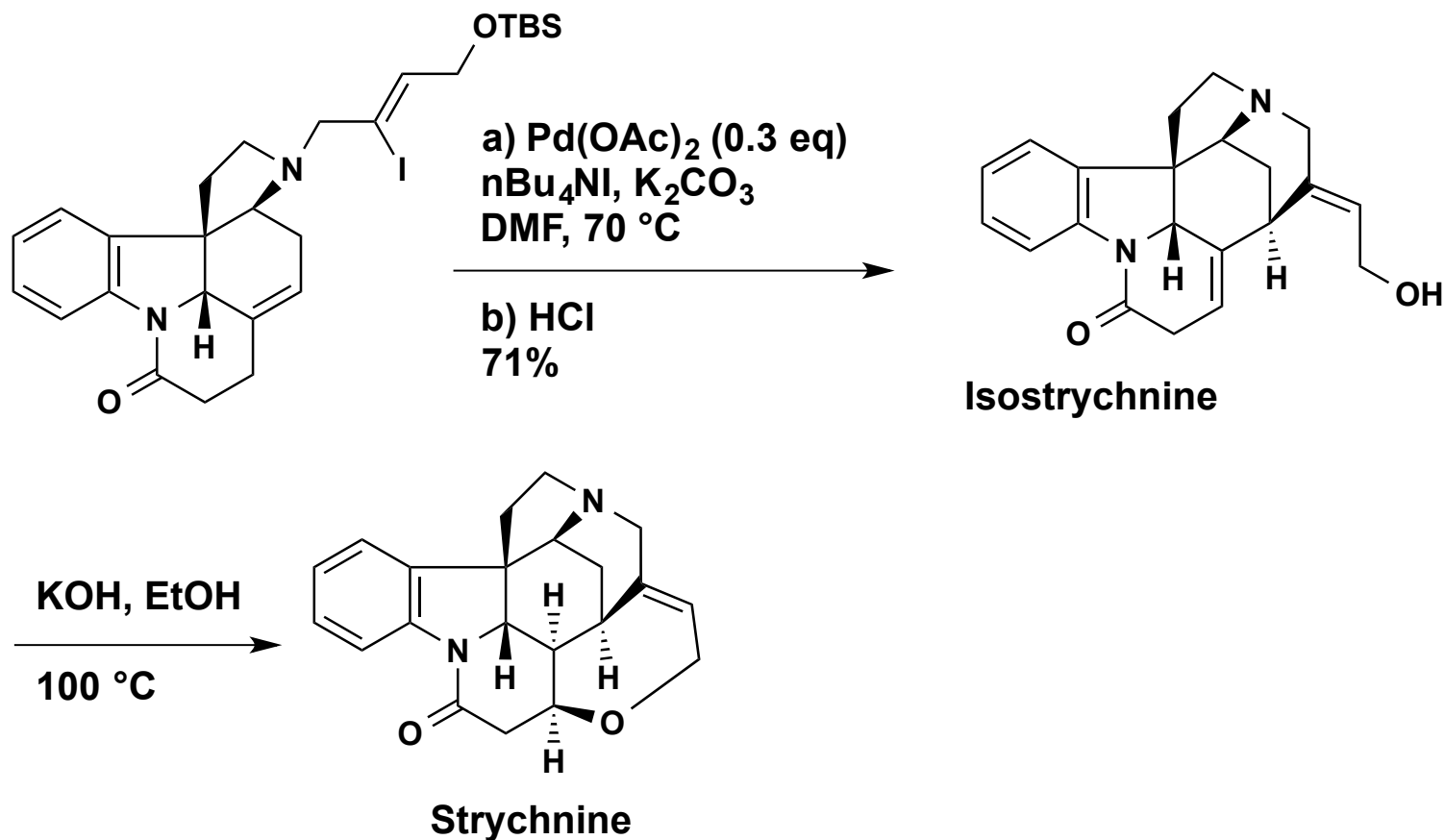
# Rawal's Total Synthesis of Strychnine



**D-A Between  
 electron rich  
 diene and dienophile!!**



# Rawal's Total Synthesis of Strychnine

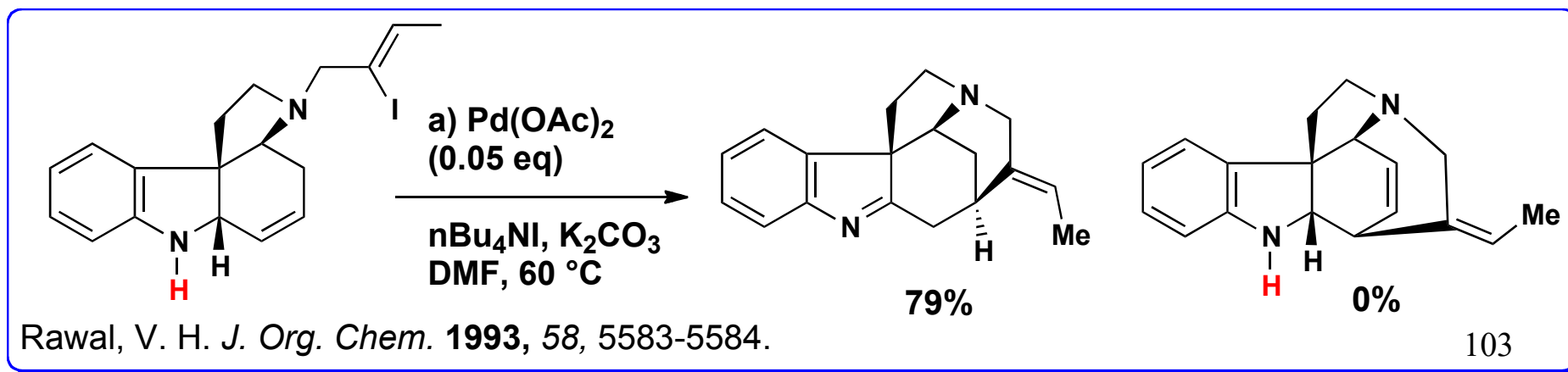
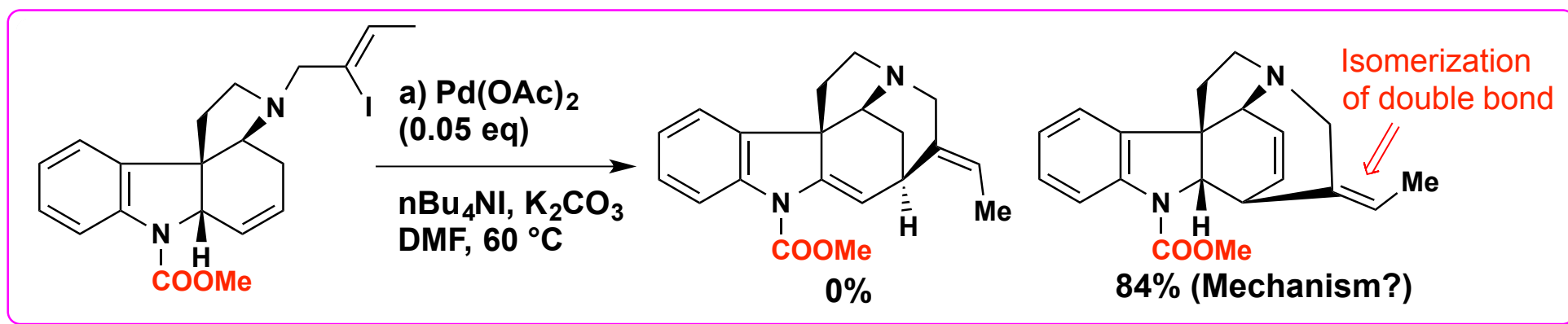
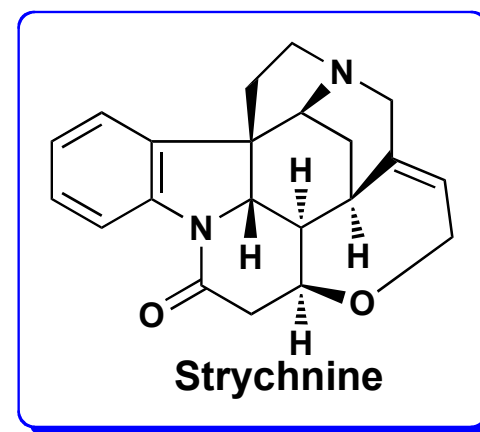
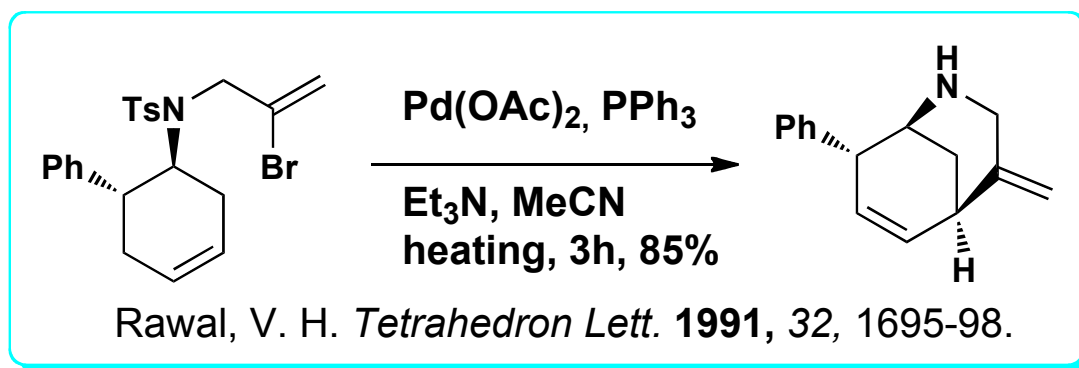


**Heck reaction:** Heck, R. F. *Org. React.* **1982**, 27, 345.  
Jeffery's conditions: *Tetrahedron Lett.* **1985**, 26, 2667.

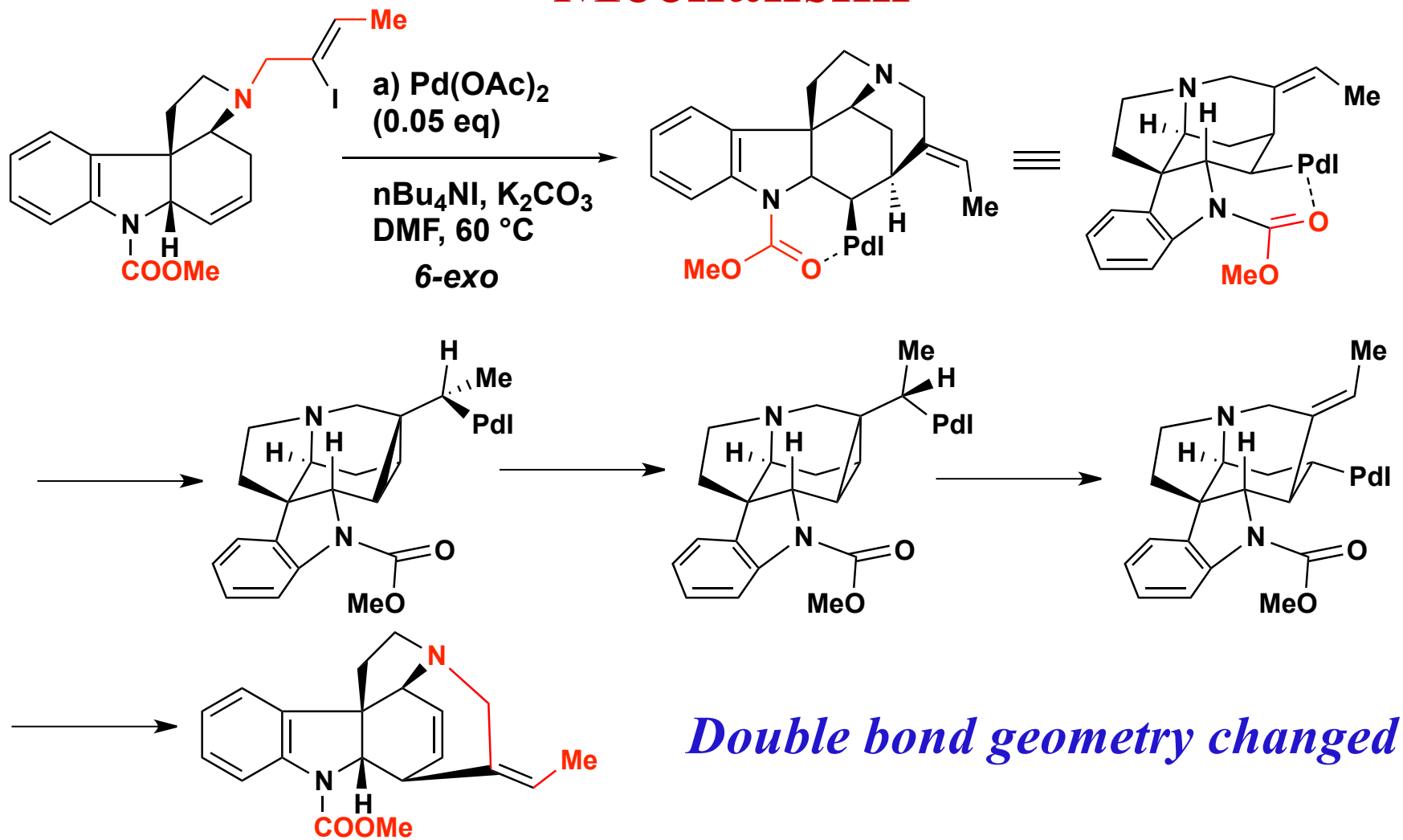
Rawal, V. H.; Iwasa, S. *J. Org. Chem.* **1994**, 59, 2685-2686.

See also: Rawal et al. *J. Am. Chem. Soc.* **1993**, 115, 3030-3031.

# Rawal's Model Studies: Influence of Protective Group



# Formation of “Formal 7-endo” Product: Mechanism



Rawal, V. H. *J. Org. Chem.* **1993**, *58*, 5583-5584.

# **Summary of Rawal's Synthesis: Reactions and Tactics**

**Cyclopropane to cyclopentene rearrangement**

**Cyclopropylimine to pyrrole rearrangement**

**Von Braun Reaction**

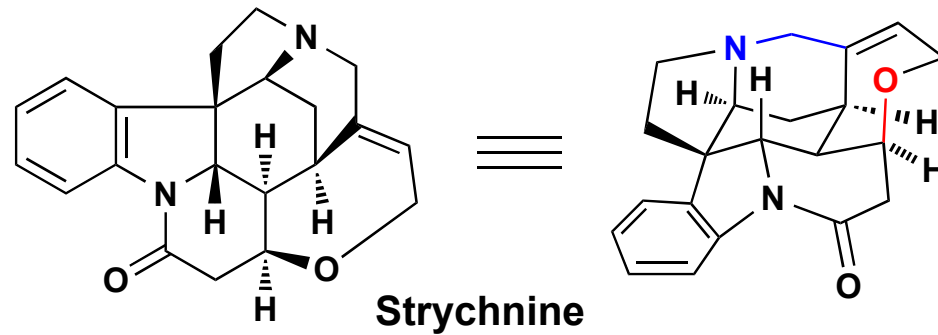
**Diels-Alder reaction**

**Heck reaction**

**Key Features:**

**Intramolecular Diels-Alder reaction for the rapid construction of tetracyclic ring system.**

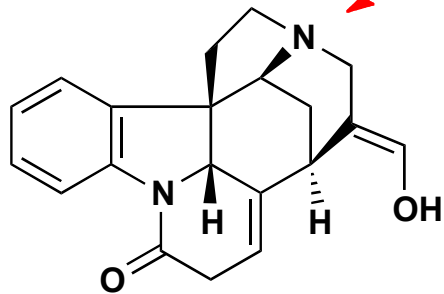
# Strychnine: Key background



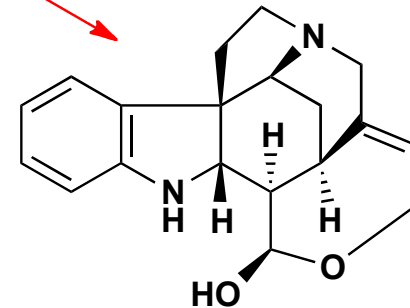
**KOH, EtOH  
85 °C**

**CH<sub>2</sub>(COOH)<sub>2</sub>, Ac<sub>2</sub>O  
NaOAc, AcOH, 110 °C**

**Degradation studies  
in 1940's**

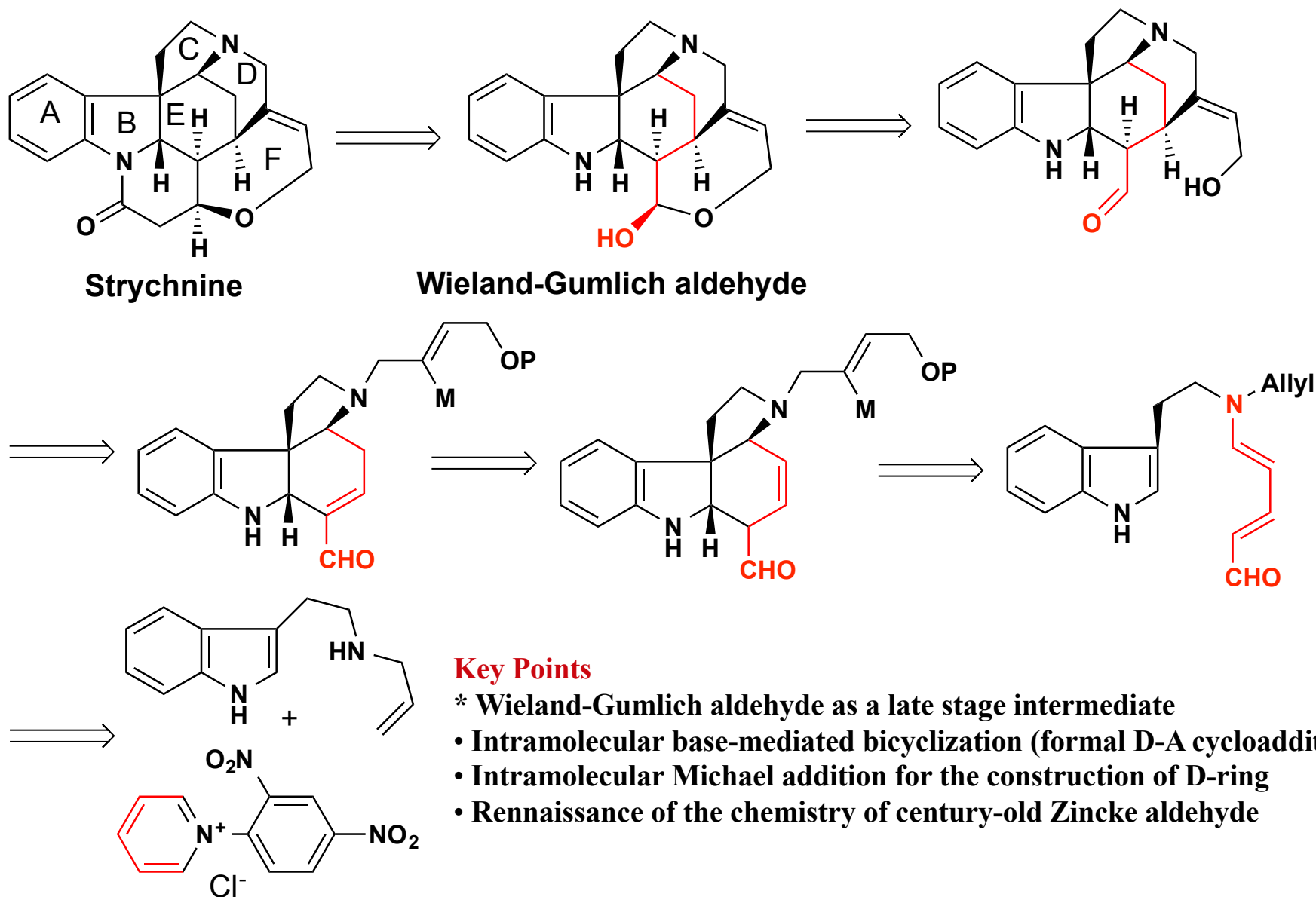


**Isostrychnine**



**Wieland-Gumlich aldehyde**

# Vanderwal's Total Synthesis of Strychnine-Retro



Martin, D. B. C.; Vandewal, C. D. *Chem. Sci.* **2011**, *2*, 649-651. *J. Am. Chem. Soc.* **2009**, *131*, 3472-3473. 107  
 Full account: *J. Org. Chem.* **2012**, *77*, 17-46.

## **Reactions Needed to Know**

**Zincke aldehyde**

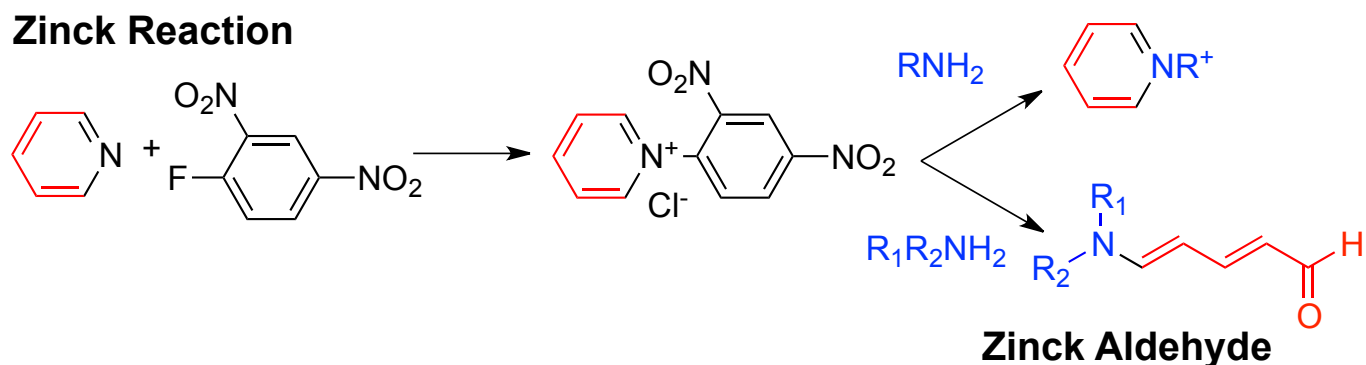
**N-Deallylation**

**Diels-Alder reaction**

**Brook rearrangement**

**Michael addition**

# Zincke Reaction and Zincke's Aldehyde



**Mechanism:** Initiated by nucleophilic addition of amine to the C-2 of pyridinium followed by ring opening ...

**Discovered by Zincke in 1904**

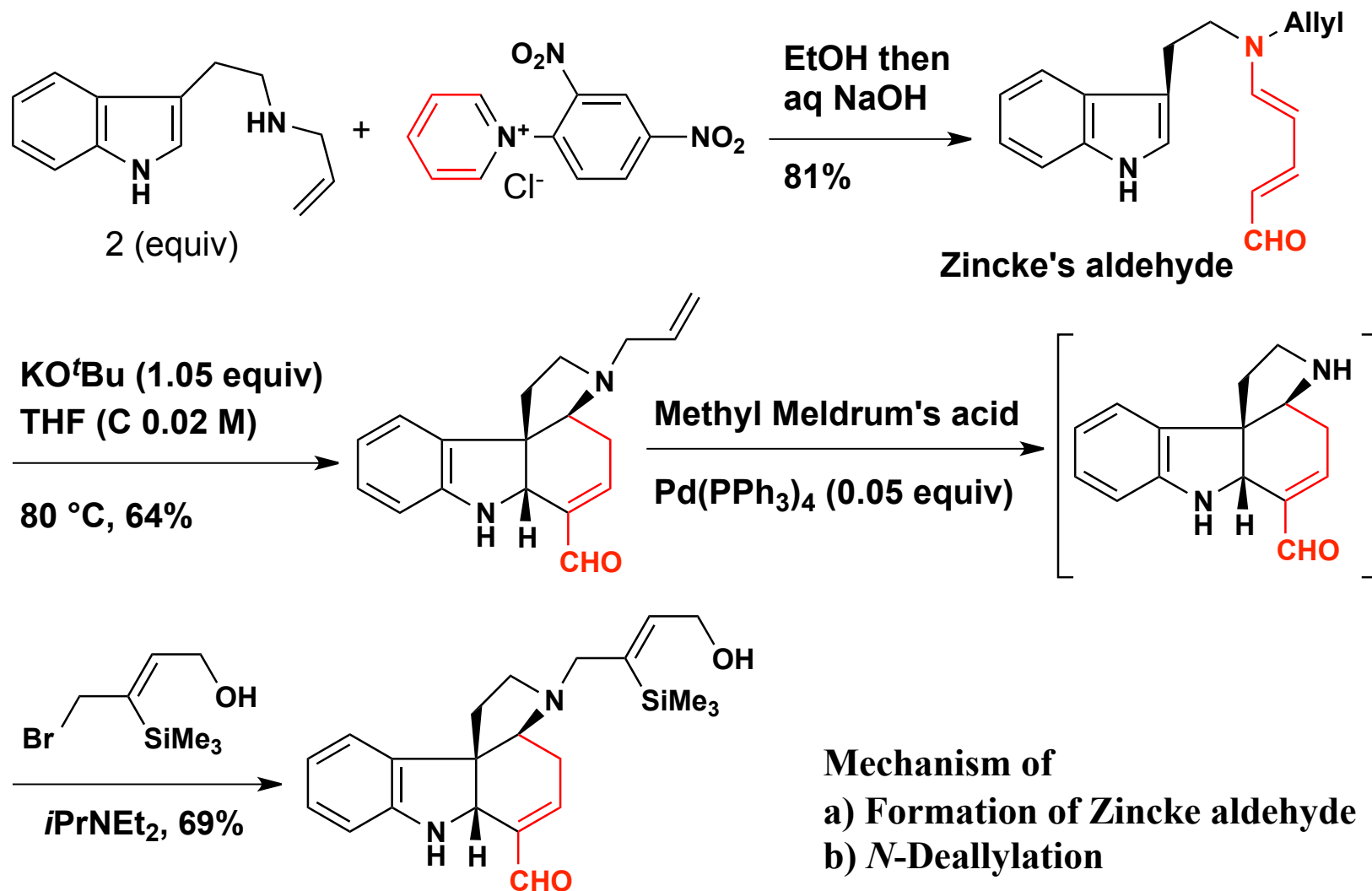
**Re-discovered in 2006 with wrong structure**

F. M. Menger, *Angew. Chem. Int. Ed.* **2007**, 46, 5889 – 5891 (VIP)

I. Yamaguchi, *Org. Lett.* 2006, 8, 4279 – 4281.

Correspondance : M. Christl, *Angew. Chem. Int. Ed.* **2007**, 46, 9152-9153.

# Vanderwal's Total Synthesis of Strychnine



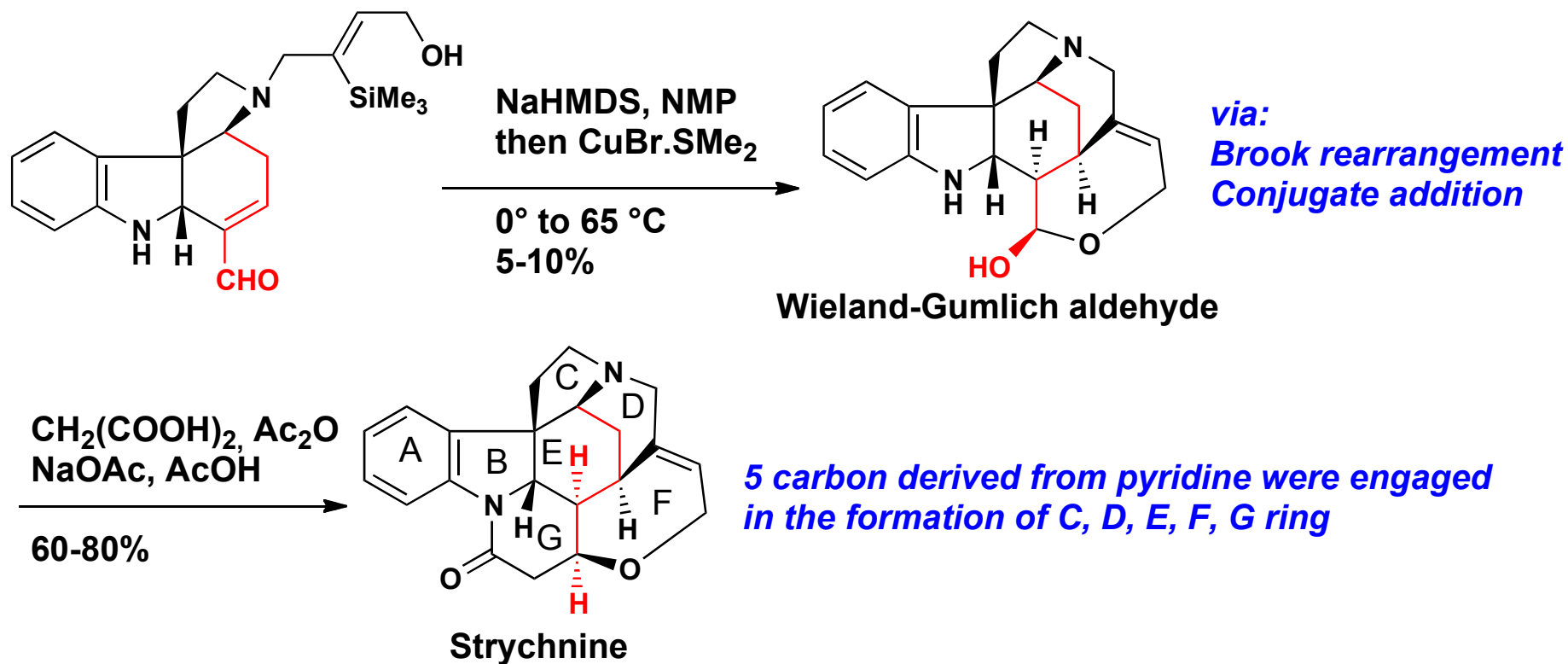
Martin, D. B. C.; Vandewal, C. D. *Chem. Sci.* **2011**, *2*, 649-651. *J. Am. Chem. Soc.* **2009**, *131*, 3472-3473.

Full account: *J. Org. Chem.* **2012**, *77*, 17-46.

Cycloaddition: Stepwise mechanism: Vandewal, C. D.; Houk, K. N. *Chem. Sci.* **2012**, *3*, DOI:

10.1039/c2sc01072k

# Vanderwal's Total Synthesis of Strychnine

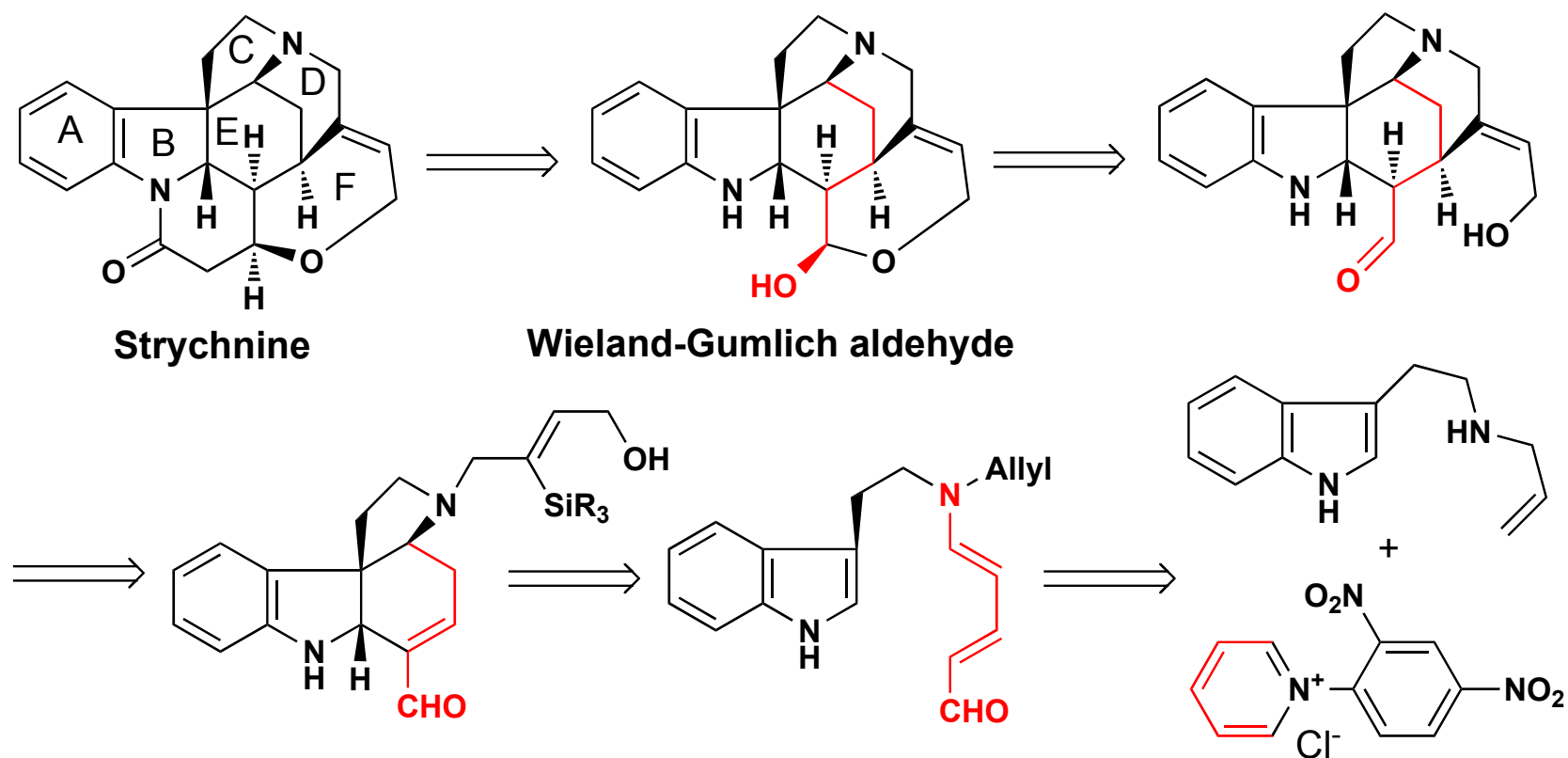


Has anybody ever imagined that strychnine can be synthesized in only **6 steps!!**

Martin, D. B. C.; Vandewal, C. D. *Chem. Sci.* **2011**, 2, 649-651. *J. Am. Chem. Soc.* **2009**, 131, 3472-3473.  
Full account: *J. Org. Chem.* **2012**, 77, 17-46.

1,4 C<sub>sp2</sub>-to-O Silyl Migration, see: Takeda, T., *J. Org. Chem.* **2002**, 67, 8450-8456.

# Vanderwal's Total Synthesis of Strychnine-Retro



**Formation of tetracycle: an Anion-Induced Biscyclization:  
Not concerted, but a **Domino** Michael addition/Mannich sequence**

Martin, D. B. C.; Vandewal, C. D. *Chem. Sci.* **2011**, *2*, 649-651. *J. Am. Chem. Soc.* **2009**, *131*, 3472-3473.

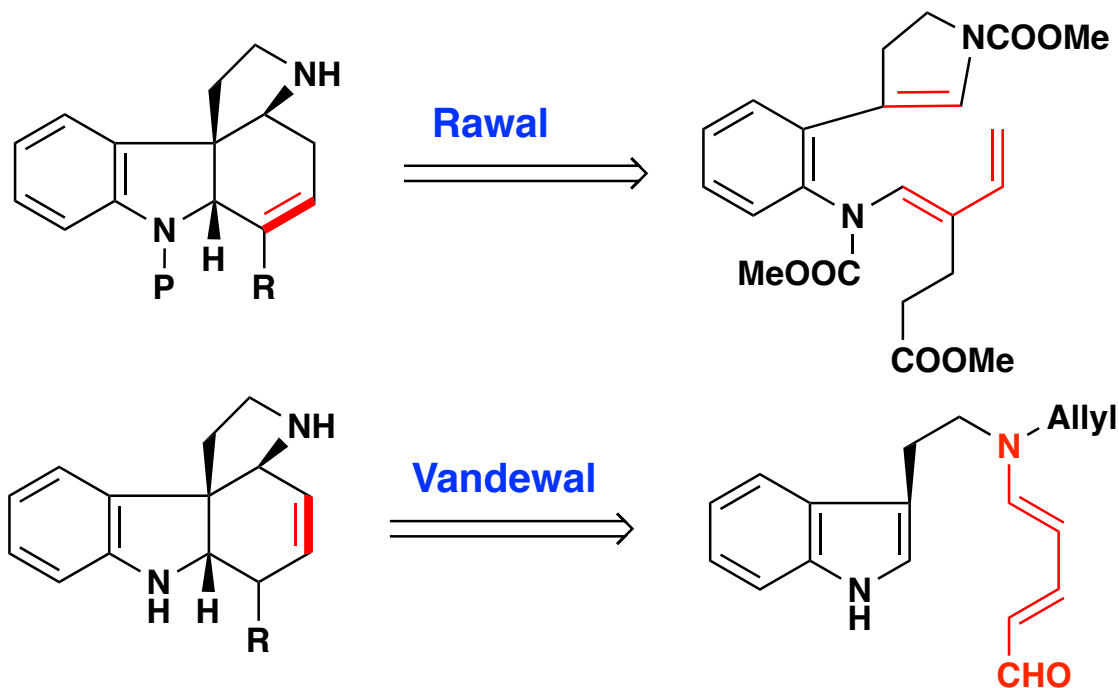
Full account: *J. Org. Chem.* **2012**, *77*, 17-46.

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# Summary of Vanderwall's Synthesis: Reactions and Tactics

- 1) Zincke aldehyde
- 2) N-Deallylation
- 3) Diels-Alder reaction
- 4) Brook-rearrangement
- 5) Michael addition

**Key Feature:** Intramolecular Diels-Alder reaction for the rapid construction of tetracyclic ring system.



# Strychnine Synthesis

	<b>Functional group manipulation</b>	<b>Protective Group</b>	<b>Redox Process</b>	<b>Multiple bond forming process</b>
<b>Woodward (1954)</b>	+++++	+++++	+++++	+
<b>Rawal (1994)</b>	+	++	++	+++
<b>Vanderwal (2011)</b>	-	+	-	+++++